

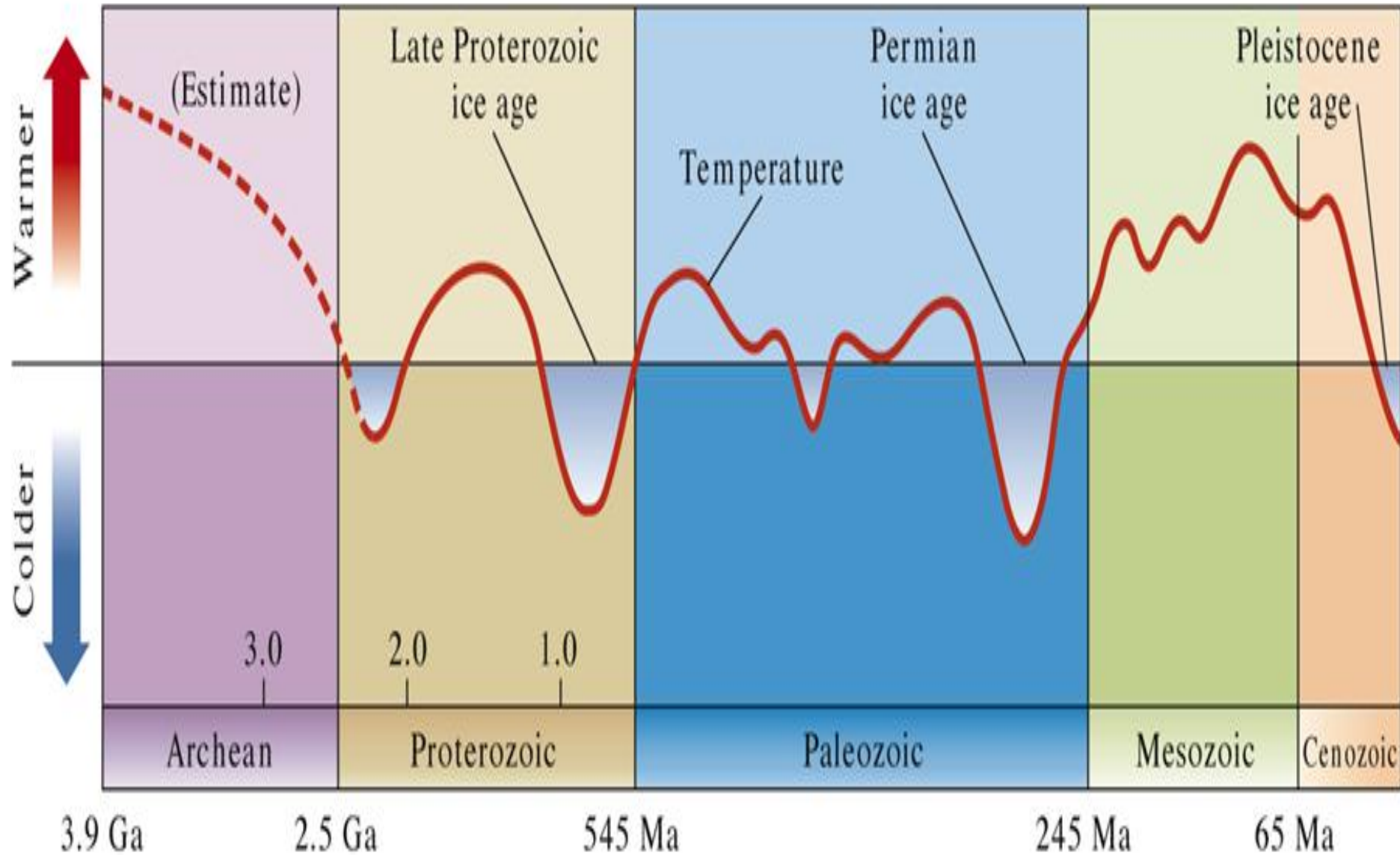
What can we learn about climate change from energy balance models?

