

# Microscopic Legendre Transform and Canonical Ensemble

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# Overview

Legendre Transform

Canonical ensemble

Microscopic Legendre Transform

Equilibrium statistical mechanics with MLT

Comparison with Jaynes' approach

# What is a Legendre Transform?

- ▶ A Legendre transform switches from one set of variables to another—especially useful when the new variables are more natural to work with.
- ▶ Common in physics when changing from quantities like velocity to momentum, or entropy to temperature.

# Mathematical Definition

The Legendre-Fenchel transform:

$$f^*(p) = \inf_x (px - f(x)).$$

If  $f(x)$  is concave and differentiable:

$$f^*(p) = px - f(x), \quad \text{where } p = \frac{df}{dx}$$

► It replaces variable  $x$  with  $p = \frac{df}{dx}$

Known simply as Legendre transform in physics.

Legendre-Fenchel transforms in a nutshell, by Hugo Touchette

# Legendre Transform in Thermodynamics

- ▶ Internal energy:  $U(S, V)$
- ▶ Temperature:  $T = \left(\frac{\partial U}{\partial S}\right)_V$
- ▶ Helmholtz free energy:  $S \rightarrow T$ .

$$F(T, V) = U(S, V) - TS$$

Legendre Transform of entropy:

$$\beta F(\beta, V) = \beta U - S(U, V).$$

Inverse Legendre Transform:

$$S(U, V) = \beta U - \beta F(\beta, V).$$

# Canonical ensemble

The system is in thermal equilibrium with the heat reservoir:

$$p_i^* = \frac{e^{-\beta \epsilon_i}}{\sum_{i=1}^W e^{-\beta \epsilon_i}},$$

The equilibrium free energy:

$$\beta F(\beta) = -\ln \sum_{i=1}^W e^{-\beta \epsilon_i}.$$

# Canonical ensemble

Mean energy:

$$U^* = \frac{\partial}{\partial \beta}(\beta F) = \sum_{i=1}^W p_i^* \varepsilon_i$$

Entropy:

$$\begin{aligned} S(U^*) &= \beta U^* - \beta F(\beta) \\ &= - \sum_{i=1}^W p_i^* \ln p_i^* \equiv \mathcal{S}^* \end{aligned}$$

Thermodynamic entropy is equal to the Shannon entropy of the equilibrium distribution.

# Jaynes' Maximum Entropy Principle

## Statement

Given a set of constraints, the most unbiased estimate of a probability distribution is the one that maximizes the Shannon entropy:

$$S = - \sum_i p_i \ln p_i$$

subject to:

$$\sum_i p_i = 1 \quad (\text{normalization})$$

$$\sum_i p_i f_i = \langle f \rangle \quad (\text{constraint on mean value})$$



# Mathematical Solution

- ▶ Use the method of Lagrange multipliers to incorporate constraints:

$$\mathcal{L} = - \sum_i p_i \ln p_i - \alpha \left( \sum_i p_i - 1 \right) - \lambda \left( \sum_i p_i f_i - \langle f \rangle \right)$$

- ▶ Solving this ( $\delta \mathcal{L} = 0$ ) gives:

$$p_i = \frac{e^{-\lambda f_i}}{Z}, \quad Z = \sum_i e^{-\lambda f_i}$$

# The Maxent principle

Constraints:  $k = 1, \dots, m$

$$\langle f_k(x) \rangle = \sum_{i=1}^n p_i f_k(x_i) = F_k$$

Solution:

$$p_i = \exp[-\lambda_0 - \sum_{j=1}^m \lambda_j f_j(x_i)]$$

$$\lambda_0 = \ln Z = \ln \sum_{i=1}^n \exp[-\sum_{j=1}^m \lambda_j f_j(x_i)].$$

$$F_k = -\frac{\partial}{\partial \lambda_k} \ln Z.$$

$$S(F_1, \dots, F_m) = \sum_{k=1}^m \lambda_k F_k + \ln Z(\lambda_1, \dots, \lambda_m).$$

## Example: Canonical Ensemble

- ▶ Maximize entropy with given average energy ( $\sum_i p_i E_i = U$ ):
- ▶ Variational condition:

$$\sum_{i=1}^W (\beta \varepsilon_i + 1 + \ln p_i + \alpha) dp_i = 0.$$

- ▶ Resulting distribution:

$$p_i = \frac{e^{-\beta E_i}}{Z}, \quad Z = \sum_i e^{-\beta E_i}$$

- ▶ Legendre Transform structure:

$$S(U) = \beta U + \ln Z(\beta).$$

# Interpretation

Canonical entropy  $\Longleftrightarrow$  Shannon entropy

- ▶ Equilibrium free energy is related to thermodynamic entropy via Legendre transform
- ▶ Thermodynamic entropy may be related to Shannon entropy via Jaynes' principle.