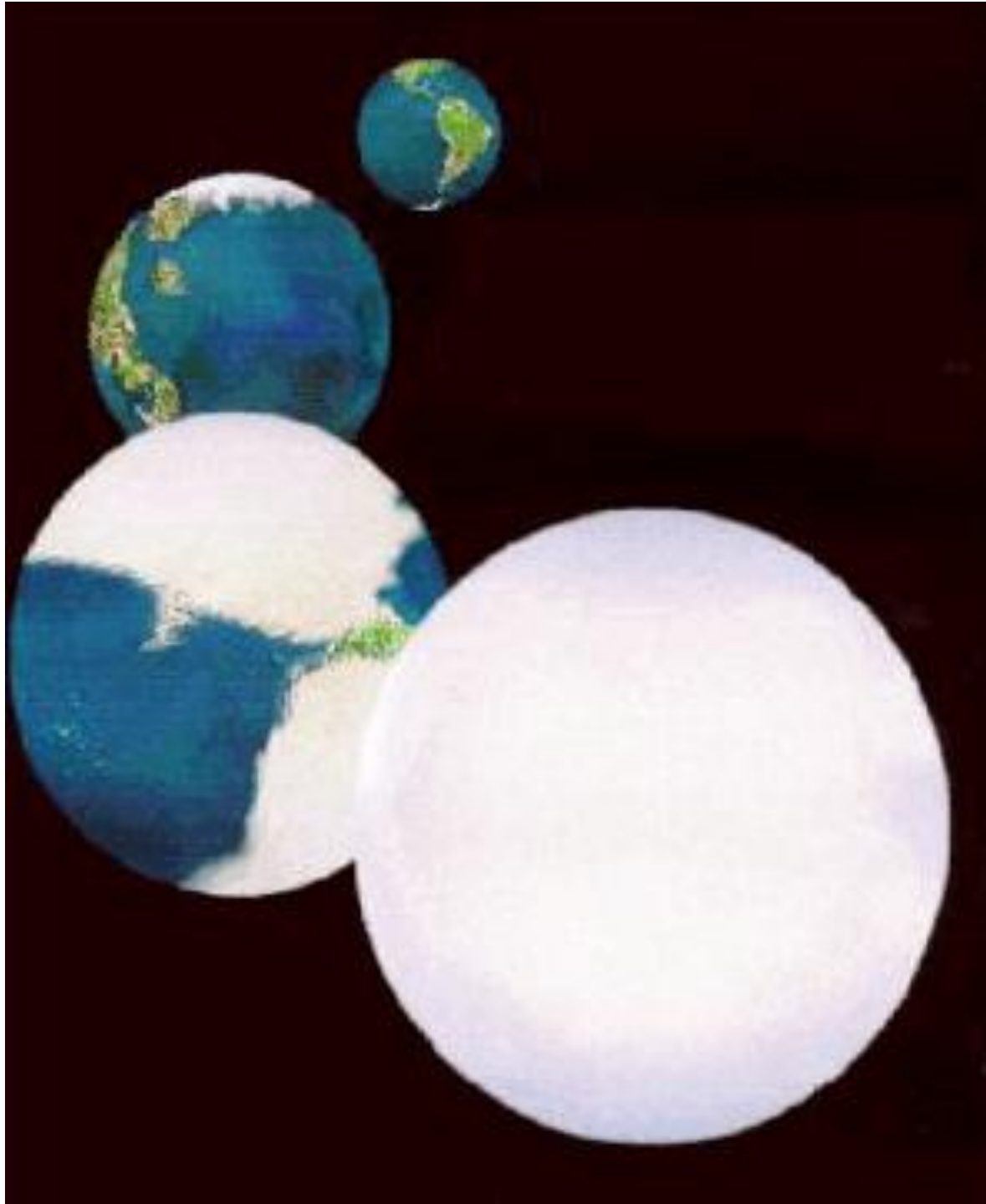
J.Srinivasan

**Divecha centre for Climate Change
&**

**Centre for Atmospheric and Oceanic Sciences
Indian Institute of Science**

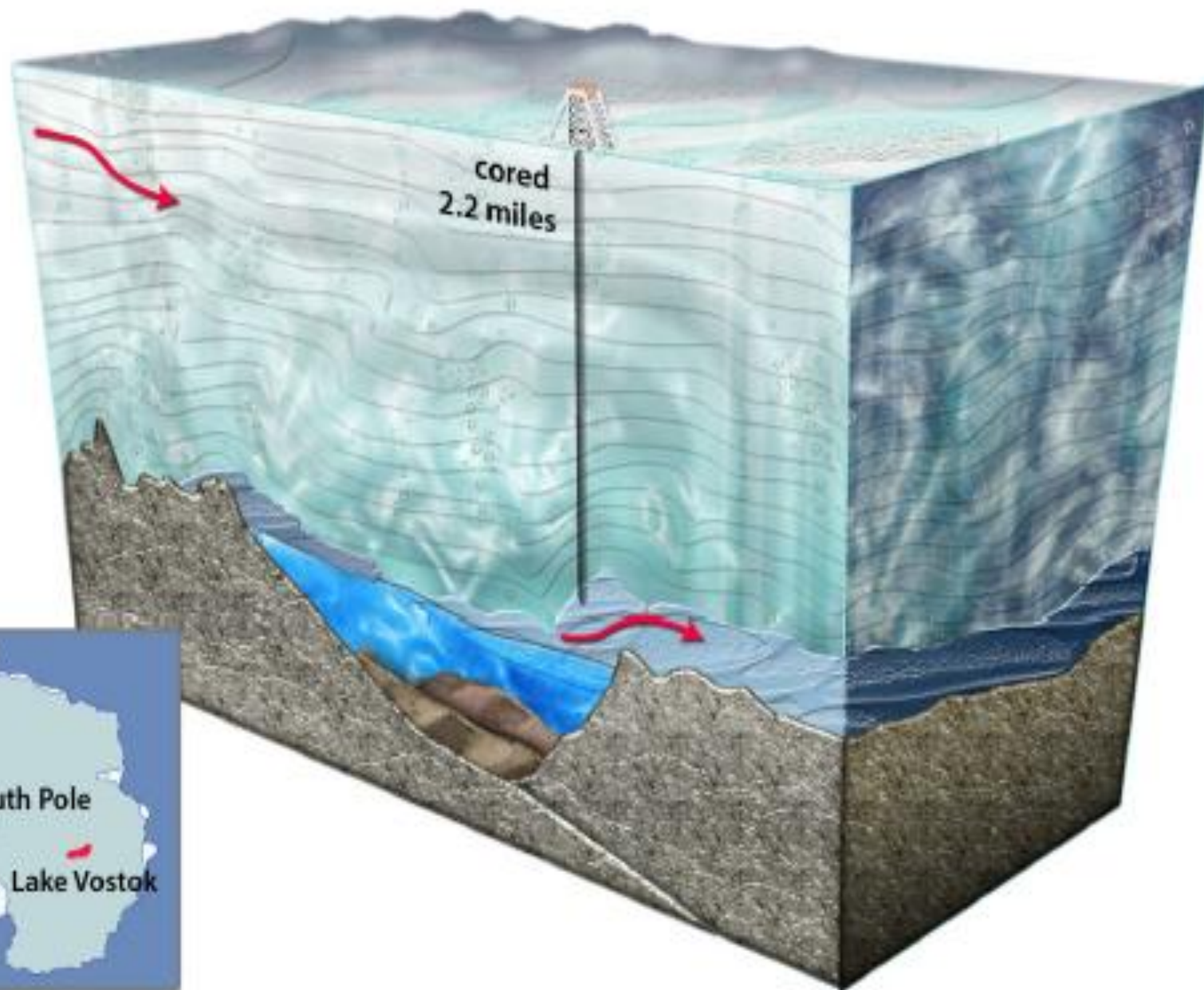
Global Climate History



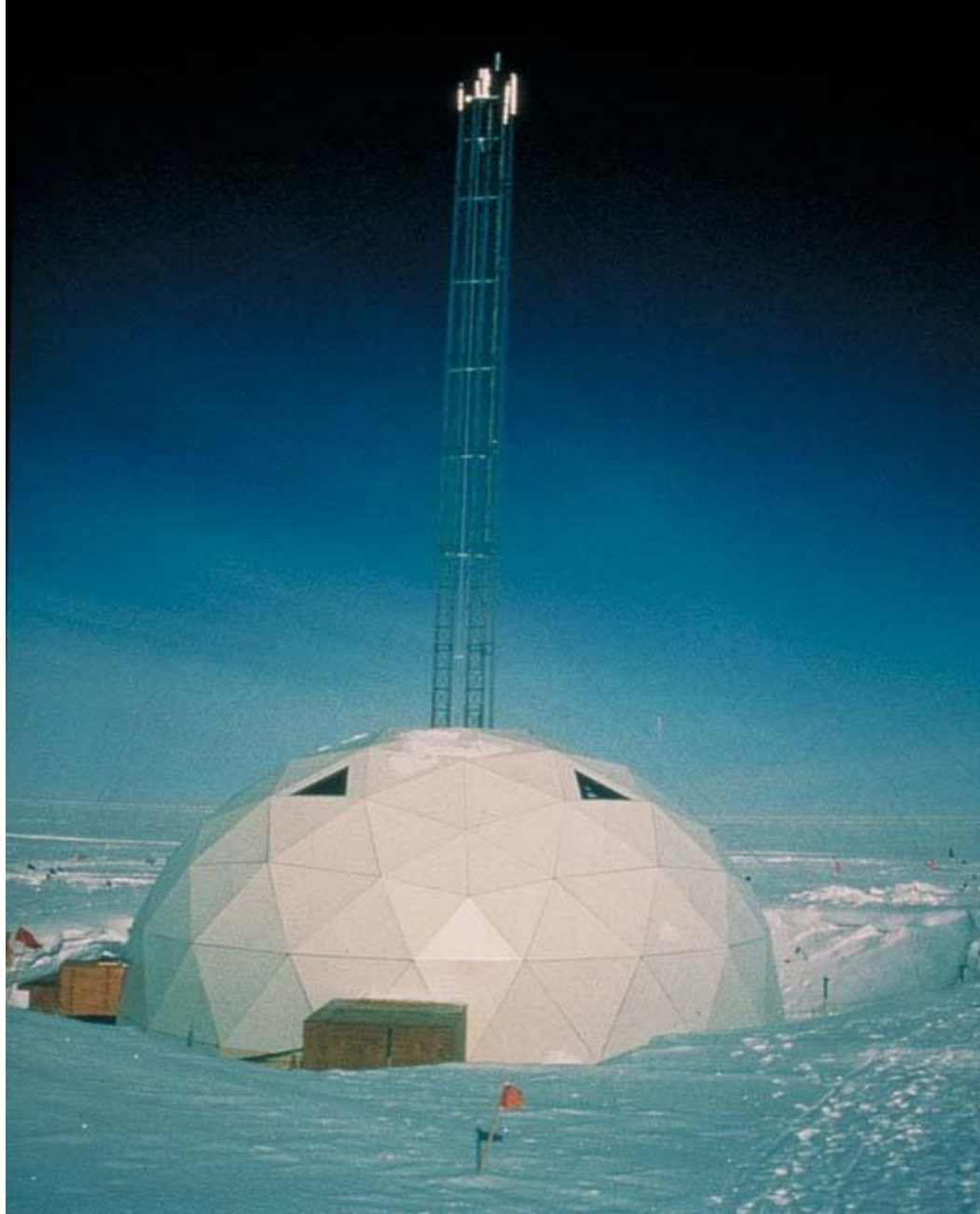


**Oscillations
between
ice- free
and
ice-
covered
earth**

Lake Vostok, Antarctica



GISP2 Drilling Project

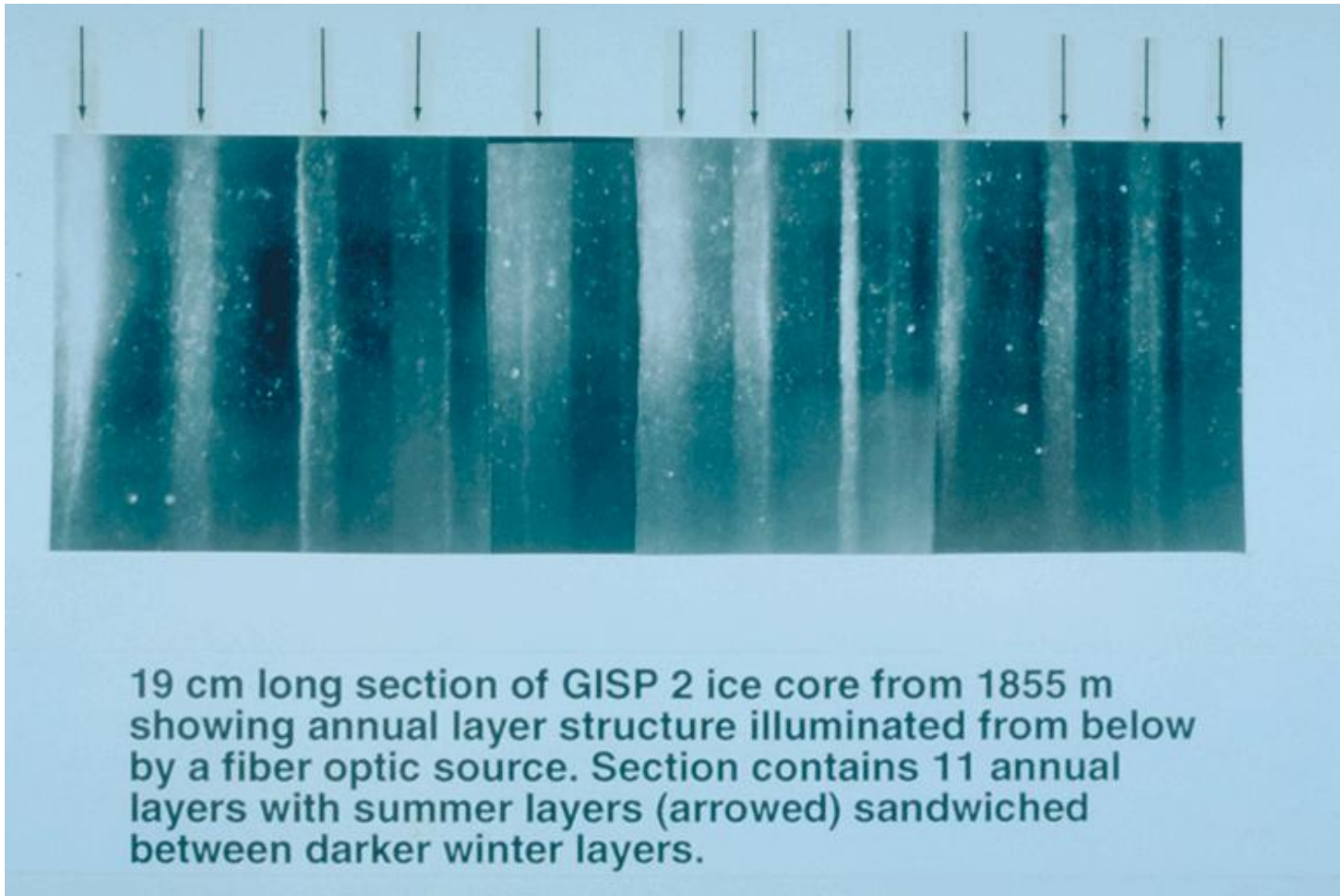


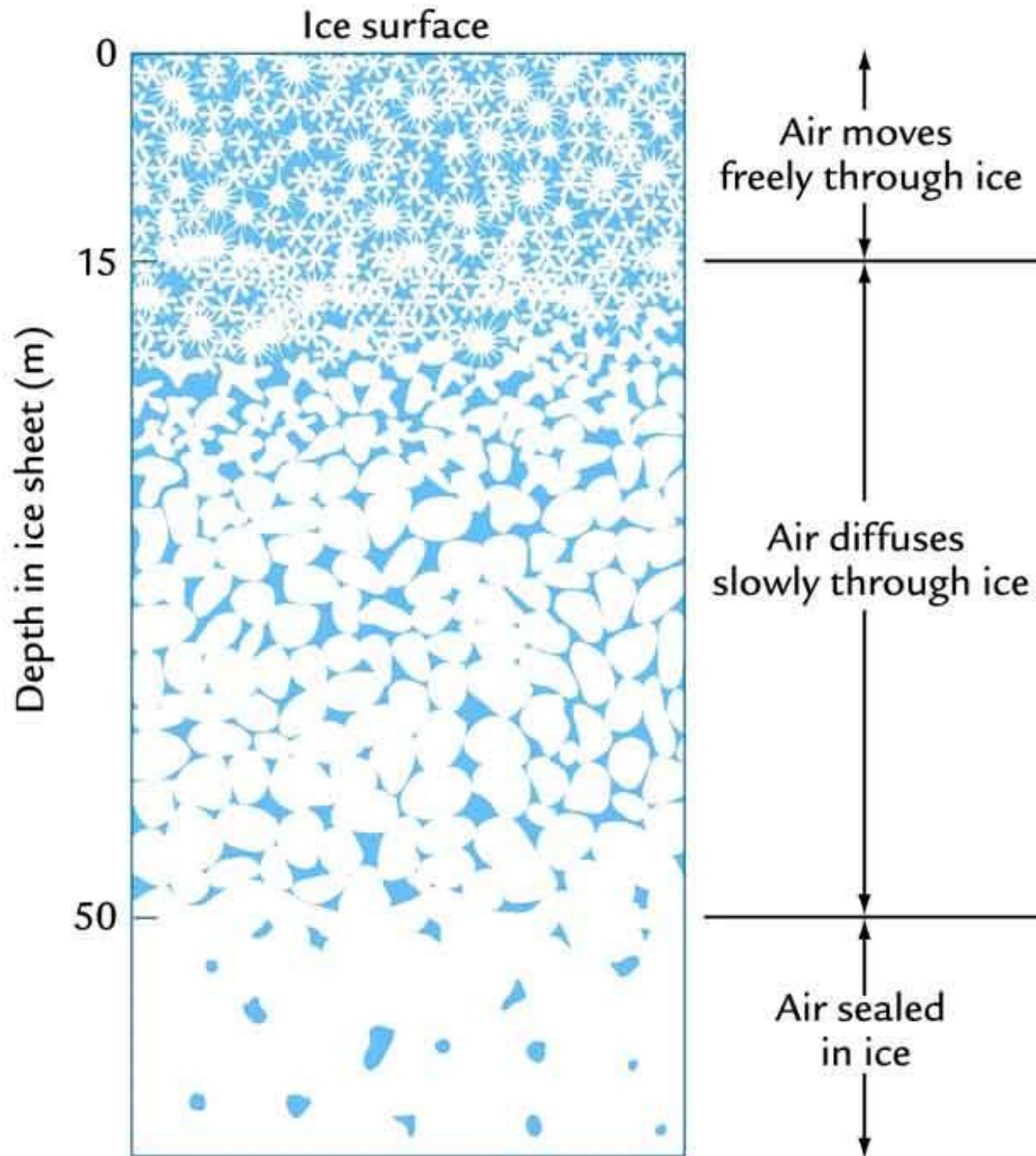


Extracting An Ice Core

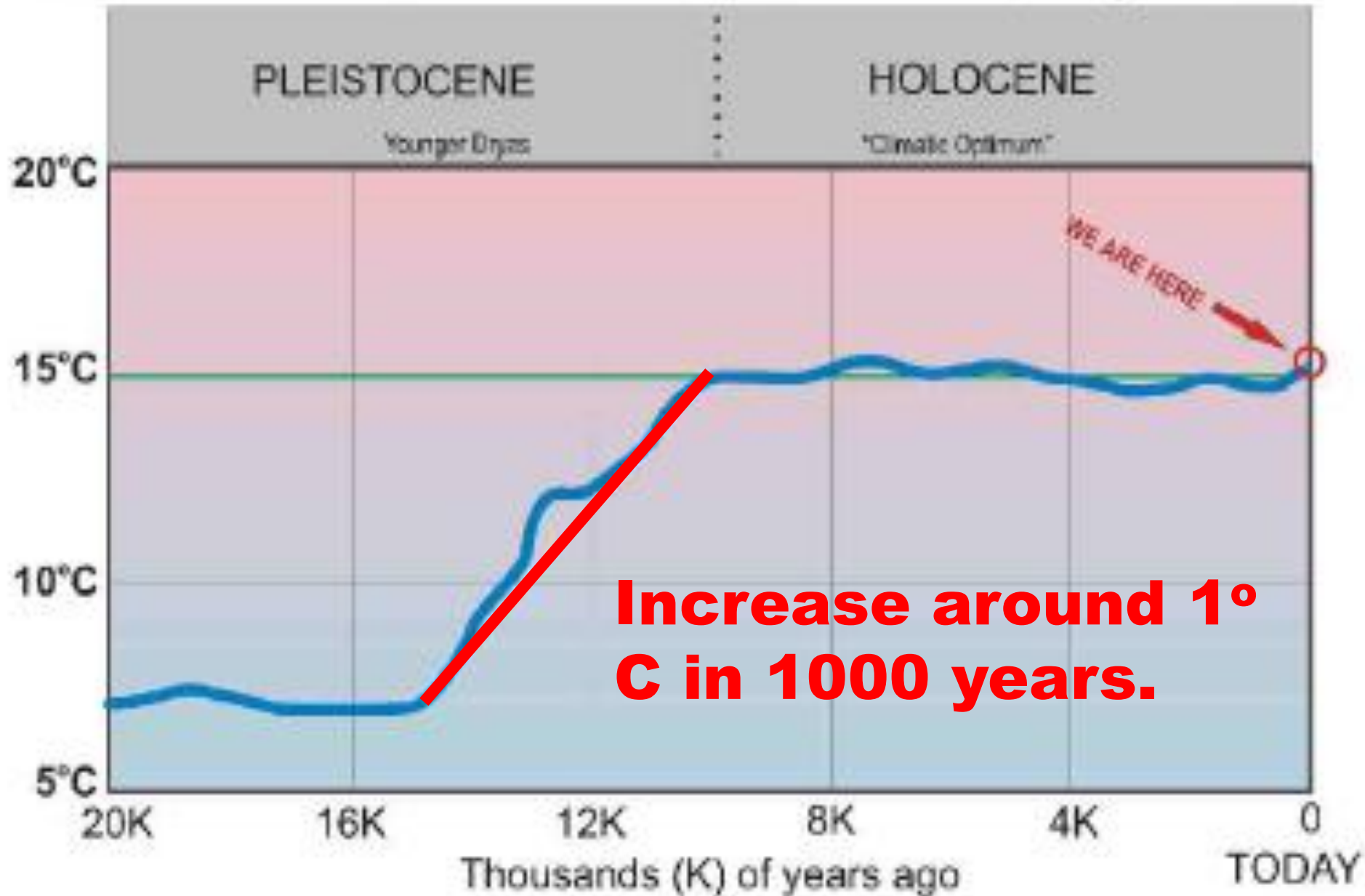


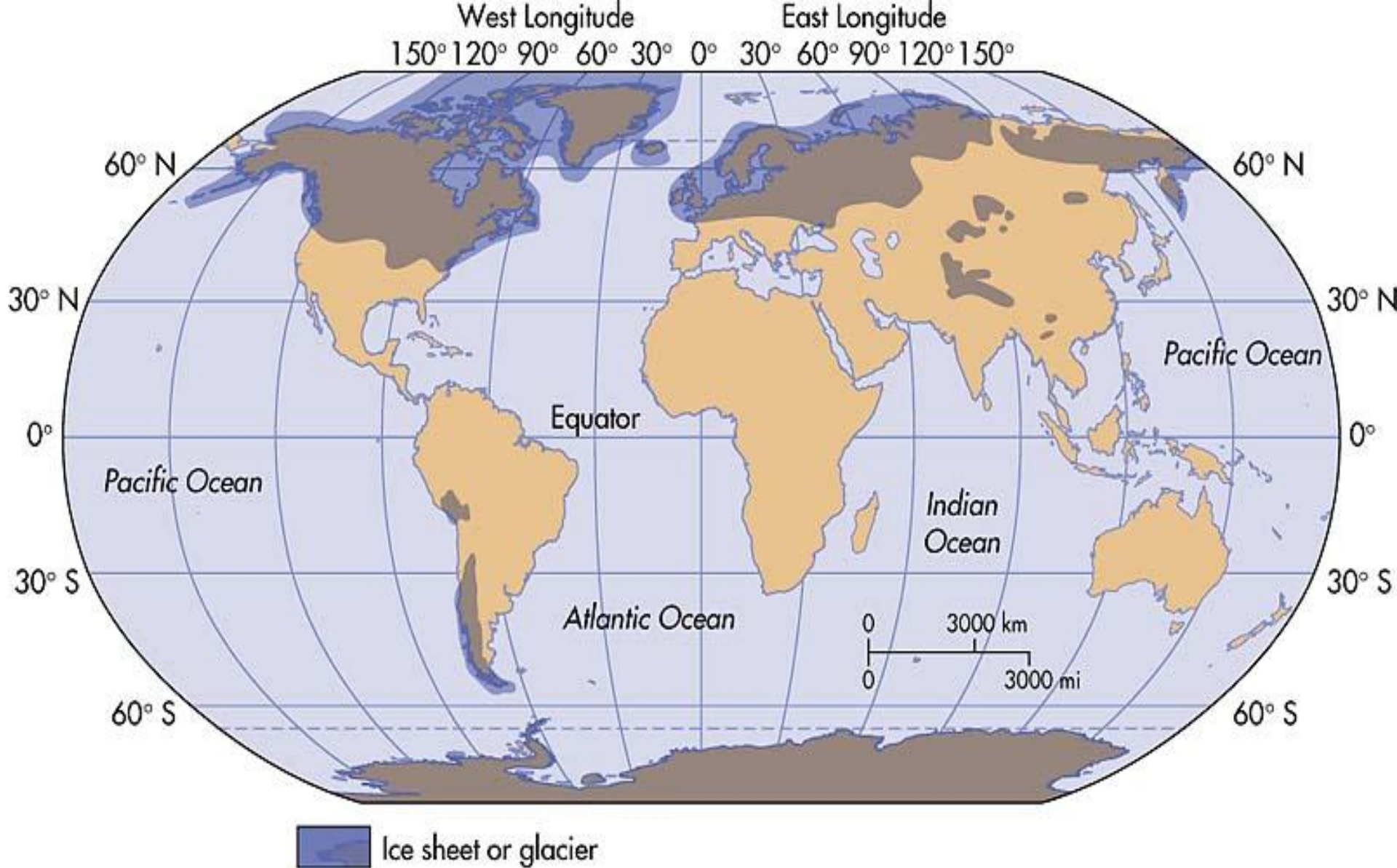
Annual Layers In Ice Core





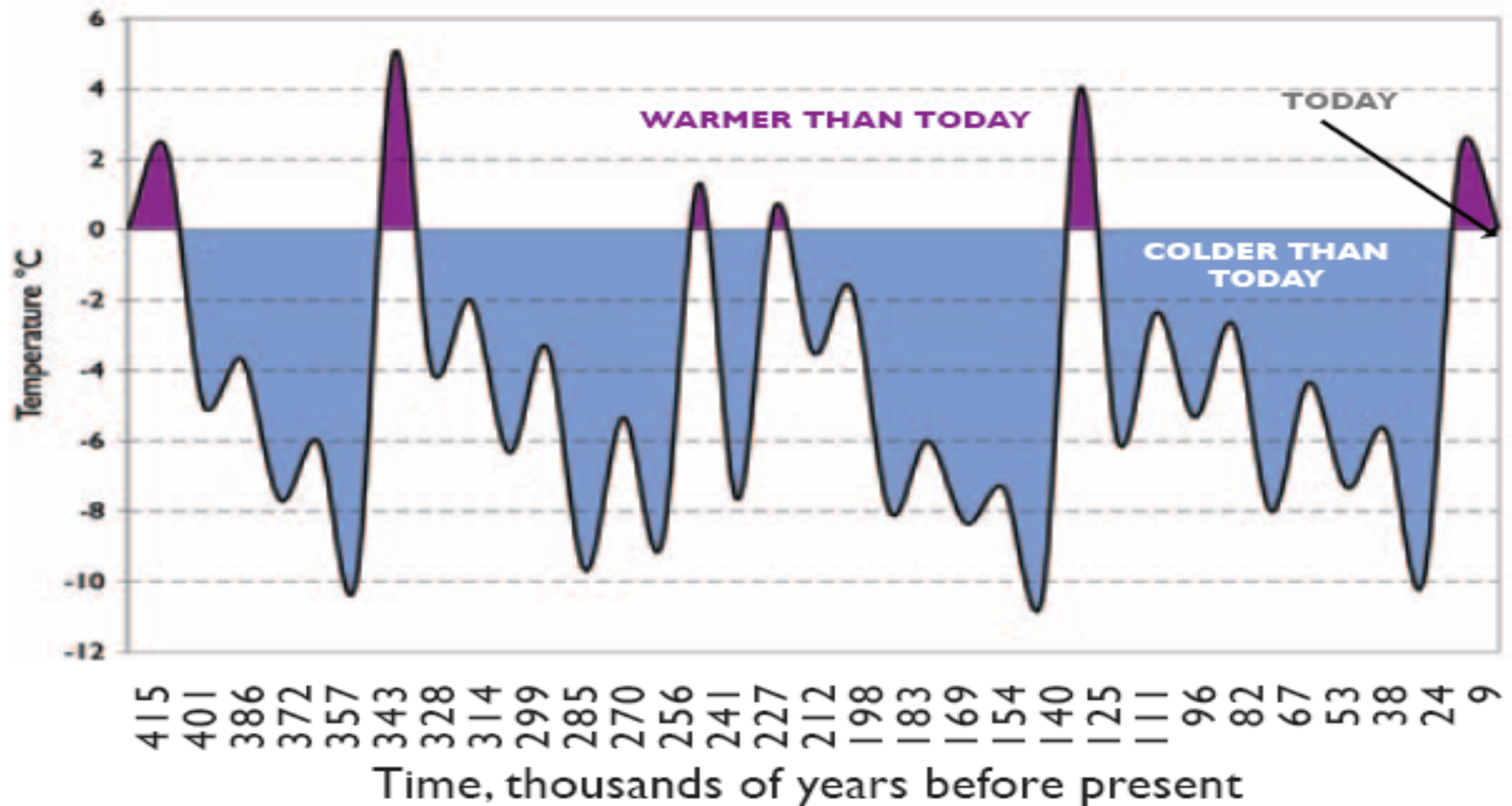
AVERAGE GLOBAL TEMPERATURES - the last 20,000 years





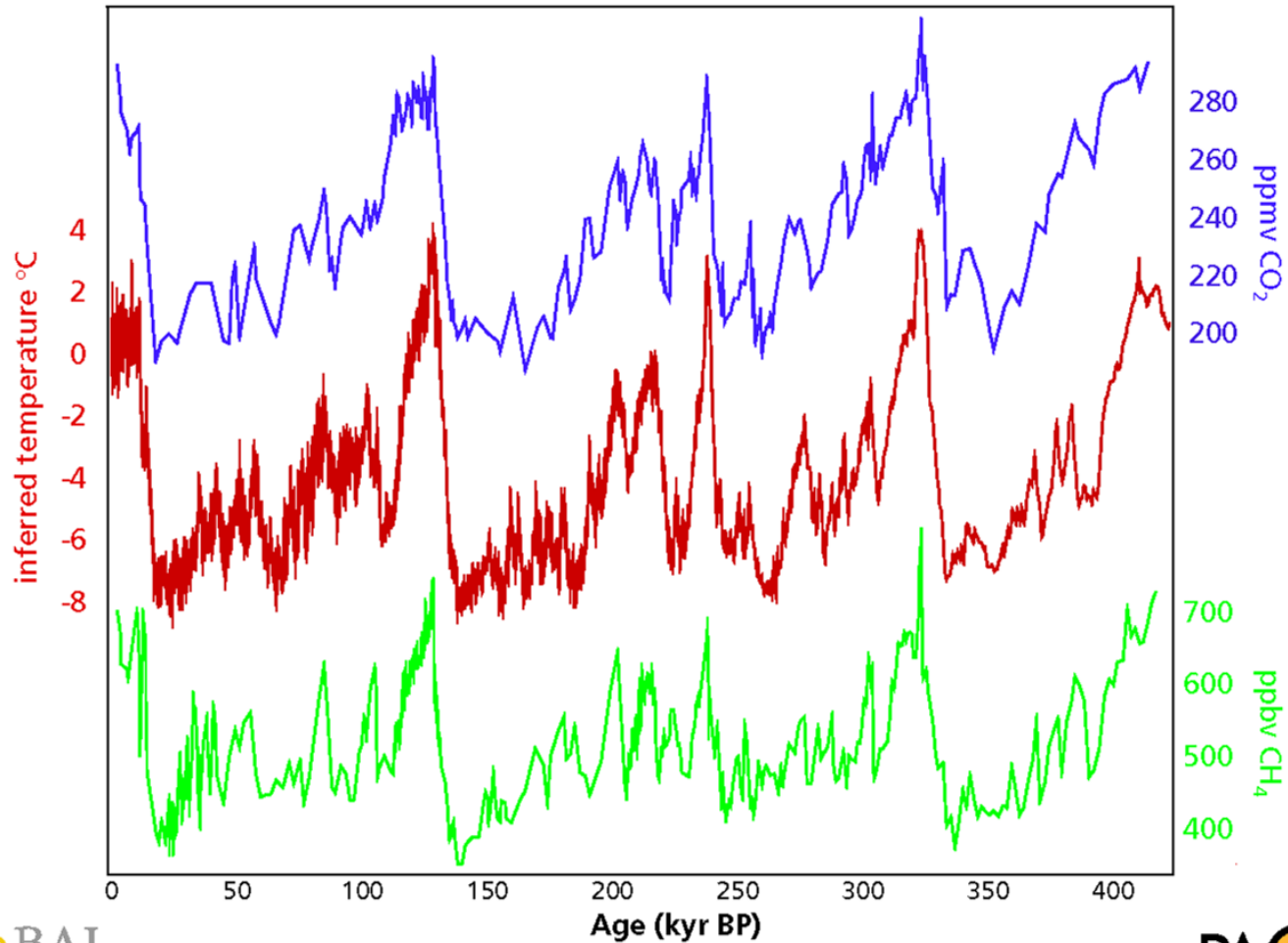
**30% of earth covered with ice
20,000 yrs ago. Today only 11%**

Figure 1: Long-term climatic (Milankovitch) cycles over the last 415,000 years from the Vostok ice core

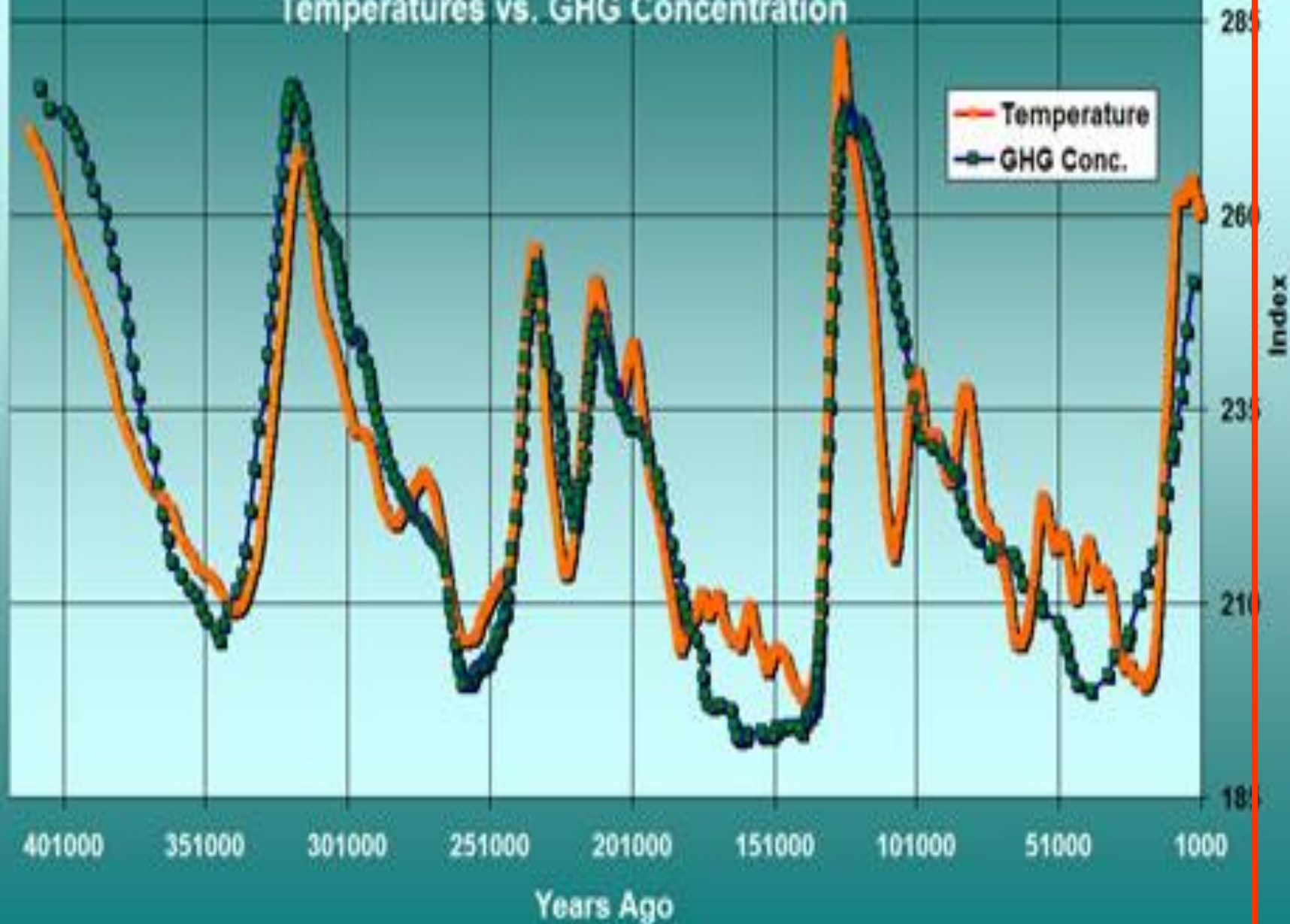


Source: Salamatin A.N., (et al.) 'Ice core age dating and paleothermometer calibration based on isotope and temperature profiles from deep boreholes at Vostok station (East Antarctica)', *Journal of Geophysical Research*, 1998, vol. 103, no D8, pp. 8963-8977.

4 glacial cycles recorded in the Vostok ice core

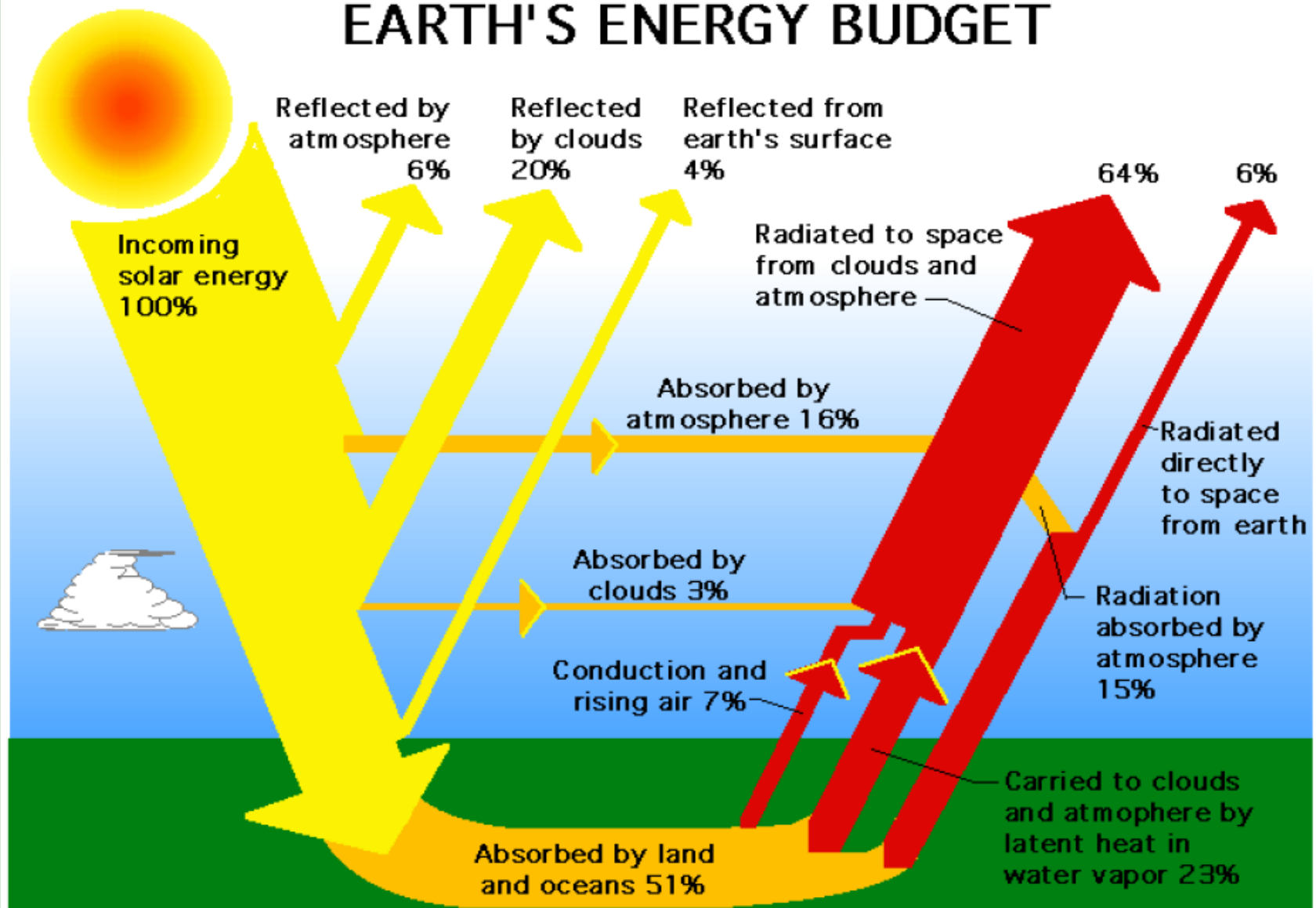


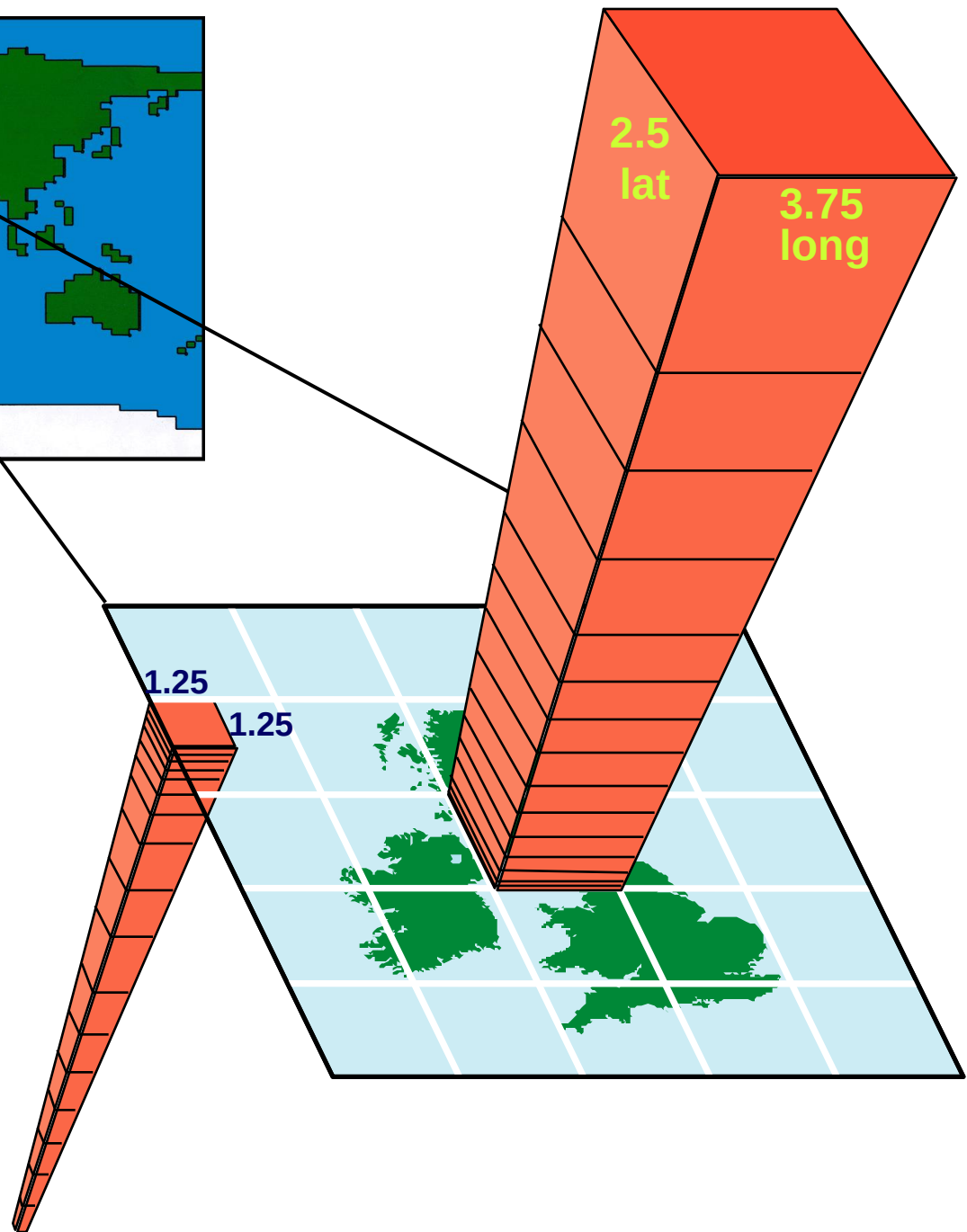
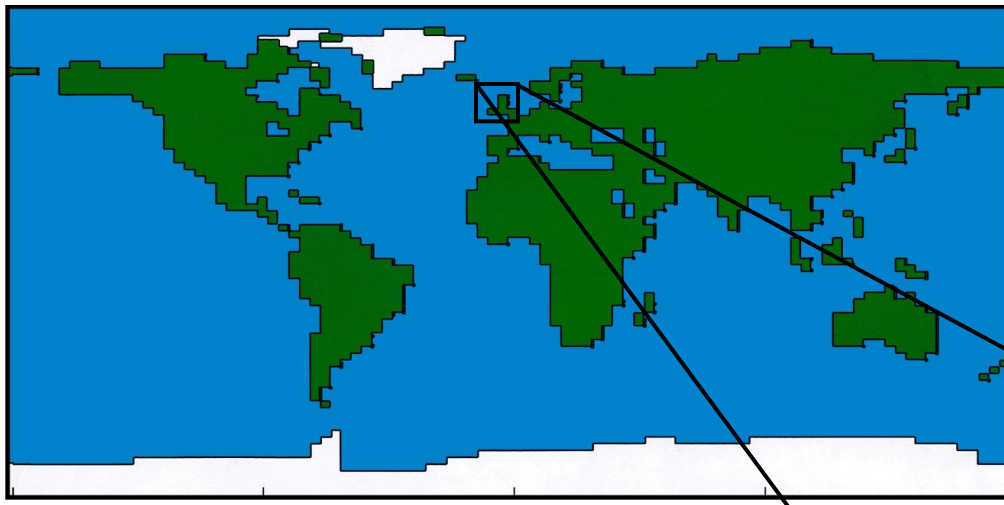
Vostok Ice Core Temperatures vs. GHG Concentration



Global Energy Balance

EARTH'S ENERGY BUDGET

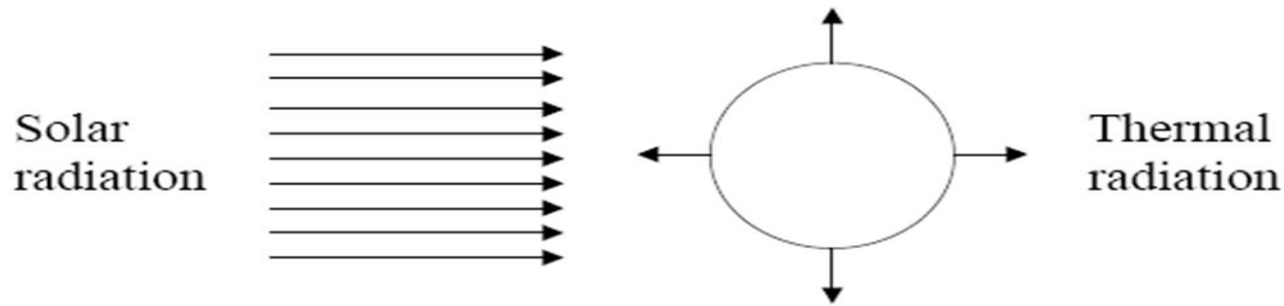




**COUPLED
OCEAN-
ATMOSPHERE
GENERAL
CIRCULATION
MODELS**

**SIMPLE ANALYTICAL
MODEL
FOR
GLOBAL MEAN SURFACE
TEMPERATURE
OF THE EARTH**

Overall energy balance of the Earth



$$(1 - \alpha) S_o \pi r^2 = 4 \pi r^2 \sigma T_{\text{eff}}^4$$

Simplifying, we find that;

$$\sigma T_{\text{eff}}^4 = (1 - \alpha) S_o / 4$$

and hence

$$T_{\text{eff}} \approx 255 \text{ K}$$

for $S = 1365 \text{ W/m}^2$, $\alpha = 0.3$



ABSORBED SOLAR RADIATION = OUTGOING LONGWAVE RADIATION

240 W/m^2

240 W/m^2

Absorbed Solar



Earth's emission

ATMOSPHERE



$\sigma T^4 = 390 \text{ W/m}^2$ for $T_s = 288\text{K}$

Greenhouse effect = $390 - 240 = 150$

GREENHOUSE EFFECT

**GHE = RADIATION EMITTED BY
PLANET'S SURFACE –
RADIATION LEAVING THE
PLANET**

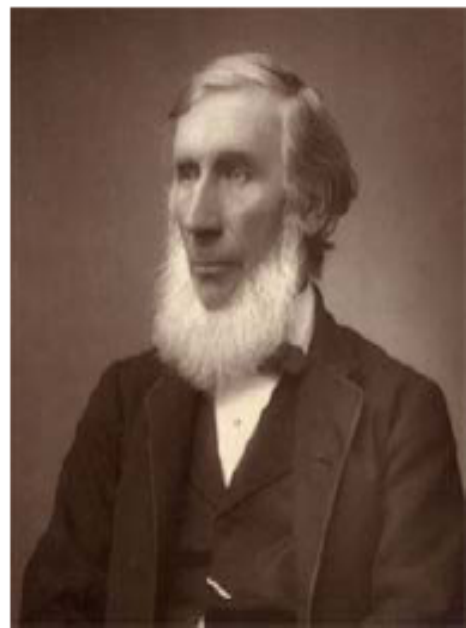
EARTH = 390 - 240 = 250 W/m²

VENUS = 16100 - 200 = 15,900 W/m²

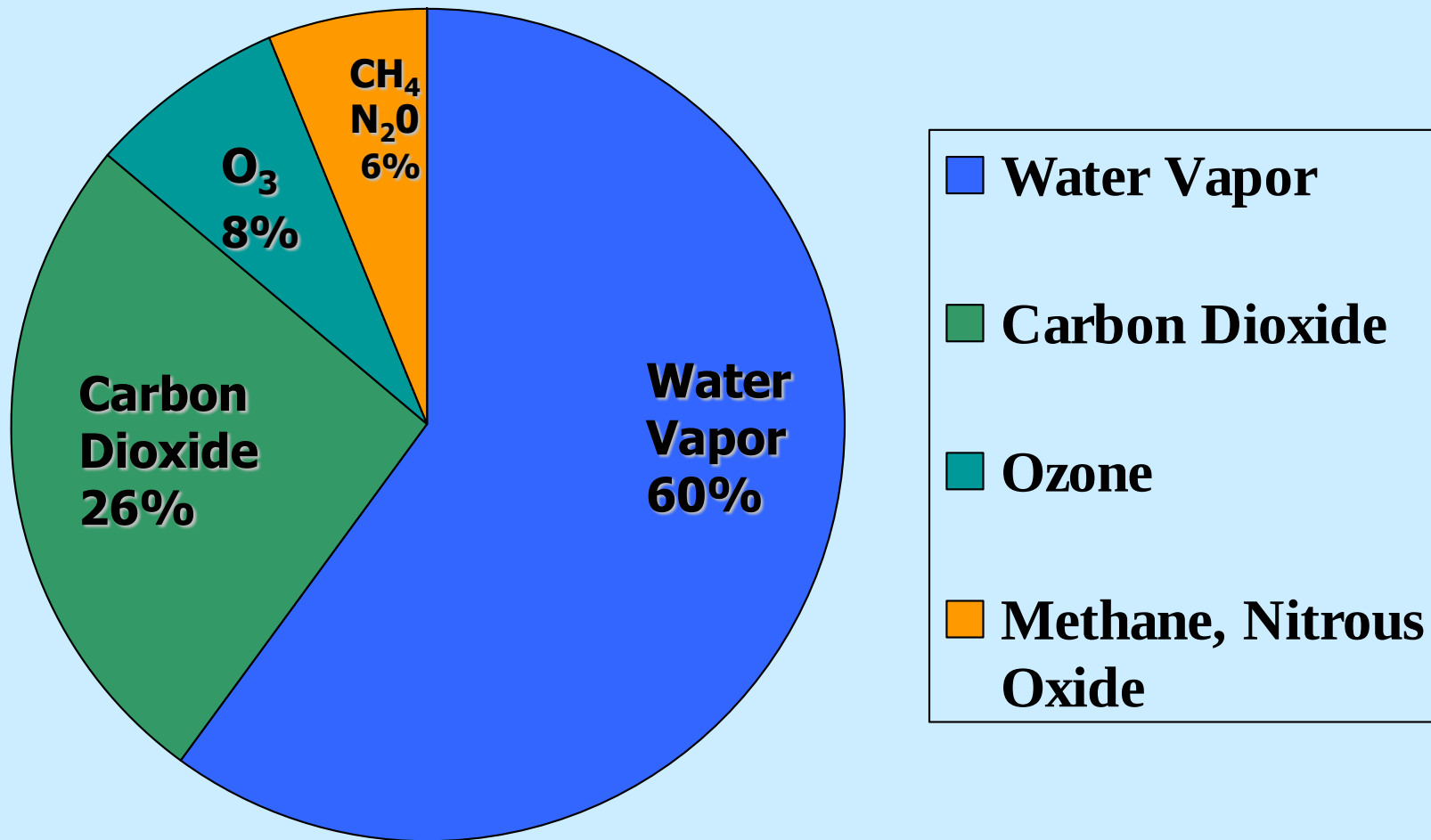
The Greenhouse Effect: Tyndall

John Tyndall*, 1862 (river analogy):

"As a dam built across a river causes a local deepening of the stream, so our atmosphere, thrown as a barrier across the terrestrial rays, produces a local heightening of the temperature at the Earth's surface."



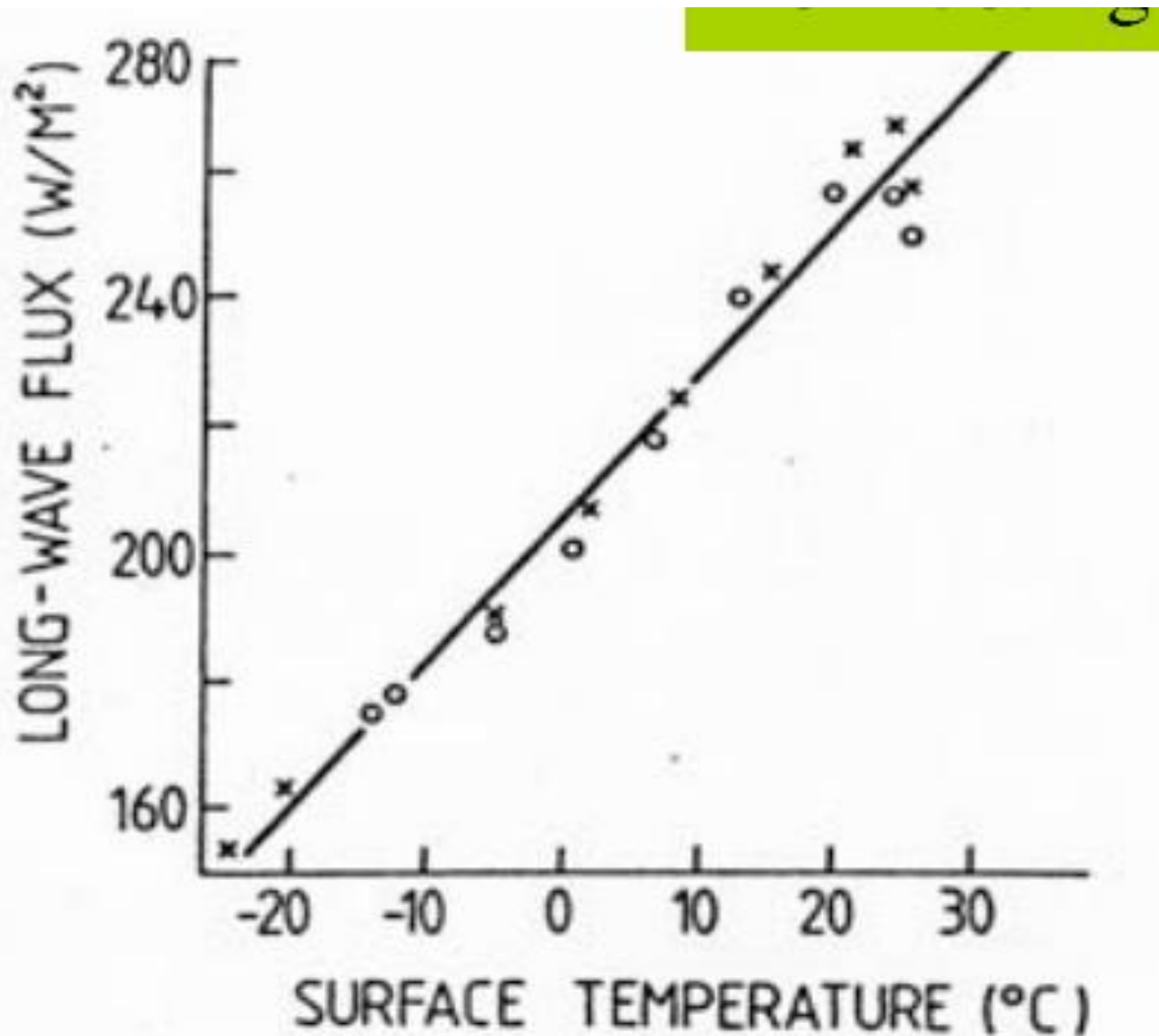
6.2 The Natural Greenhouse Effect: clear sky



Clouds also have a greenhouse effect

Kiehl and Trenberth 1997

LINEARIZATION OF OUTGOING LONGWAVE RADIATION



$$\text{OLR} = A + BT$$

$$S/4 (1 - \alpha) = A + B^*T$$

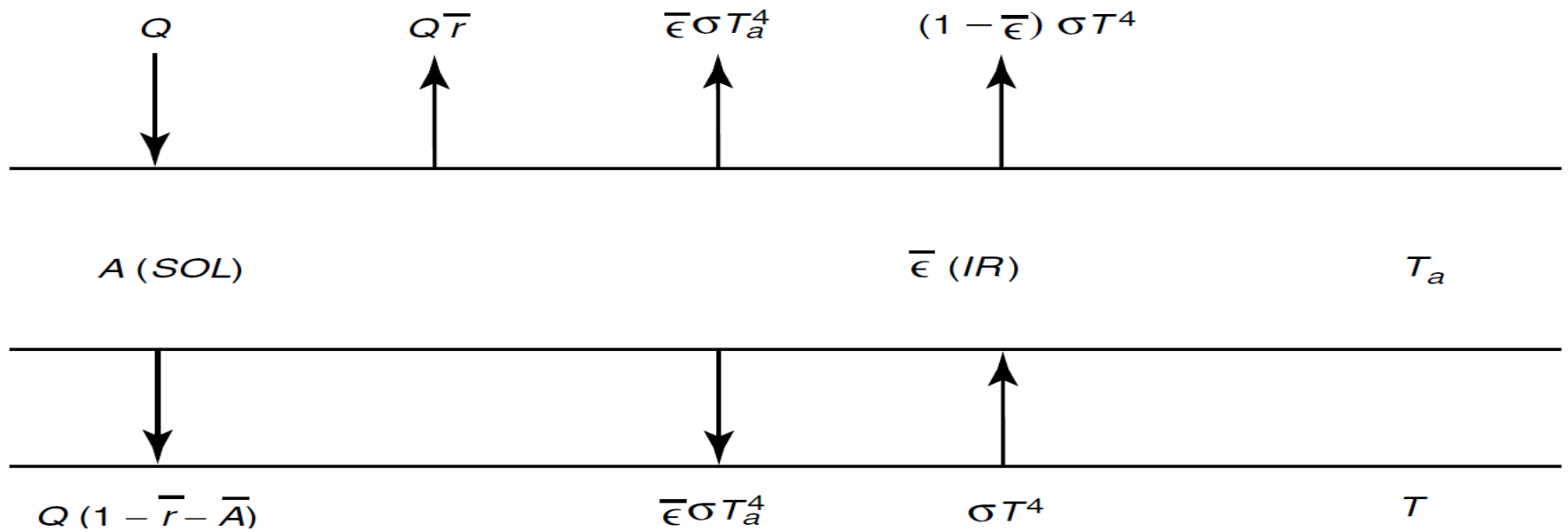
A and B include

Greenhouse effect

$$I = A + BT - 5.397 \ln \left(\frac{c_a}{c_{a,P}} \right)$$

$$\mathbf{A = 203.34 \text{ W /m}^2}$$

$$\mathbf{B = 2.09 \text{ W /m}^2 \text{ K}}$$



$$Q(1 - \bar{r}) - \bar{\epsilon}\sigma T_a^4 - (1 - \bar{\epsilon})\sigma T^4 = 0,$$

$$Q(1 - \bar{r} - \bar{A}) + \bar{\epsilon}\sigma T_a^4 - \sigma T^4 = 0.$$

$$T^4 = Q[2(1 - \bar{r}) - \bar{A}]/[\sigma(2 - \bar{\epsilon})],$$

$$T_a^4 = Q[\bar{A} + \bar{\epsilon}(1 - \bar{r} - \bar{A})]/[\sigma\bar{\epsilon}(2 - \bar{\epsilon})].$$

Where $Q = S/4$, r = albedo

A = solar absorptivity ϵ = atmos emissivity

For $r=0.3$, $A=0.2$, $\epsilon= 0.95$

$T = 288 \text{ K}$ close to the observation

Limiting Cases :