Can the equilibrium free energy be directly related to Shannon entropy via an optimization procedure?

# Microscopic Legendre Transform ( $\mathcal{L}_{\mathcal{M}}$ )

$$\beta F(\beta, V, N) = \underset{U}{\text{Min}} \{ \beta U - S(U, V, N) \}.$$

$$\beta U = \beta \sum_{i=1}^W p_i \varepsilon_i \rightarrow \sum_{i=1}^W p_i \cdot (\beta \varepsilon_i)$$

$$\beta F(\beta \varepsilon_1, \dots, \beta \varepsilon_W) = \underset{p_1, \dots, p_W}{\text{Min}} \left\{ \sum_{i=1}^W p_i \cdot (\beta \varepsilon_i) - \tilde{S}(p_1, \dots, p_W) \right\}$$

$$\tilde{S}(p_1, \dots, p_W) = - \sum_{i=1}^W p_i \ln p_i - \alpha \left( \sum_{i=1}^W p_i - 1 \right)$$

Optimum condition:

$$\left. \frac{\partial \tilde{S}}{\partial p_i} \right|_{p_i = p_i^*} = \beta \varepsilon_i.$$

## Exact differential of entropy

$$\tilde{\mathcal{S}} = - \sum_{i=1}^W p_i \ln p_i - \alpha \left( \sum_{i=1}^W p_i - 1 \right)$$

$$d\tilde{\mathcal{S}} = \sum_{i=1}^W \frac{\partial \tilde{\mathcal{S}}}{\partial p_i} dp_i,$$

where

$$\frac{\partial \tilde{\mathcal{S}}}{\partial p_i} = -(1 + \ln p_i + \alpha).$$

At the optimal point ( $p_i = p_i^*$ ):

$$\frac{\partial \tilde{\mathcal{S}}}{\partial p_i} = \beta \varepsilon_i.$$

$$\beta \varepsilon_i + 1 + \ln p_i^* + \alpha = 0 \quad \Longrightarrow \quad p_i^* = e^{-(1+\alpha)} \cdot e^{-\beta \varepsilon_i}$$

## Definition of heat

$$\delta Q = \sum_{i=1}^W \varepsilon_i dp_i$$

$$d\tilde{S} = \sum_{i=1}^W \frac{\partial \tilde{S}}{\partial p_i} dp_i.$$

At equilibrium,

$$\frac{\partial \tilde{S}}{\partial p_i} = \beta \varepsilon_i.$$

$$\therefore d\tilde{S}^* = \beta \sum_{i=1}^W \varepsilon_i dp_i^*.$$

## MLT and Jaynes' approach

Consider a system in equilibrium with the heat reservoir.

A reversible process,

$$d\tilde{S} + dS_R = 0.$$

$$d\tilde{S} = - \sum_{i=1}^W (1 + \ln p_i + \alpha) dp_i.$$

$$\delta Q = \sum_{i=1}^W \varepsilon_i dp_i.$$

Since this heat is exchanged with the heat reservoir, so the change in the entropy of the latter is  $dS_R = -\beta \delta Q$ .

$$\sum_{i=1}^W (\beta \varepsilon_i + 1 + \ln p_i + \alpha) dp_i = 0.$$



# Conclusion

- ▶ The microscopic Legendre transform offers a novel perspective on the relationship between Shannon entropy and Helmholtz free energy.
- ▶ It provides an alternative derivation of the canonical ensemble and underscores the connection between information-theoretic and thermodynamic entropy.
- ▶ Future applications: Non-equilibrium processes. Generalized entropies.

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## RESEARCH

# Microscopic Legendre Transform, Canonical Ensemble and Jaynes' Maximum Entropy Principle

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Thank You  
for your attention