No atmosphere : $A= 0$ and $\epsilon= 0$

$$T^4 = S/4 \{ 1 - r \} \text{ for } r = 0.3 \quad T = \mathbf{255 \text{ K}}$$

No atm solar absorp, $\epsilon = 1$, & $r=0.3$

$$T^4 = S/4 \{ 2 (1 - r) \} \quad T = \mathbf{303 \text{ K}}$$

SENSITIVITY

$$\frac{1}{T_s} \frac{\partial T_s}{\partial S} = \frac{1}{4S}$$

$$\Delta S = .01, \Delta T_s = 0.75K$$

$$\frac{1}{T_s} \frac{\partial T_s}{\partial A} = \frac{-S/4}{4\sigma T_s^4 (2 - \bar{\epsilon})}$$

$$\Delta A = 0.01 \Delta T_s = -0.60K$$

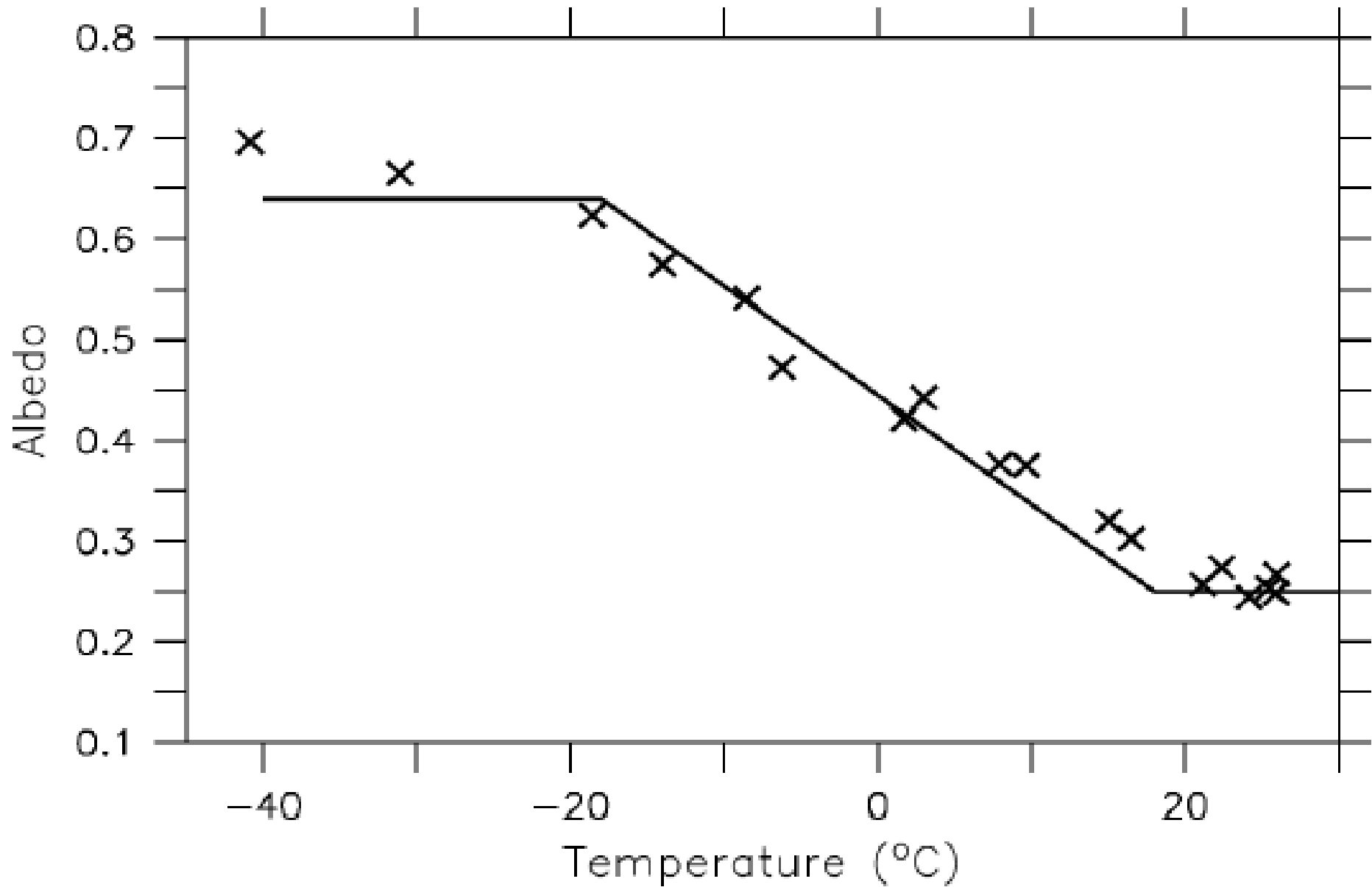
$$\frac{1}{T_s} \frac{\partial T_s}{\partial \bar{n}} = \frac{-S/2}{4\sigma T_s^4 (2 - \bar{\epsilon})}$$

$$\Delta r = 0.01 \Delta T_s = -1.20K$$

$$\frac{1}{T_s} \frac{\partial T_s}{\partial \bar{\epsilon}} = \frac{1}{4(2 - \bar{\epsilon})}$$

$$\Delta \epsilon = 0.01 \Delta T_s = 0.75K$$





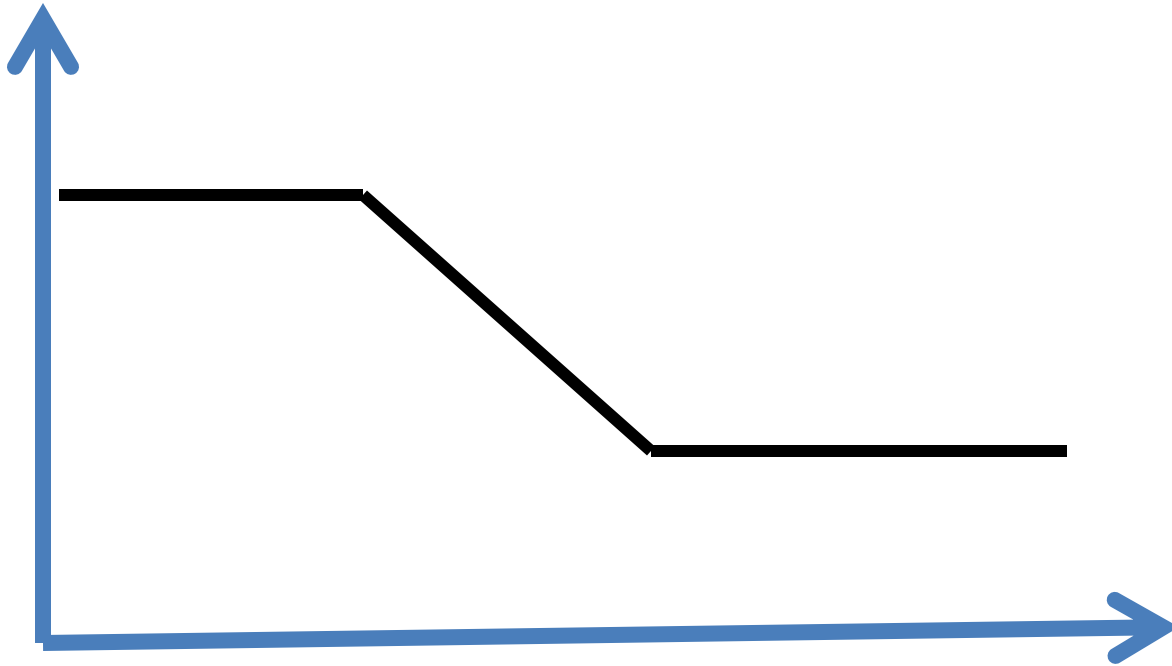
ALBEDO FROM SATELLITE (ERBE) DATA

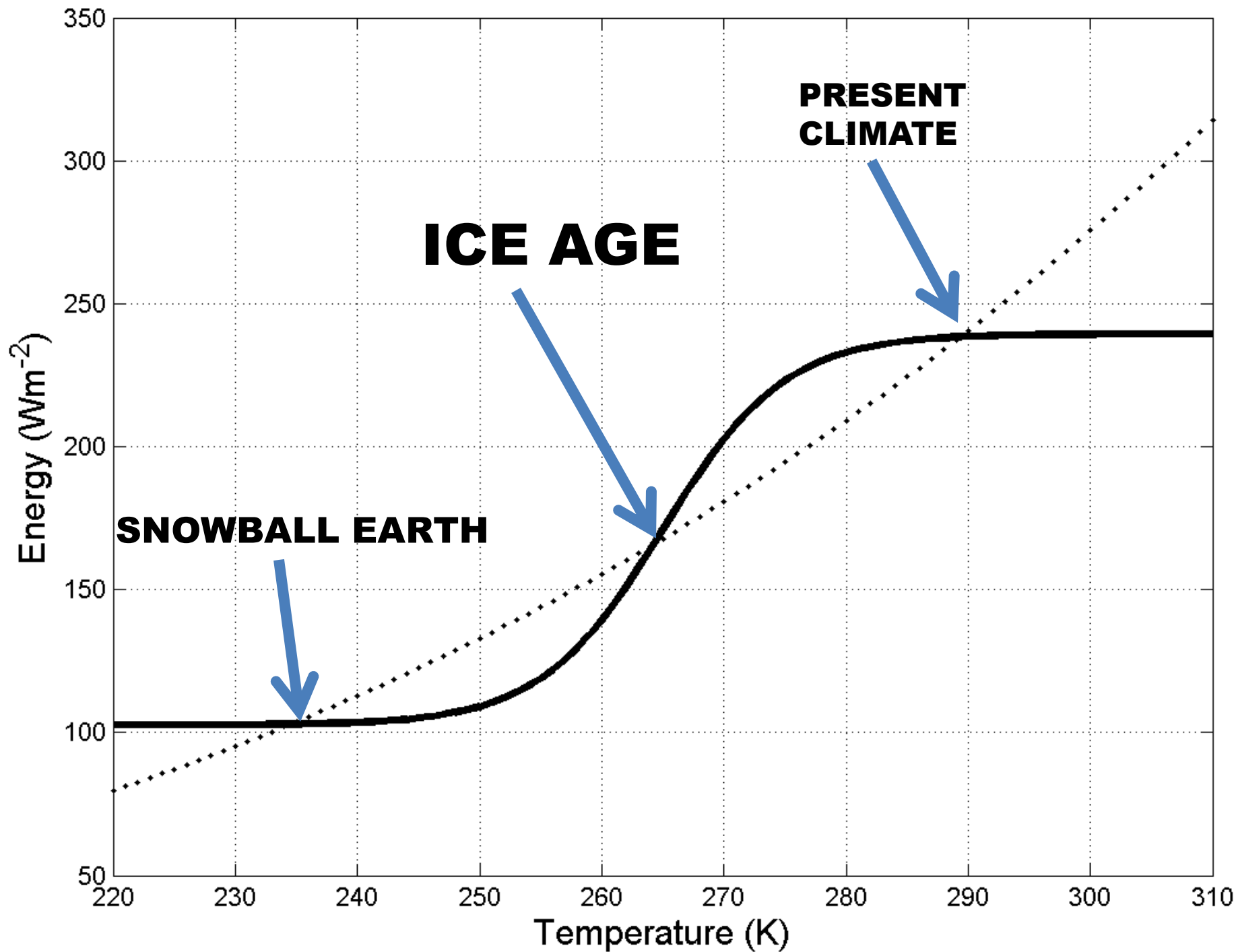
$$S/4 \{1 - \rho(T)\} = A + B * T$$

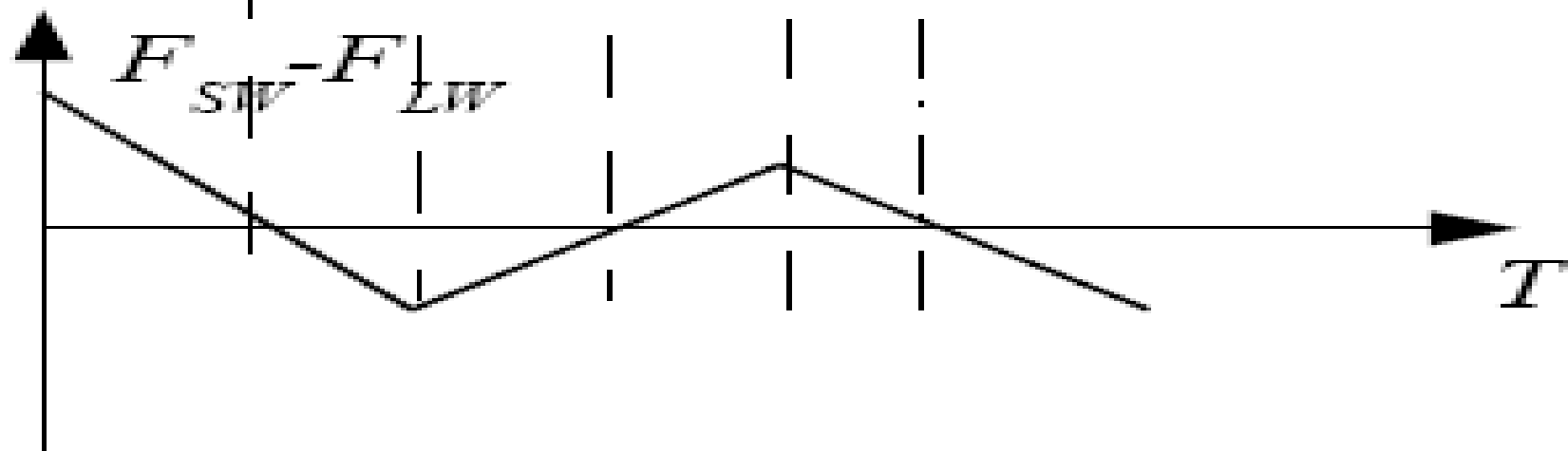
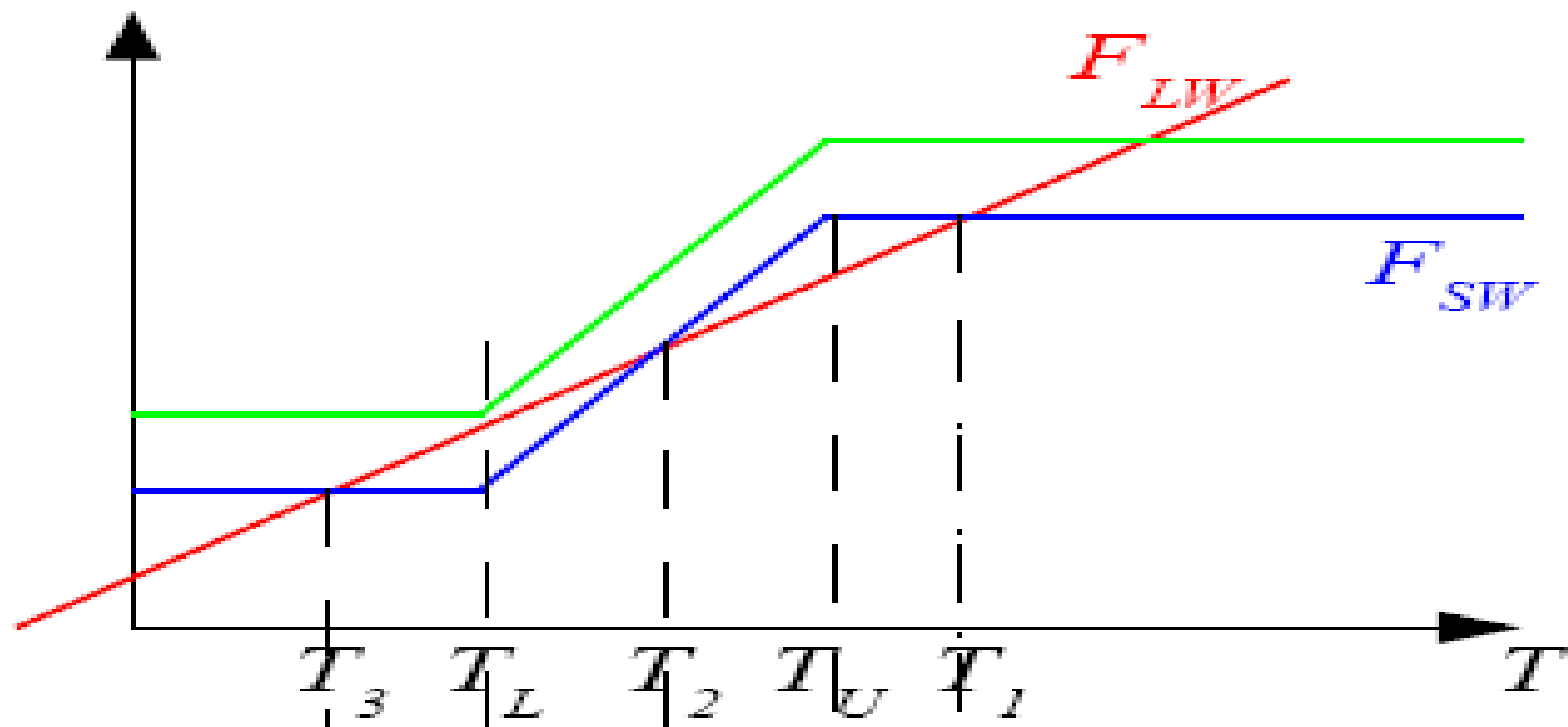
$$\rho(T) = \rho_0 \text{ for } T < T_0$$

$$\rho(T) = \rho_1 \text{ for } T > T_1$$

$$\rho(T) = \rho_0 + (\rho_1 - \rho_0)(T - T_0)/(T_1 - T_0)$$



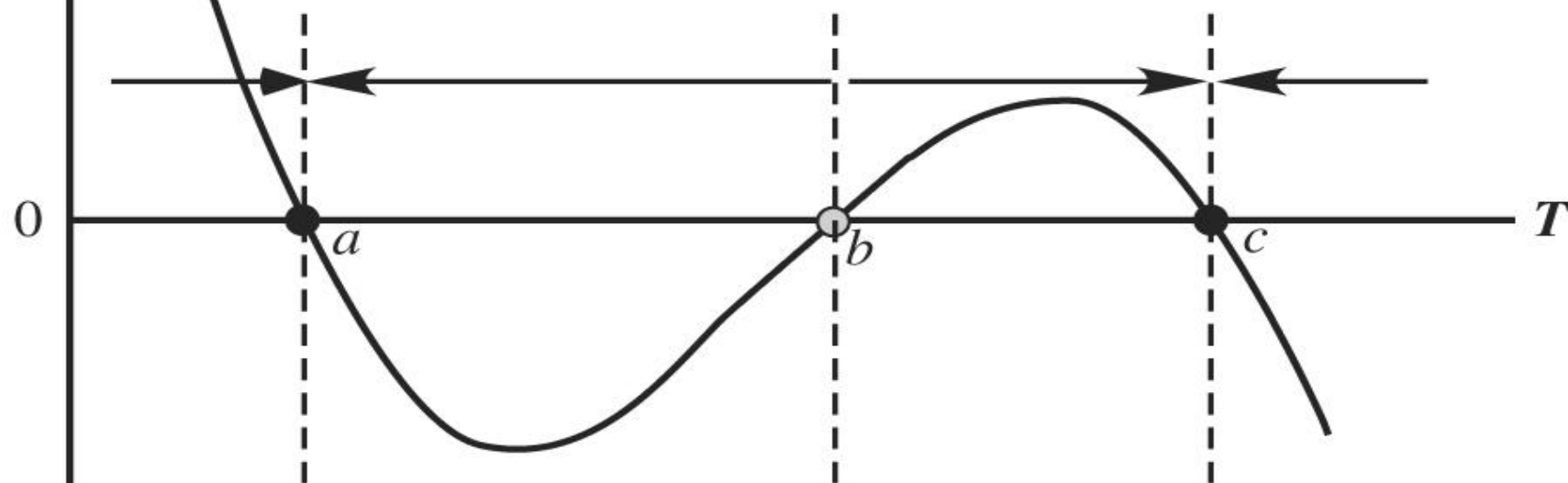


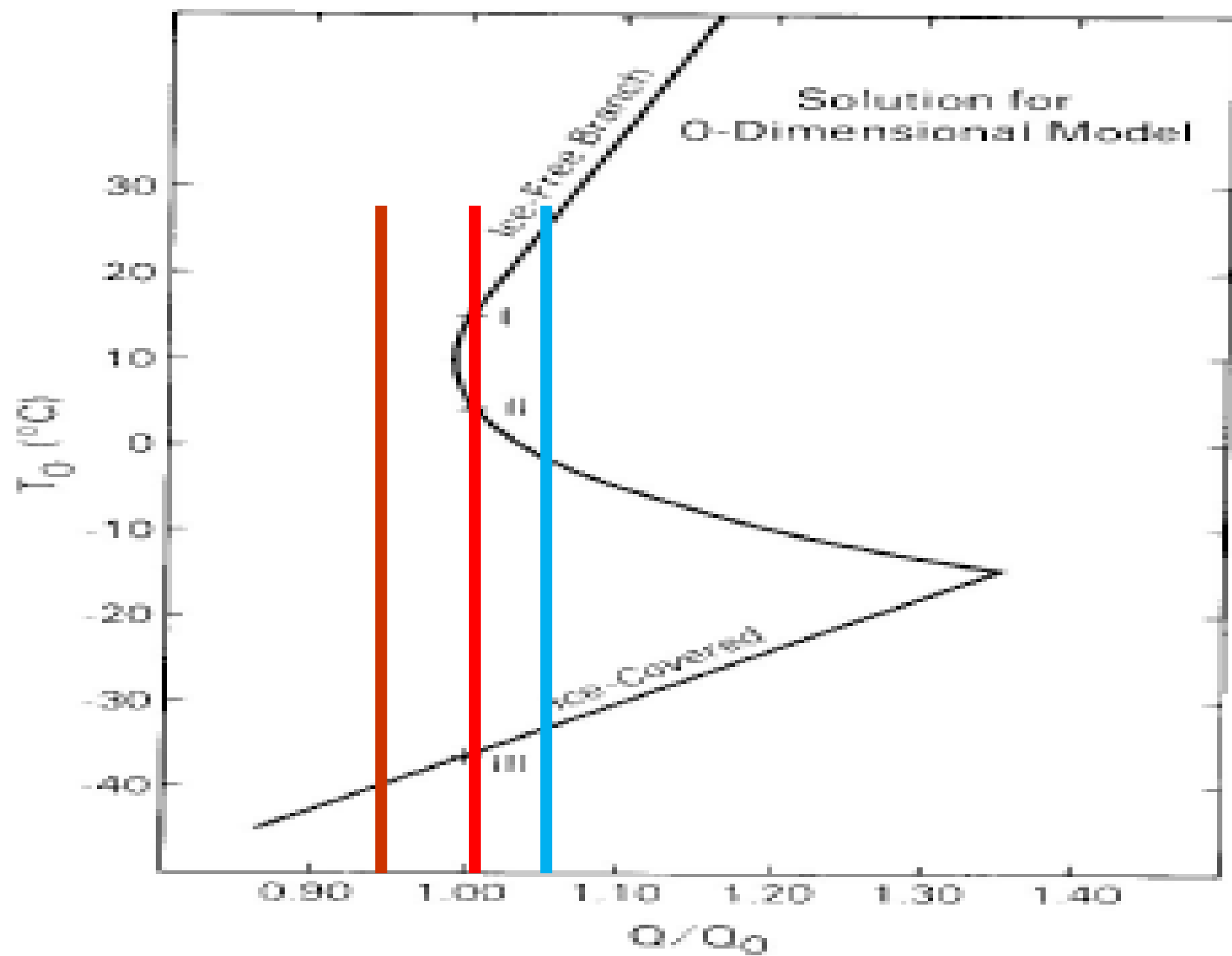


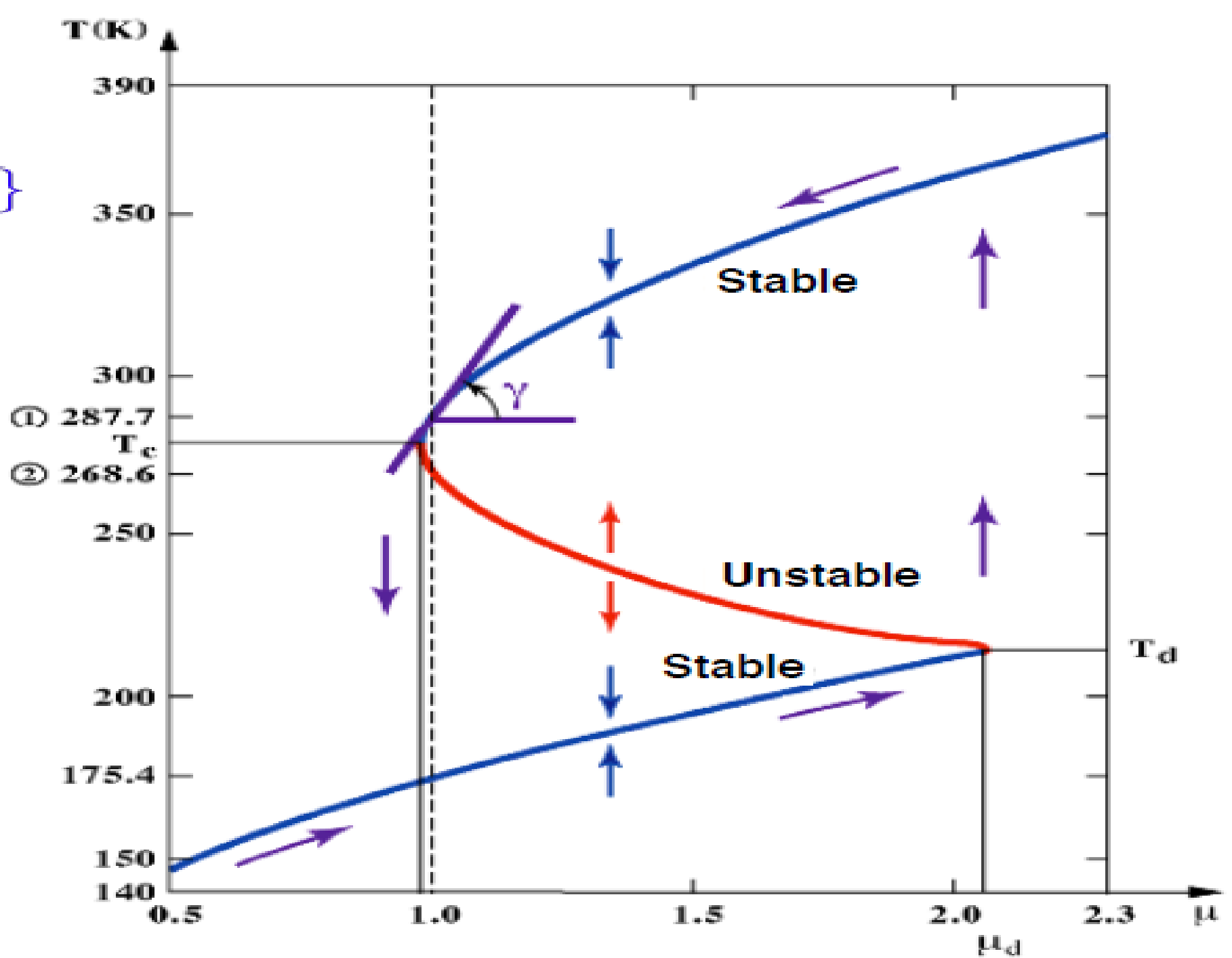
Net flux

○ Unstable equilibrium

● Stable equilibrium







Stability characteristics

The stability of these equilibria can be investigated by **linearizing** the equation

$$C \frac{dT}{dt} = Q(1 - \alpha(T)) - a * -b * T$$

C is the heat capacity of the system per unit surface area.

This is a nonlinear first order differential equation. The standard method of investigating the stability of the equilibrium solution is to add a perturbation:

$$T(t) = T_{eq} + T'(t)$$

The albedo may be written as:

$$\alpha(T) \cong \alpha(T_{eq}) + \frac{d\alpha}{dT} T'$$

Inserting the above two equations into the top equation we obtain

$$\frac{dT'}{dt} = -\lambda T' \quad \text{with} \quad \lambda = \frac{b^* + Q \frac{d\alpha}{dT}}{C}$$

The solution of this differential equation is $T' = T'(0)\exp(-\lambda t)$

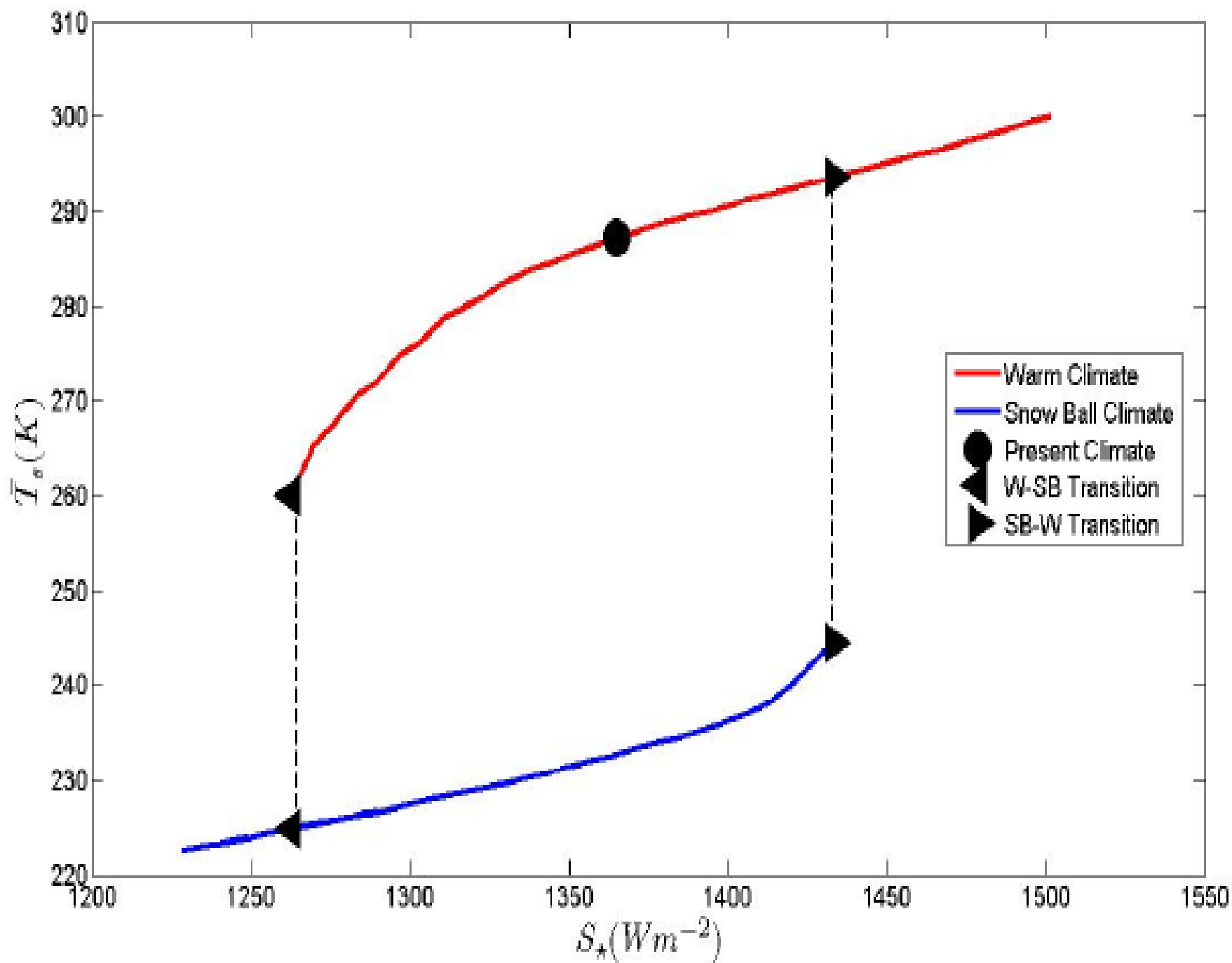
Therefore there is decay back to equilibrium if $\lambda > 0$ and **exponential growth of the amplitude of the perturbation (instability) if $\lambda < 0$** .

The sign of λ is determined by the sign of $d\alpha/dT$.

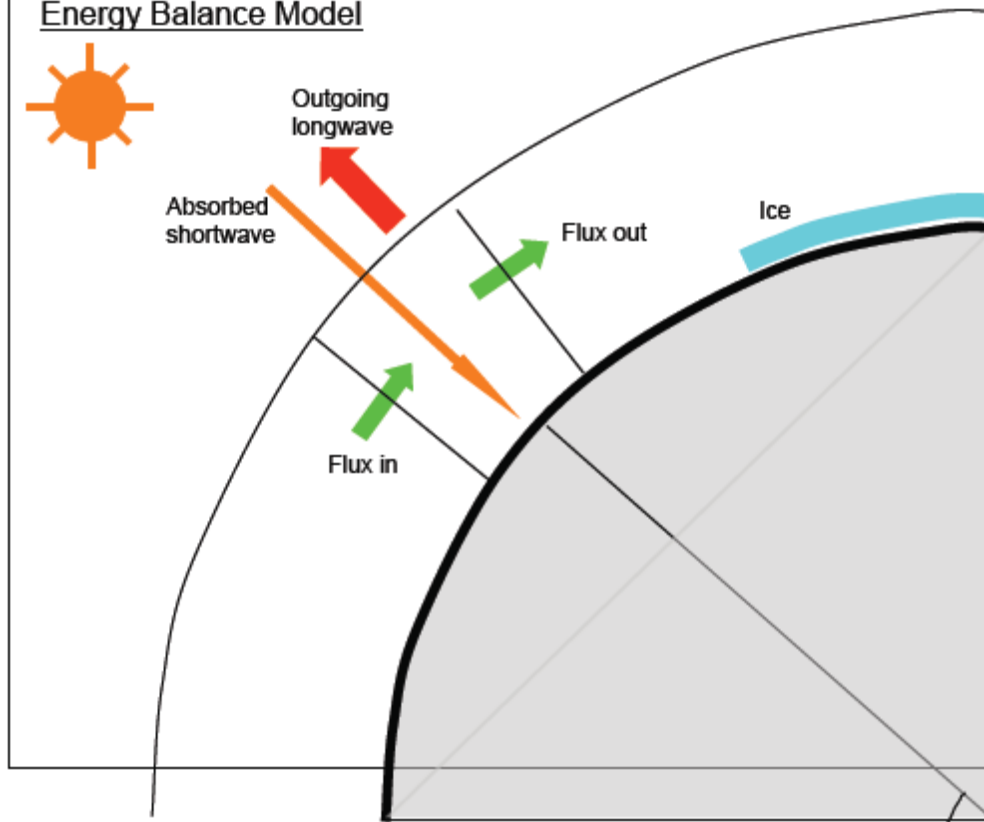
For $T > T_1$ or $T < T_0$ $d\alpha/dT = 0$. Therefore, the solution for these equilibria are stable (because $b^* > 0$).

For $T_0 < T < T_1$ $d\alpha/dT < 0$. Therefore, the solution for this intermediate equilibrium state is unstable if

$$\frac{d\alpha}{dT} < \frac{b^*}{O}$$



Energy Balance Model



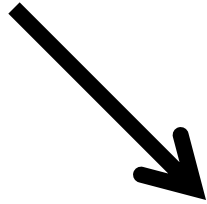
$$\underbrace{Q_0 / 4(1 - \alpha) S(\phi)}_{\text{Net heat absorbed from the sun}} = \underbrace{(A + BT)}_{\text{Longwave emitted to space}} - \underbrace{D \frac{d^2 T}{dy^2}}_{\text{Heat divergence: net heat transported out of the box by atmosphere and ocean.}}$$

Net heat
absorbed
from the sun

Longwave
emitted to
space

Heat divergence: net
heat transported out
of the box by
atmosphere
and ocean.

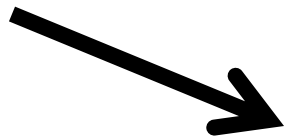
$$-\frac{d}{dx} D(1-x^2) \frac{dT(x)}{dx} + A + BT(x) = QS(x)a(x, x_s)$$



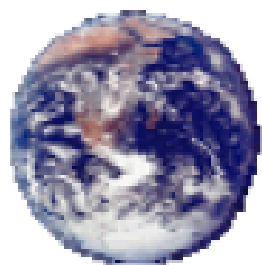
DIFFUSIVE

OR

CONVECTIVE



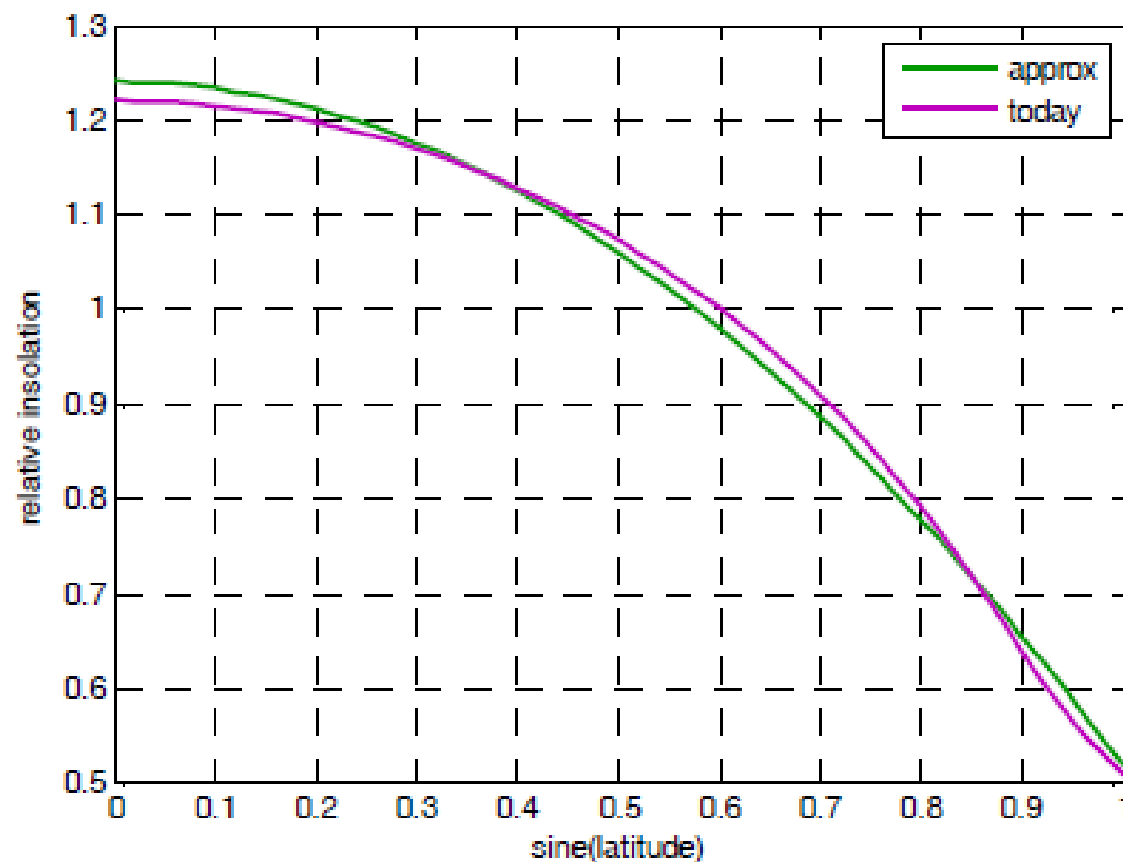
$$\gamma[T(x) - T_0]$$



Energy Balance Models

Inhomogeneous Earth

Relative Insolation Function



green = quadratic
approximation (Tung
and North)

mauve = formula using
obliquity of 23.5°

Variation of incident solar radiation with latitude

$$341 (0.53 + 0.71 \cos^2 \theta)$$

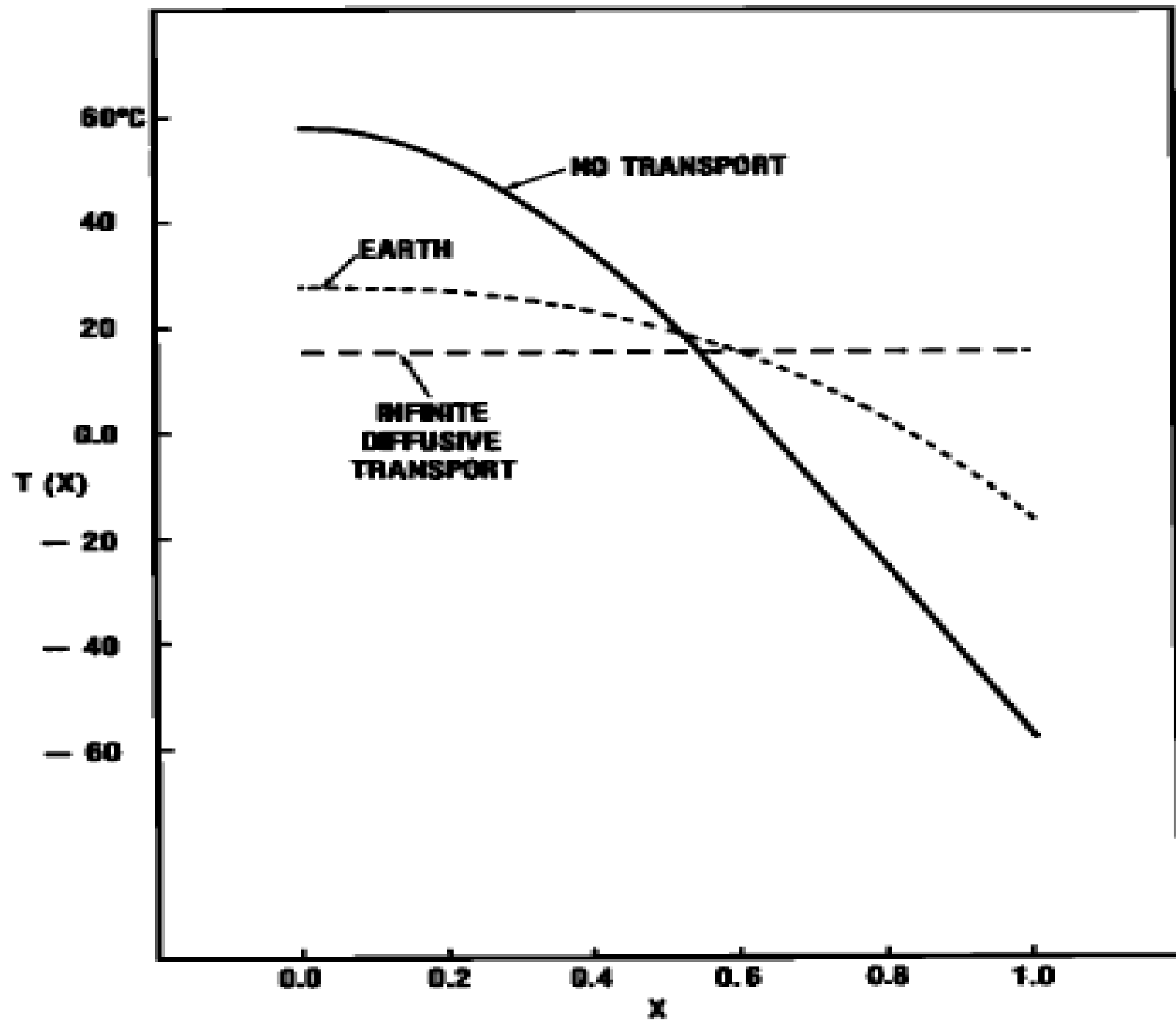
$$\text{Albedo} = 0.6 - 0.4 \cos \theta$$

$$\text{OLR} = 216 + 1.58 T$$

$$\text{Meridional Heat transfer} = C (T - T_{av})$$

$$C = 3.8 \text{ W/m}^2 \text{ } ^\circ\text{C}$$

$$T = \{ 341 (0.53 + 0.71 \cos^2 \theta)(0.4 + 0.4 \cos \theta) - A - C \} / \{ A + B \}$$



$$-\frac{d}{dx}D(x)(1-x^2)\frac{d}{dx}T(x) + A + BT(x) = QS(x)a(x)$$

Boundary Conditions:

$$\left. \sqrt{(1-x^2)} \frac{d}{dx}T(x) \right|_{x=0,1} = 0$$

(No Heat Flux into Poles or across Equator)

$S(x)$ =Distribution of Solar Insolation

Very Simple Solution:

$$\text{If } a(x) = a_0$$

Legendre Polynomial
 $P_2(x)$



$$\text{Then } T(x) = T_0 + T_2 \cdot \frac{1}{2}(3x^2 - 1)$$

$$T_0 = \frac{Qa_0 \cdot 1 - A}{B}, \quad T_2 = \frac{Qa_0 S_2}{6D + B}$$

$$T_0 \approx 15^\circ \text{C}, \quad T_2 \approx -28^\circ \text{C}$$



Energy Balance Models

Homogeneous Earth

$$R \frac{dT}{dt} = Q(1 - \alpha) - (A + BT)$$

T = global mean temperature (°C)

Q = mean solar input (W/m²)

α = mean albedo

$A + BT$ = outward radiation (linear approximation)

R = heat capacity of Earth's surface

Tung's values:

T = global mean temperature (°C)

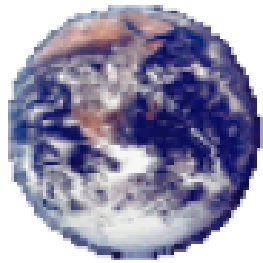
Q = 343 W/m²

A = 202 W/m²

B = 1.9 W/(m² °C)

$\alpha = \alpha_1 = 0.32$ (water and land)

$\alpha = \alpha_2 = 0.62$ (ice)



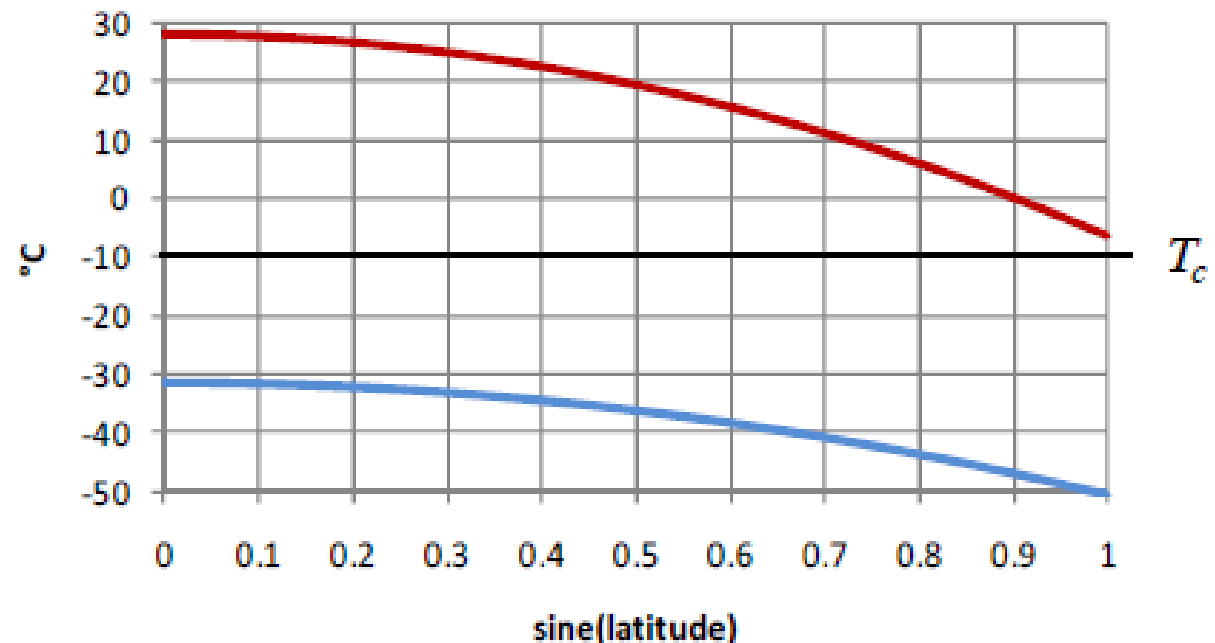
Energy Balance Models

Inhomogeneous Earth

Assume that $\alpha(y) = \alpha_i$ (constant).

$$T^*(y) = \frac{1}{B+C} \left(Q_s(y)(1-\alpha_i) - A + C\overline{T^*} \right)$$

Surface Temperature



Both ice-free
($\alpha = \alpha_1 = 0.32$)
and snowball
($\alpha = \alpha_2 = 0.62$)
states look pretty
stable.

— ice free — snowball

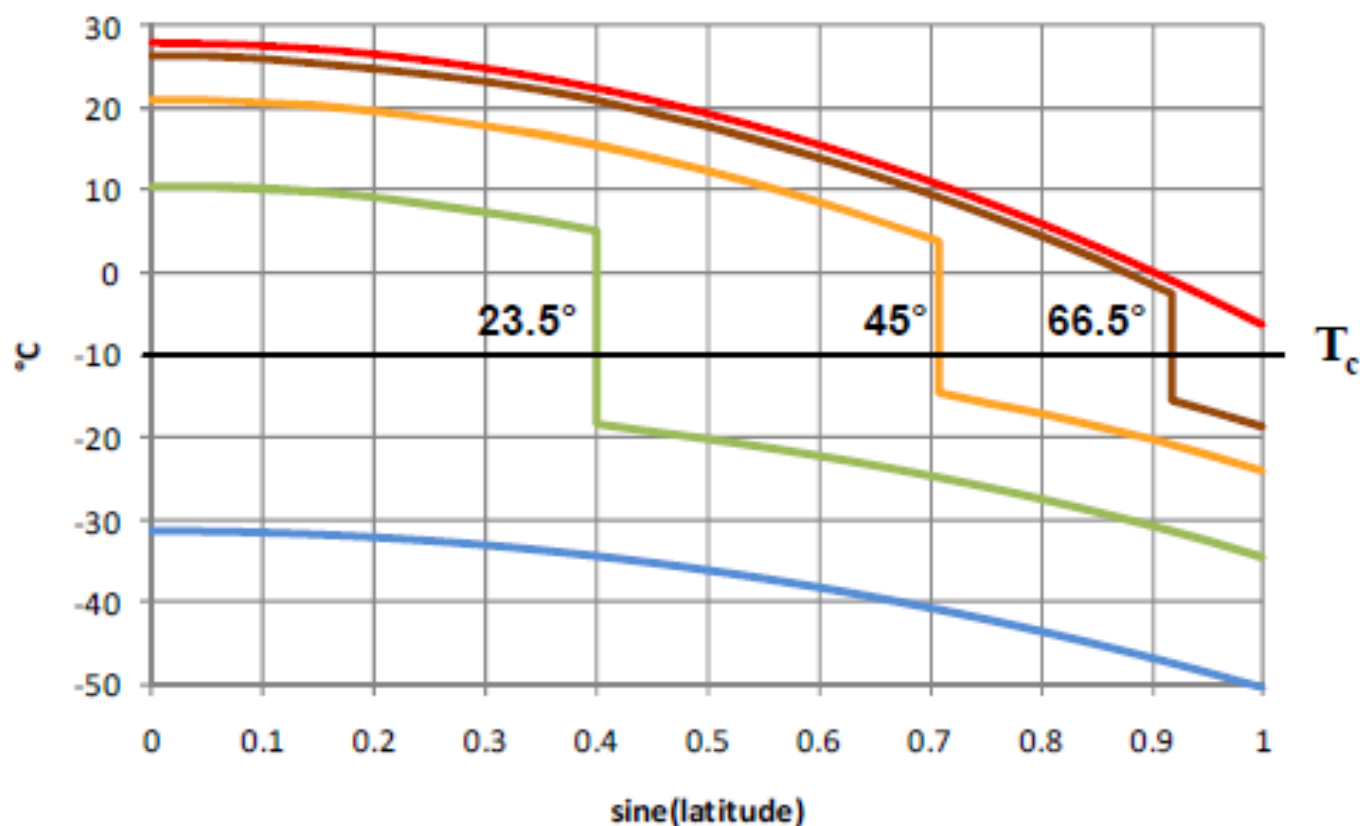


Energy Balance Models

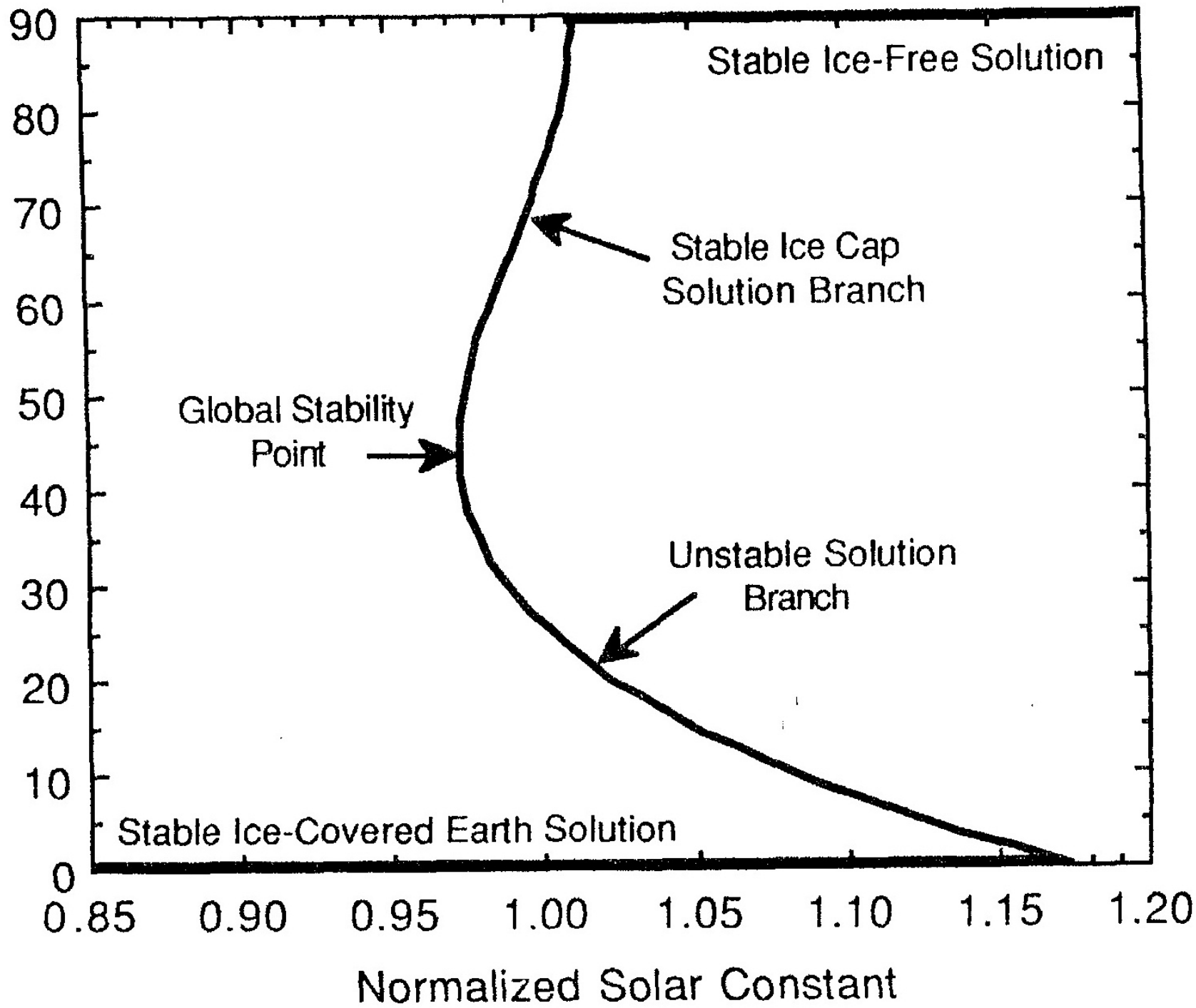
Inhomogeneous Earth

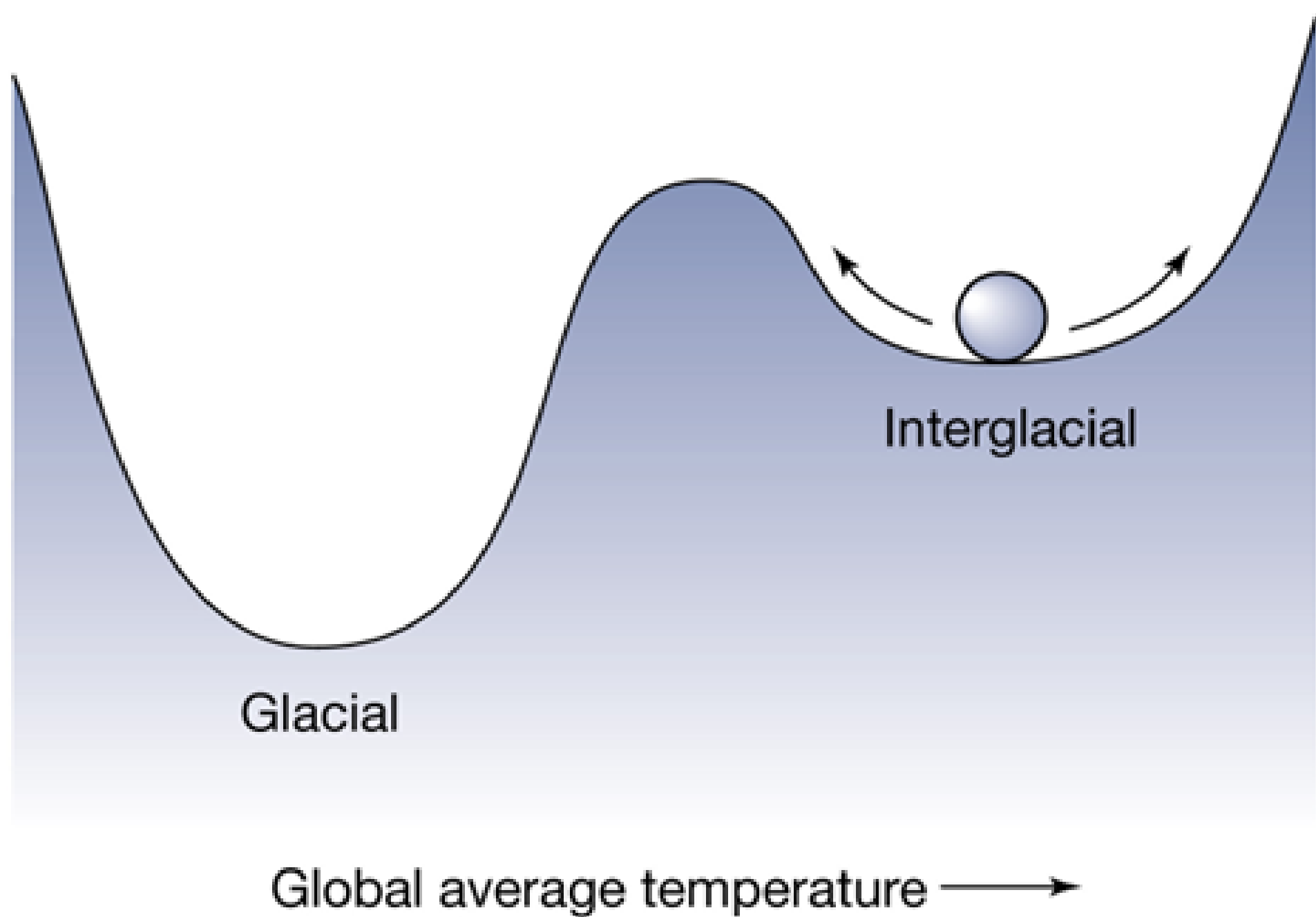
Temperature profiles for various ice lines

The standard argument seems to imply that these are all stable.

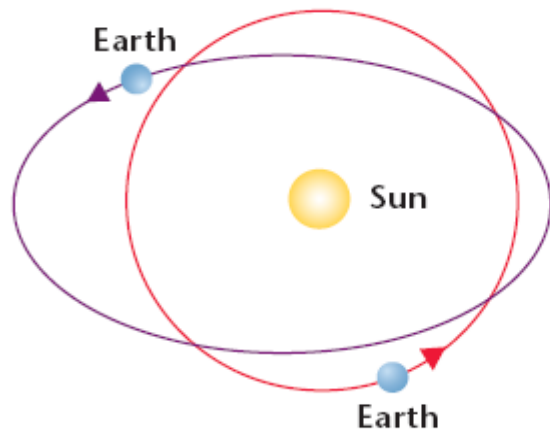


Latitude of Ice Boundary

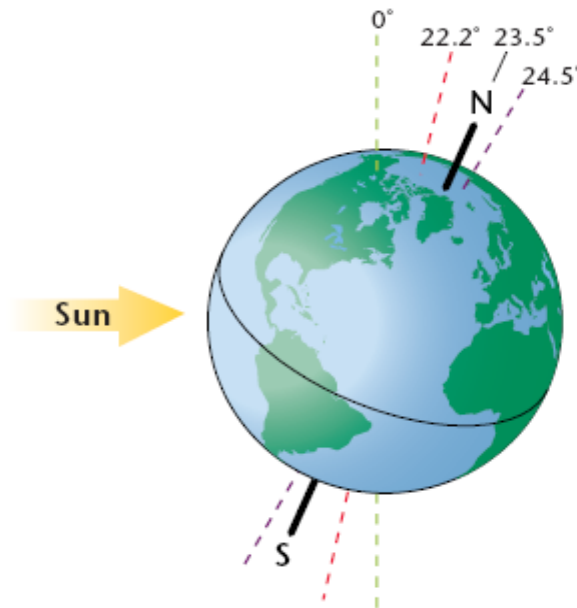




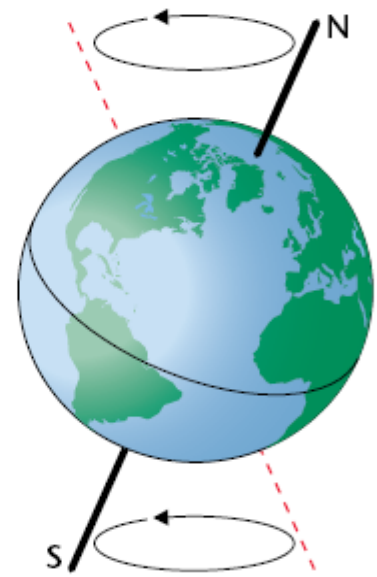
Changes in Earth's ORBIT around the Sun can Change the Climate



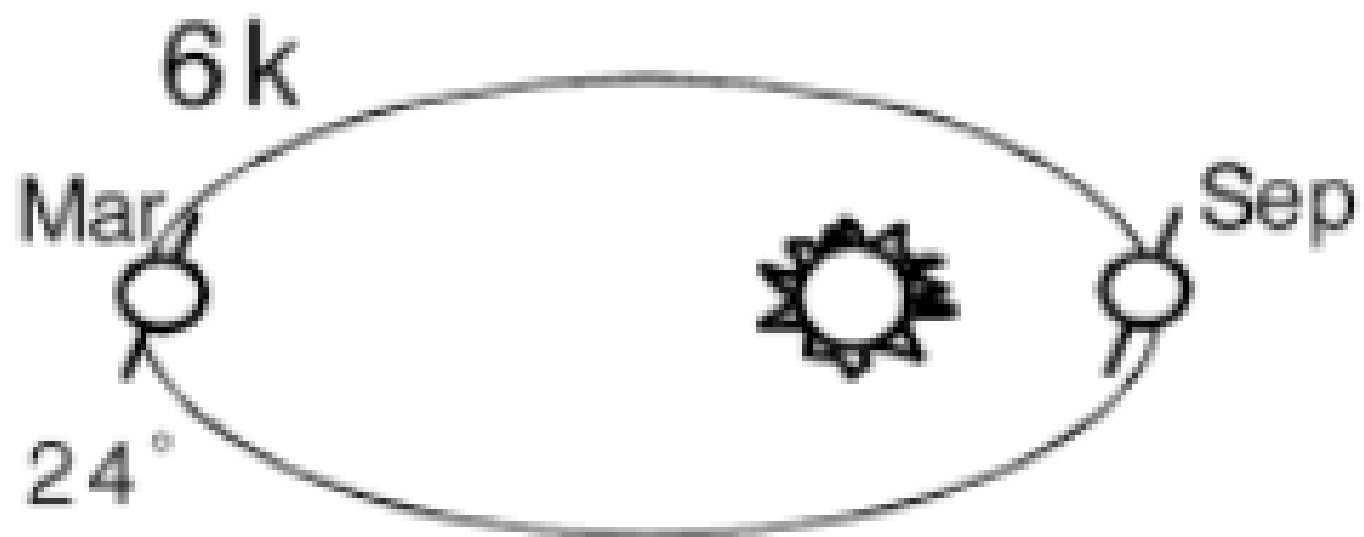
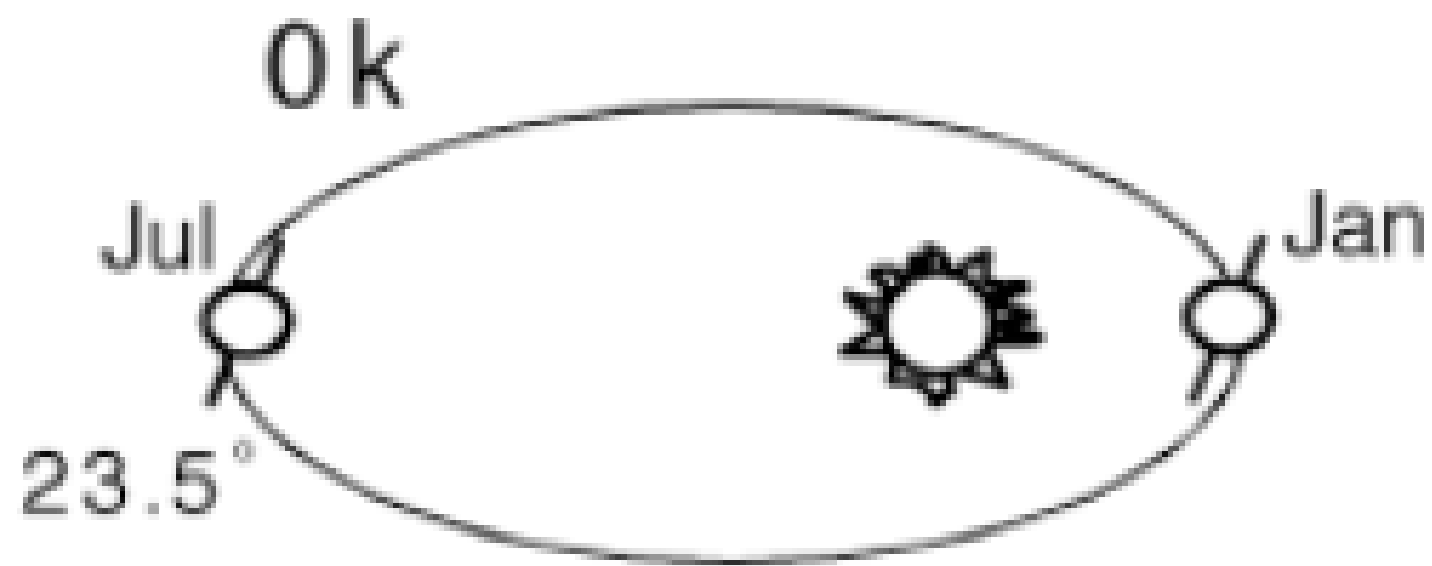
Eccentricity Changes in orbital eccentricity cause an increase in seasonality in one hemisphere and reduce seasonality in the other hemisphere.

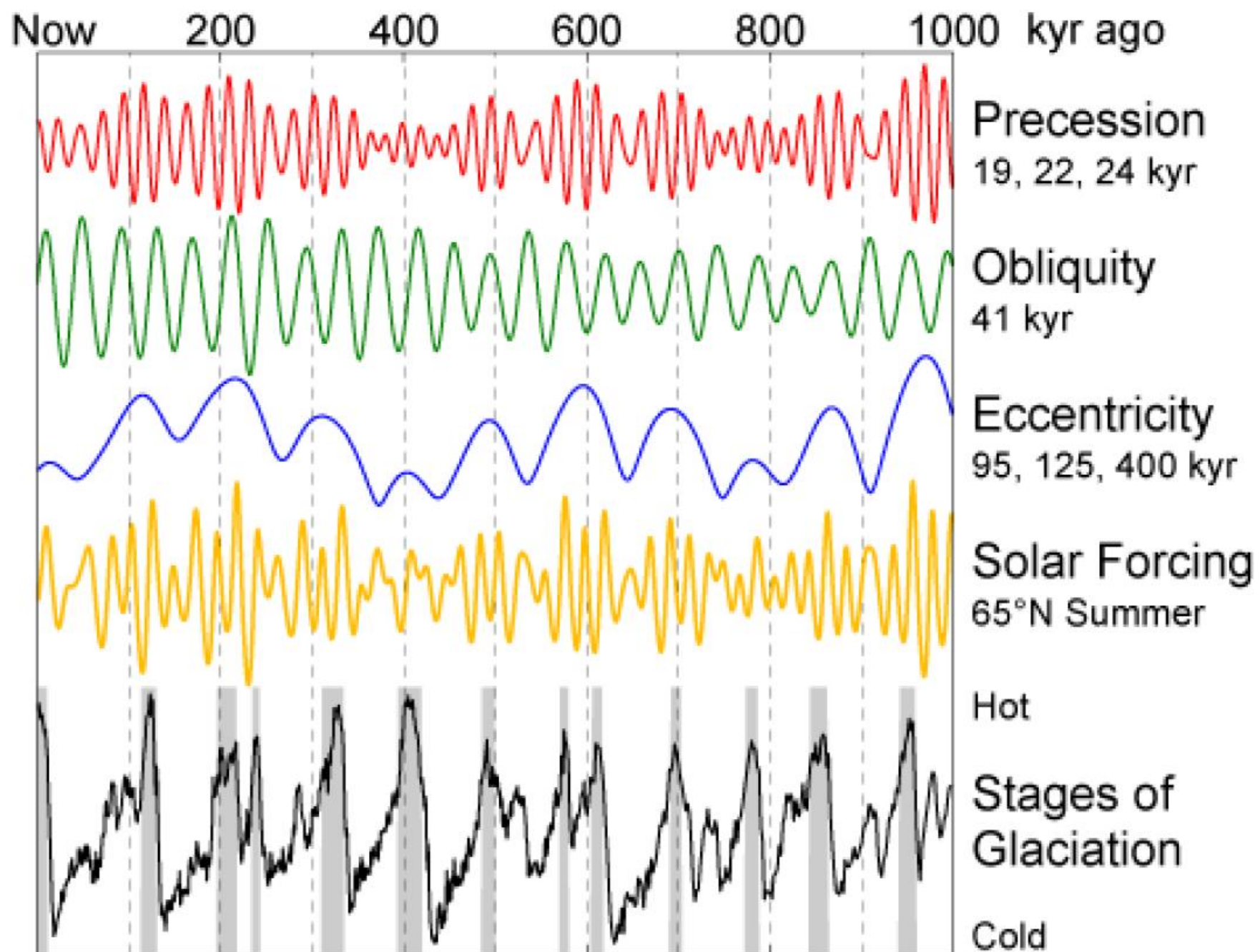


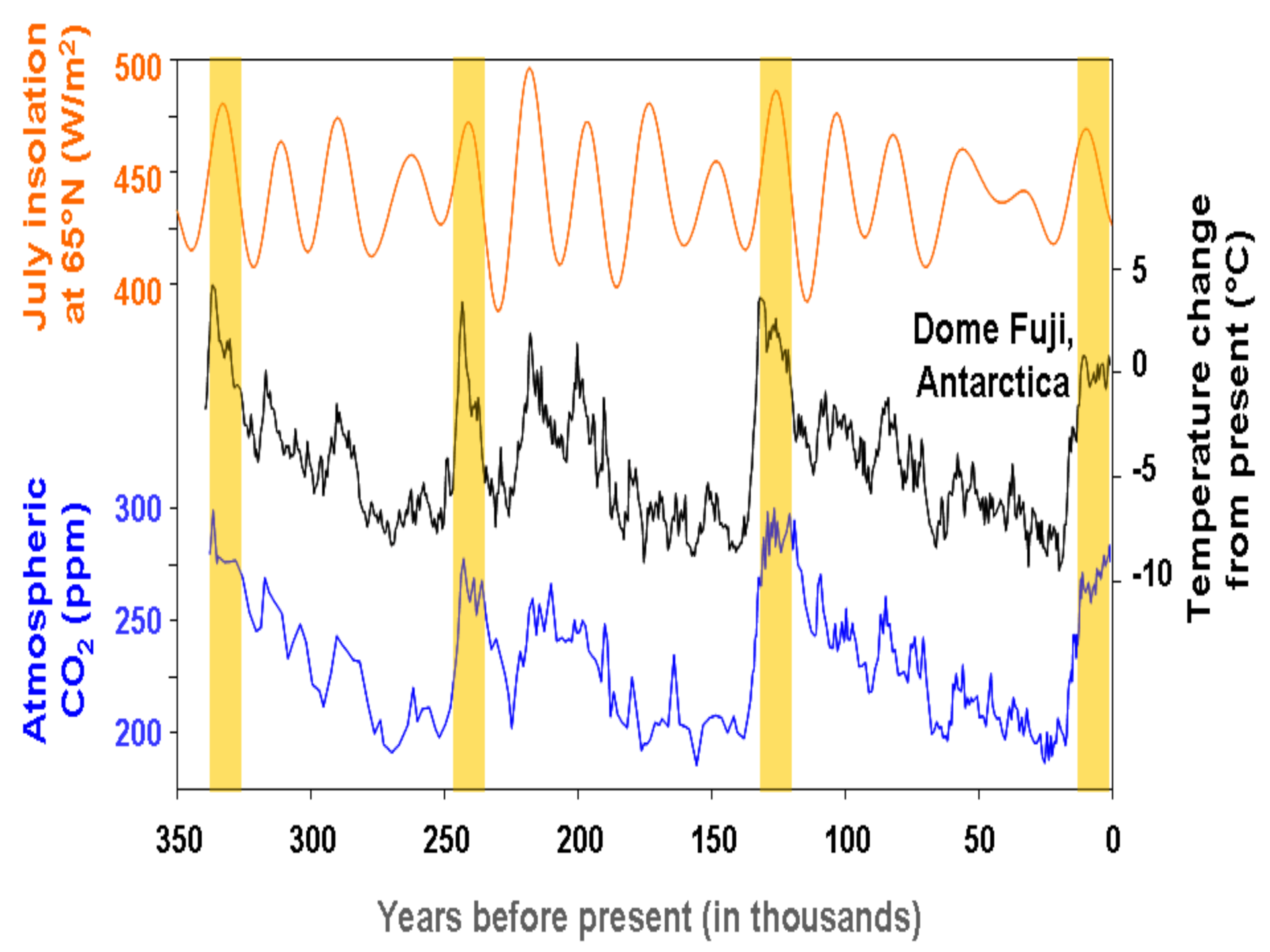
Tilt Over a period of 41,000 years, the tilt of Earth's axis varies between 22.2° and 24.5° . The poles receive more solar energy when the tilt angle is greater.

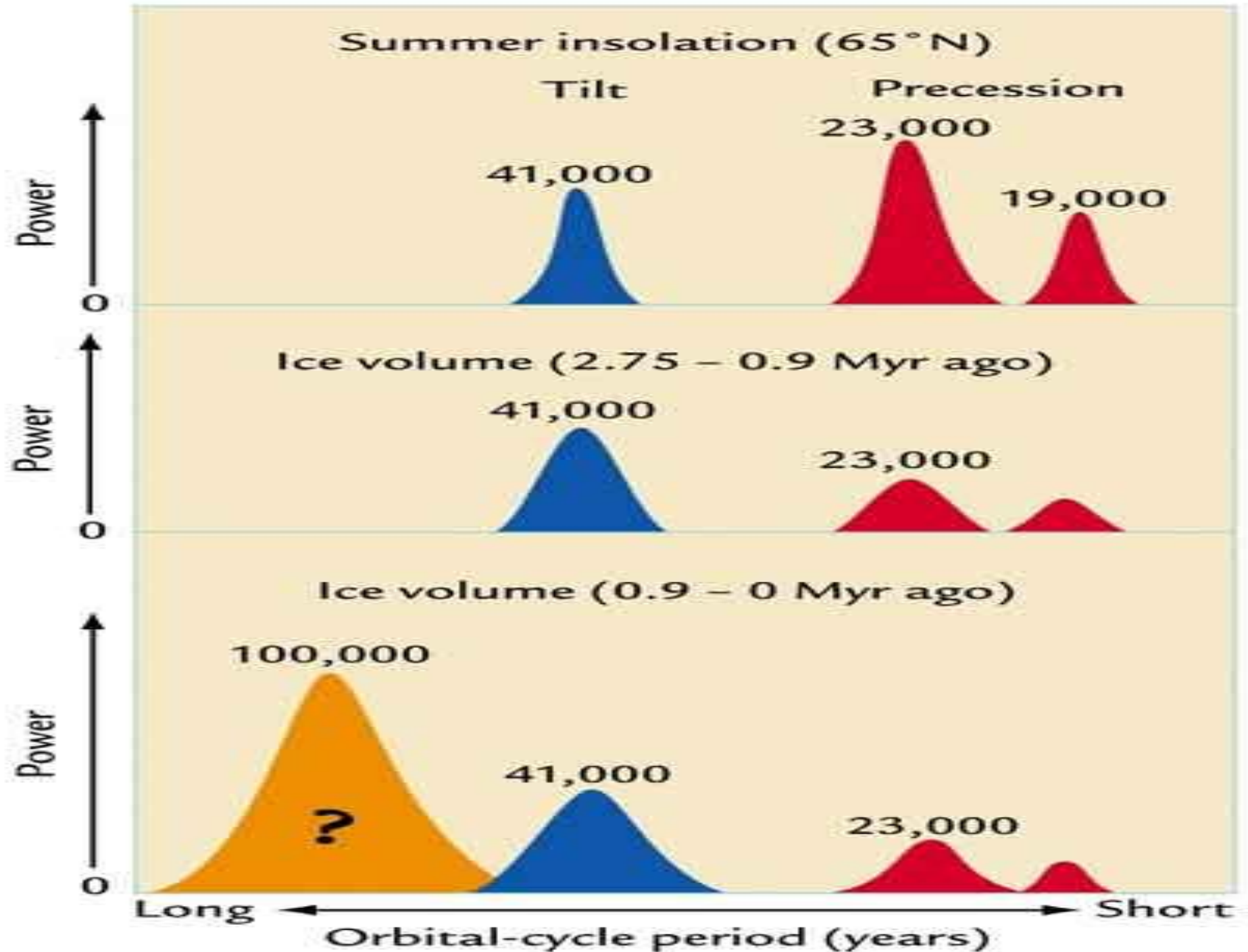


Precession The wobble of Earth's axis affects the amount of solar radiation that reaches different parts of Earth's surface at different times of the year.

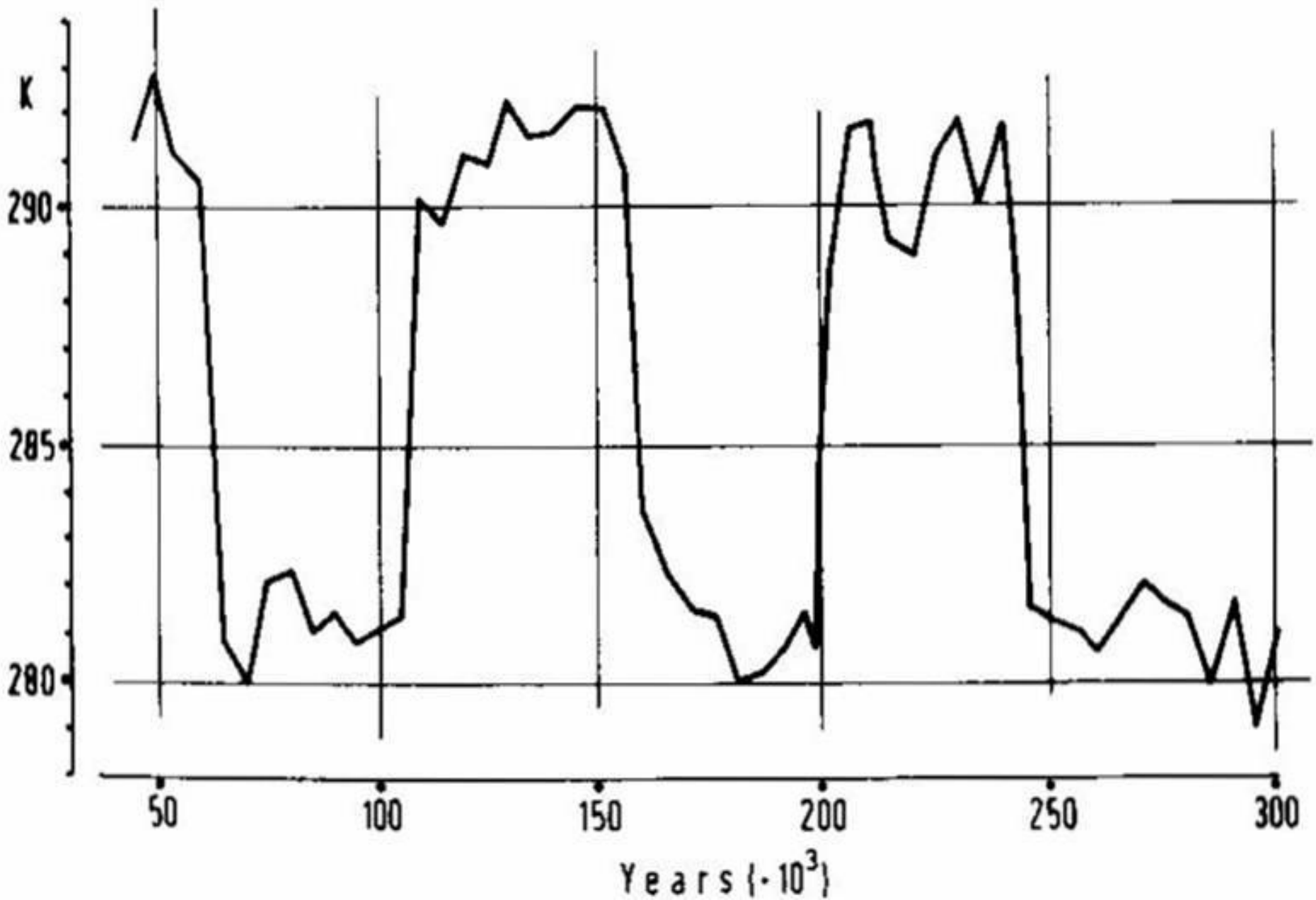








STOCHASTIC RESONANCE



A zero-dimensional energy balance model

$$\frac{dT}{dt} = F(T)$$

Solutions of $F(T)=0$ represent steady or equilibrium states .

To investigate the stability properties, introduce the the concept of pseudo-potential

$$\Phi = -\int F(T)dT$$

$T_1 < T < T_2: F < 0$

$\Rightarrow T \rightarrow T_1$

$T_2 < T < T_3: F > 0$

$\Rightarrow T \rightarrow T_3$

$$\frac{dT}{dt} = F(T) = -\frac{\partial \Phi}{\partial T}$$

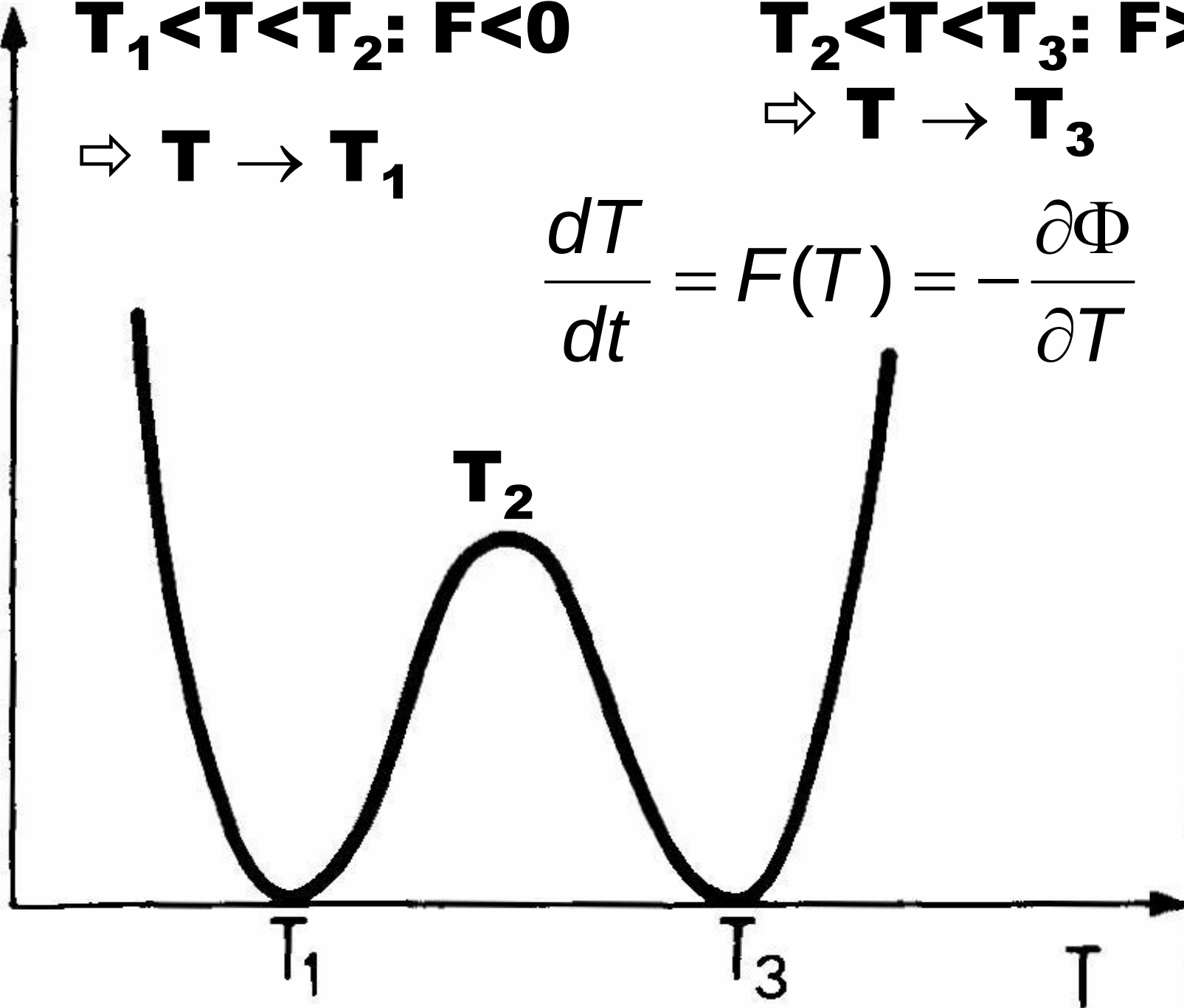
Φ

T_2

T_1

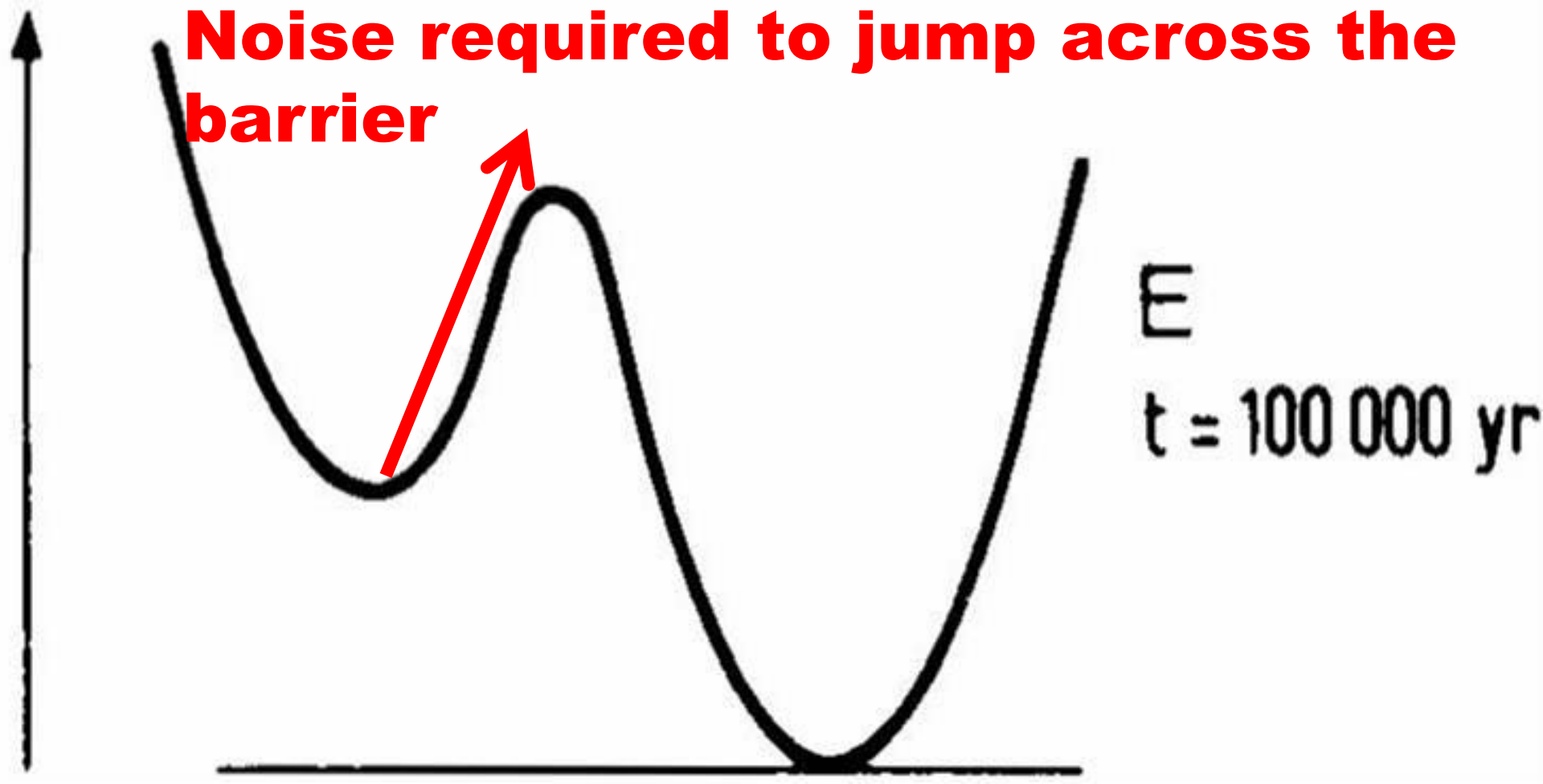
T_3

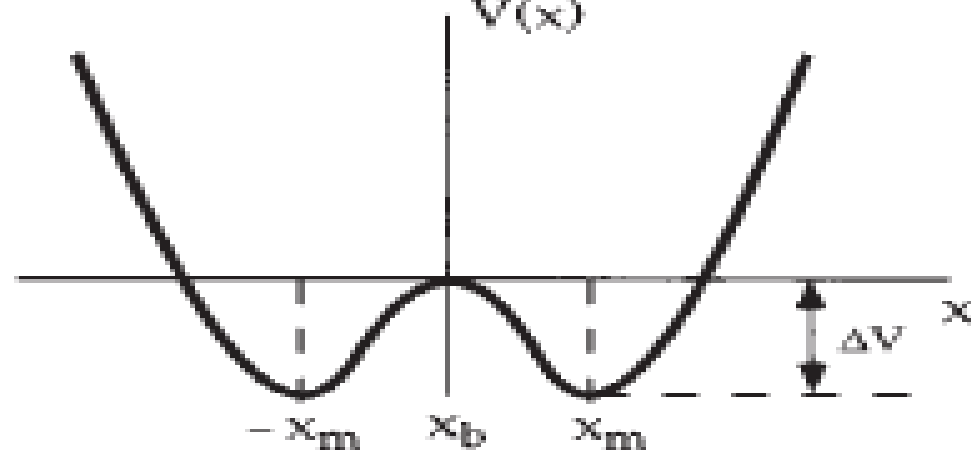
T



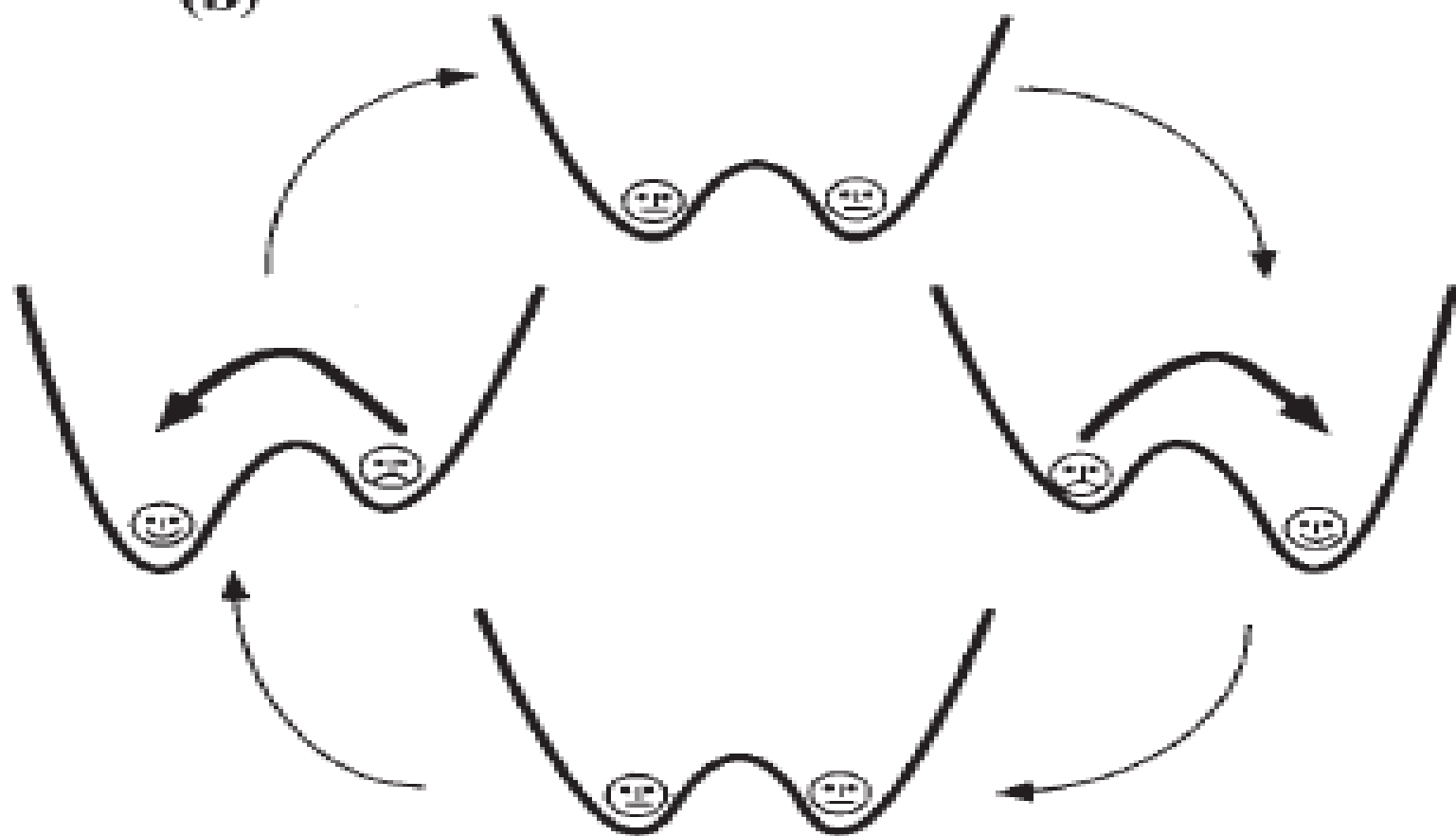
$$\tilde{F}(T, t) = F(T) \times (1 + 0.0005 \cos \varpi t)$$

$$\varpi = 2\pi / 10^5 \text{ years}$$

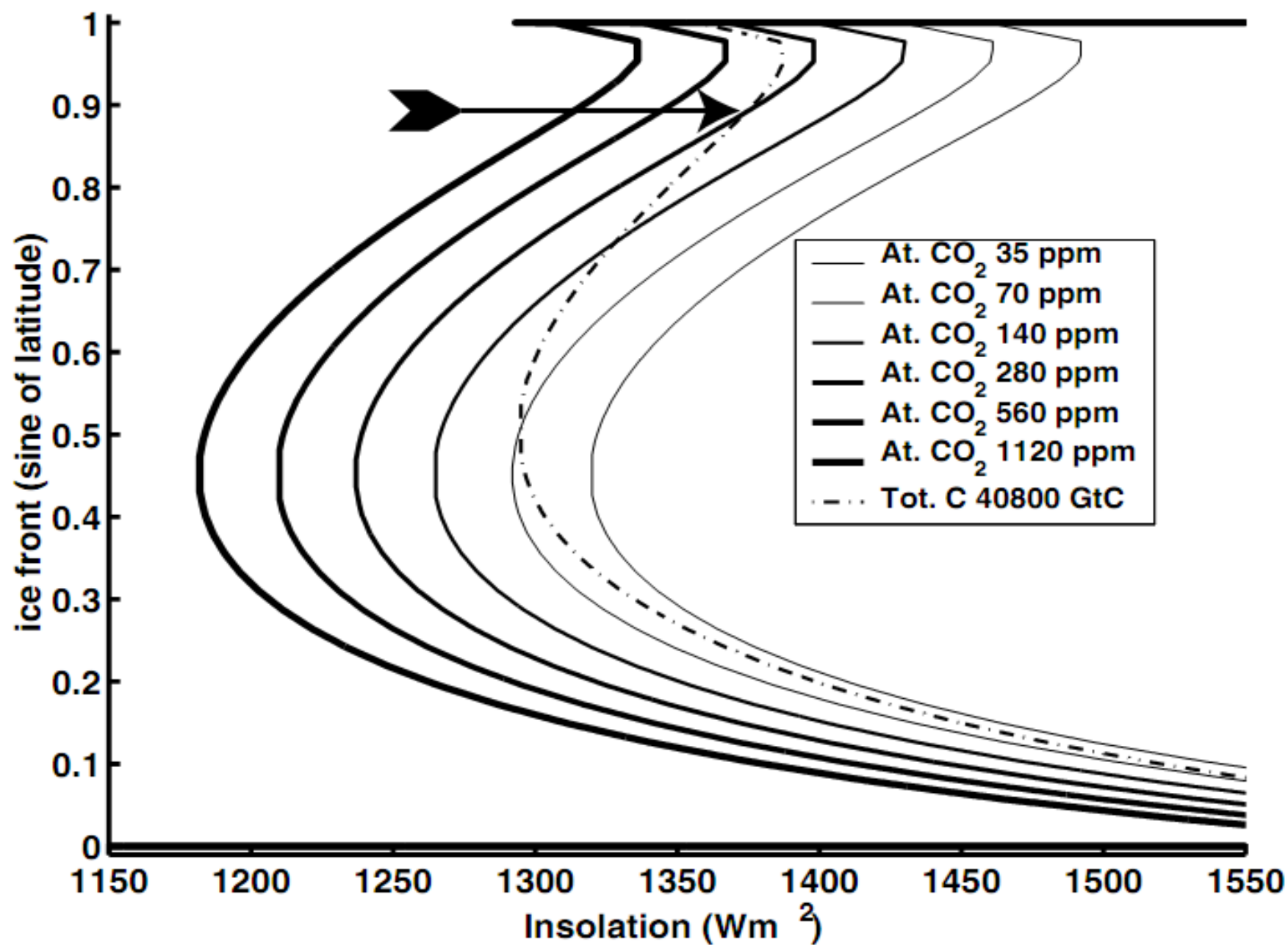




(b)



- **Orbital forcing of a simple energy balance model results in the right spectrum, but the amplitude is too small.**
- **Noise added to a simple model with prescribed stable equilibrium states results in the right amplitude, but the spectrum shows no peak.**
- **Combination of both is able to explain both amplitude and frequency of observed climate shifts.**



SALTZMAN'S MODEL

$$\frac{dX}{dt} = -\alpha_1 Y - \alpha_2 Z - \alpha_3 Y^2$$

$$\frac{dY}{dt} = -\beta_0 X + \beta_1 Y + \beta_2 Z - (X^2 + 0.004Y^2)Y + F_Y$$

$$\frac{dZ}{dt} = X - \gamma_2 Z$$

where in this particular case X , Y and Z are the ice mass, deep ocean temperature and atmospheric carbon dioxide.

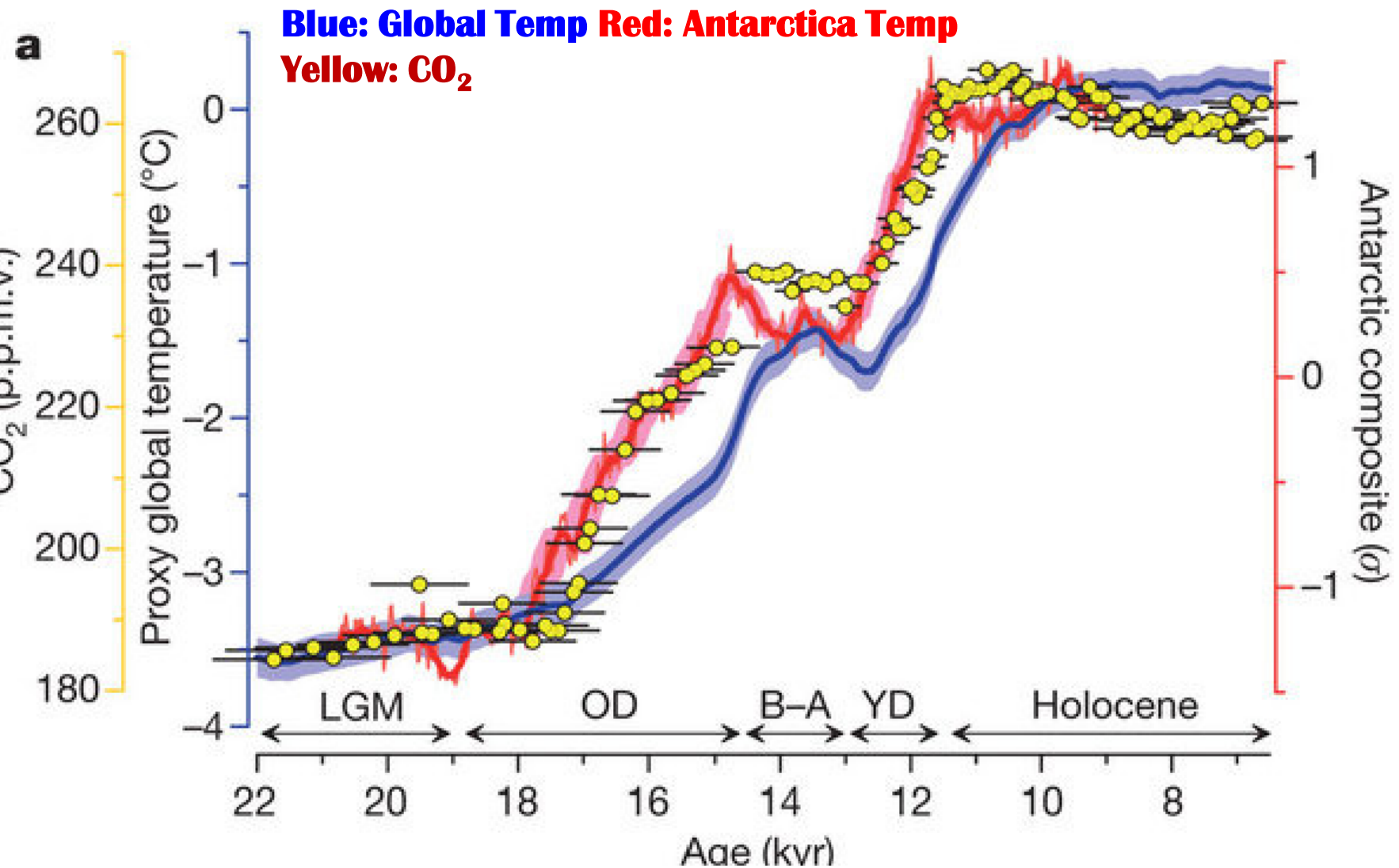
where X is ice mass,

Y is ocean temperature

Z is CO_2

Common Misconception

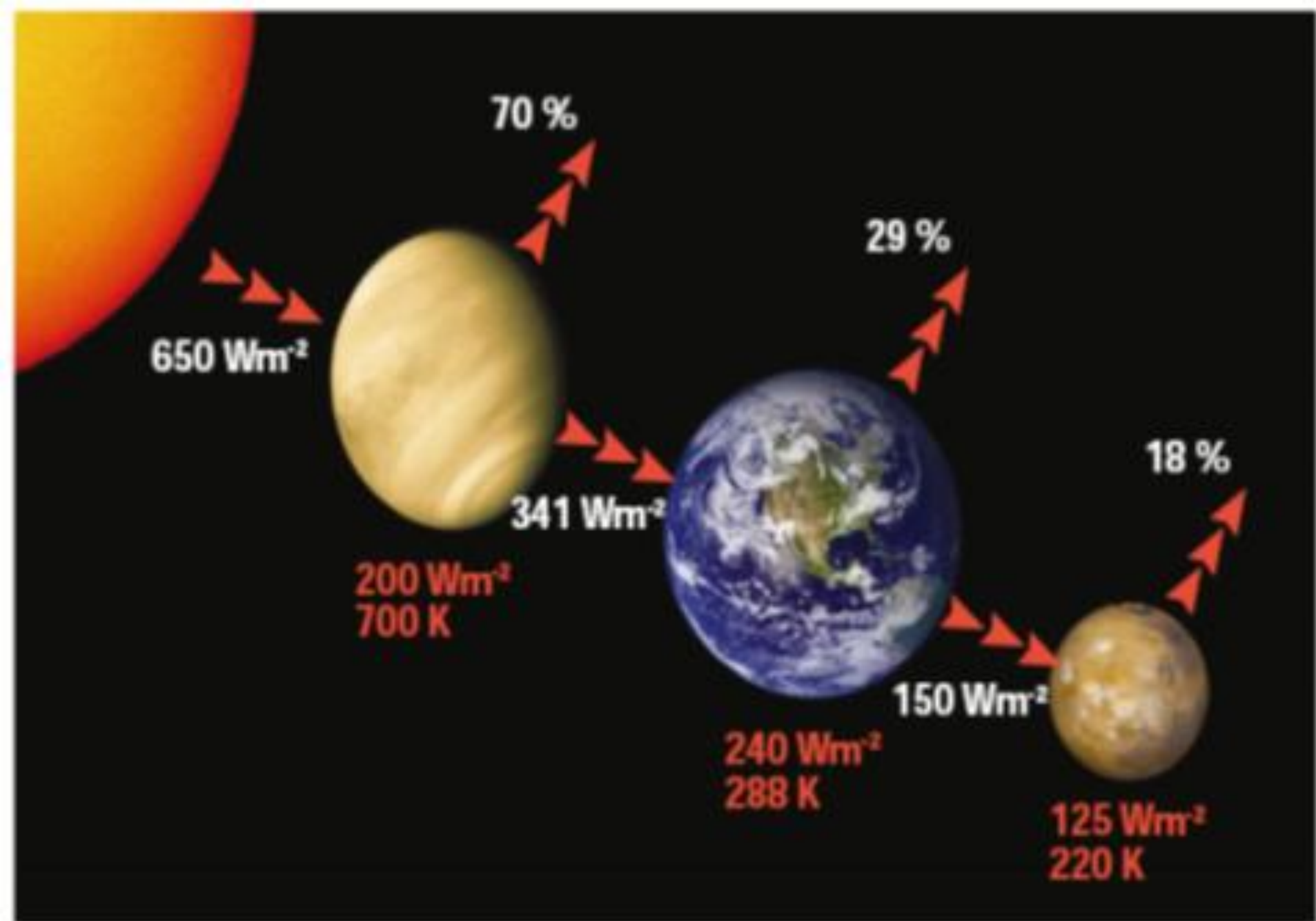
CO₂ is a very minor constituent of the atmosphere and hence cannot influence climate change

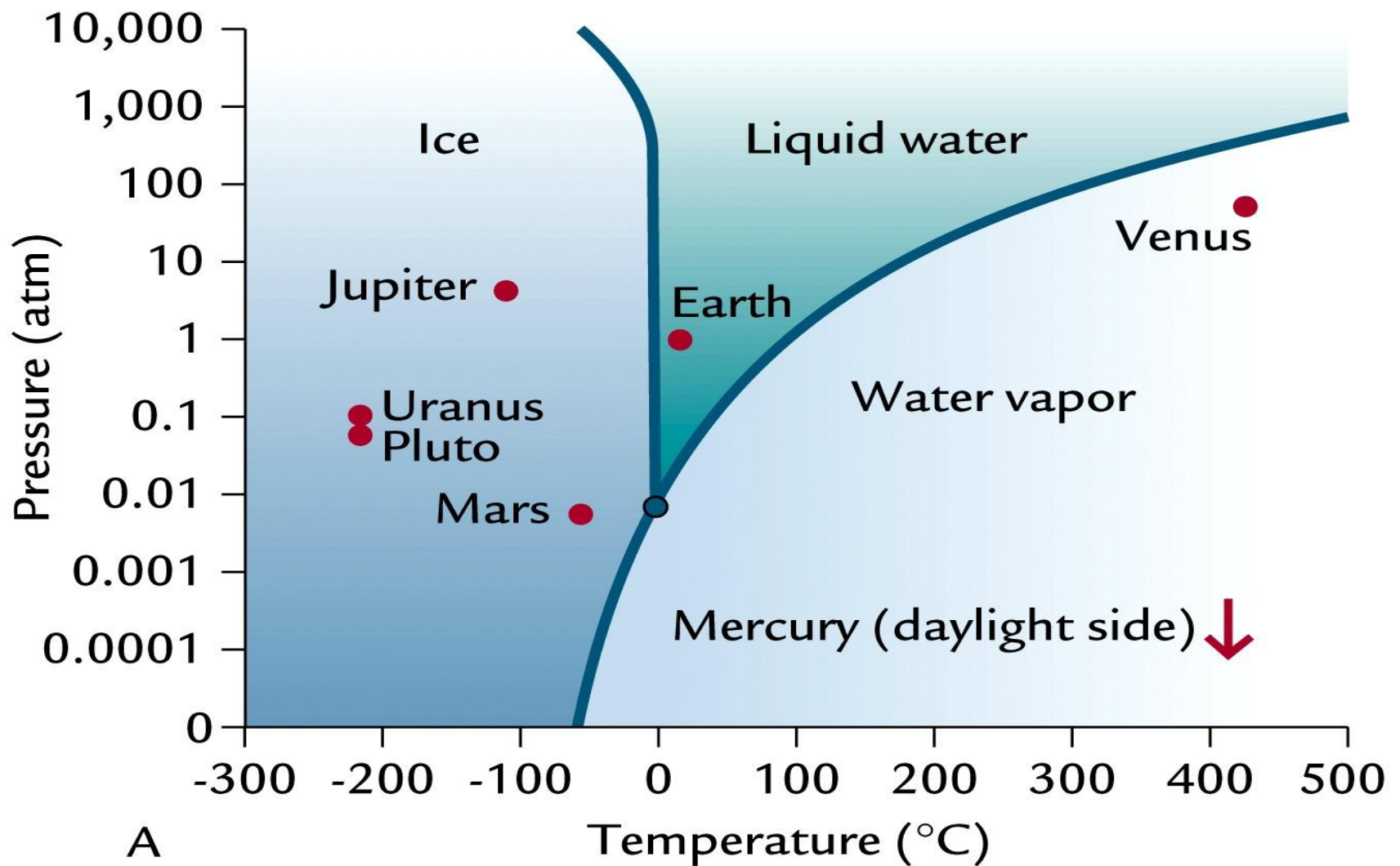


The global proxy temperature stack (blue) as deviations from the early Holocene (11.5–6.5 kyr ago) mean, an Antarctic ice-core composite temperature record⁴² (red), and atmospheric CO₂ concentration (refs 12, 13; yellow dots).

Earth vs Venus

- Although Earth and Venus started with similar composition
 - Earth evolved such that **carbon safely buried in early sediments**
 - Avoiding runaway greenhouse effect
- Venus built up CO_2 in the atmosphere
 - Build-up led to high temperature





Water exists in solid, liquid and gaseous form only on **earth**

**“If we burn all our reserves
of fossil fuels, there is a
substantial chance that we
will initiate the runaway
greenhouse”**

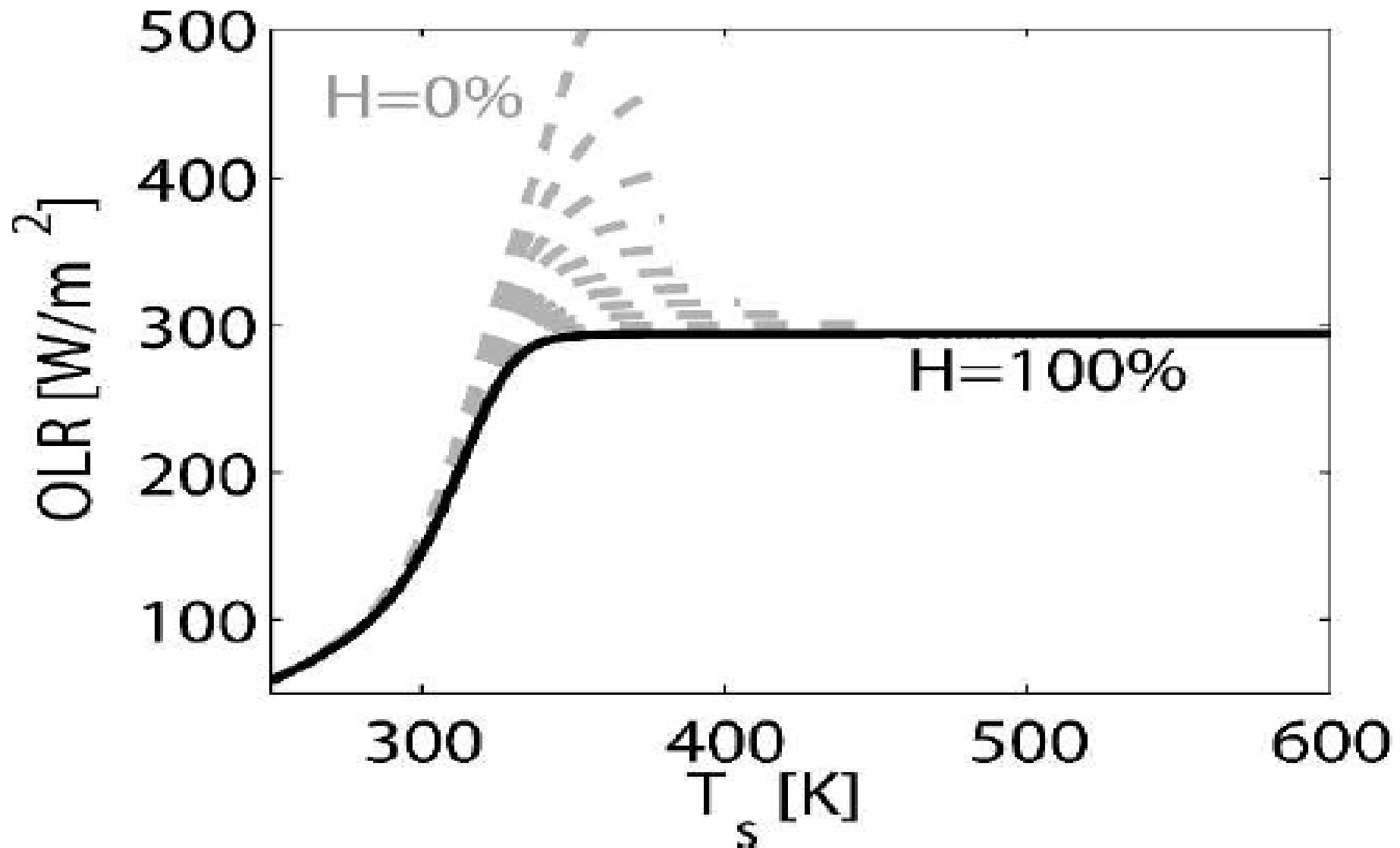
James Hansen

**“Storms of
my grand children”
2009 ,New York**



Dr. James Hansen

Note that $OLR \neq \sigma T^4$



Runaway Greenhouse

$$T_s^4 = S_o / 4 (1 - \alpha) (1 + 0.75 \delta_a)$$

$$\delta_a = \text{optical depth} = \int K_{\text{abs}} \rho_v dz$$

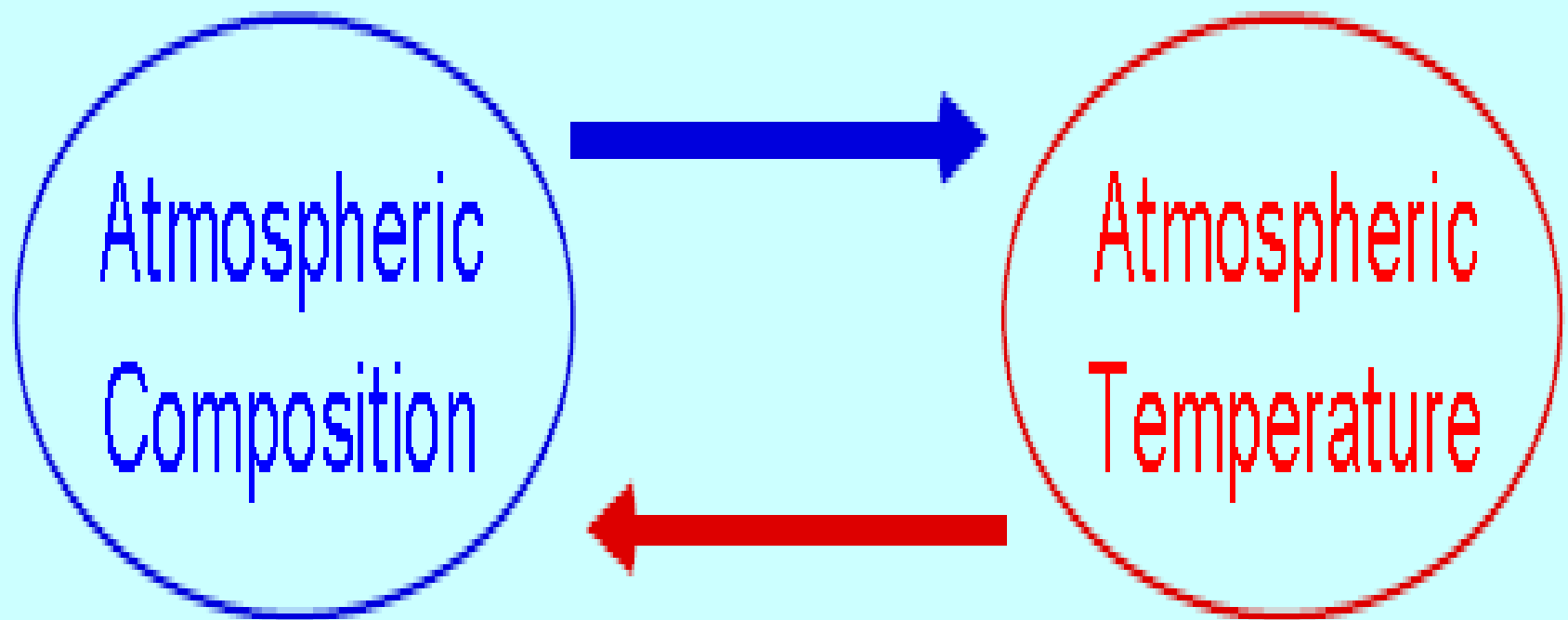
$$\delta_a = K_{\text{abs}} P_s (T_s) M_v / (M_a g)$$

If the **atmosphere is saturated with vapor**, then the **optical depth will increase monotonically with surface temperature**

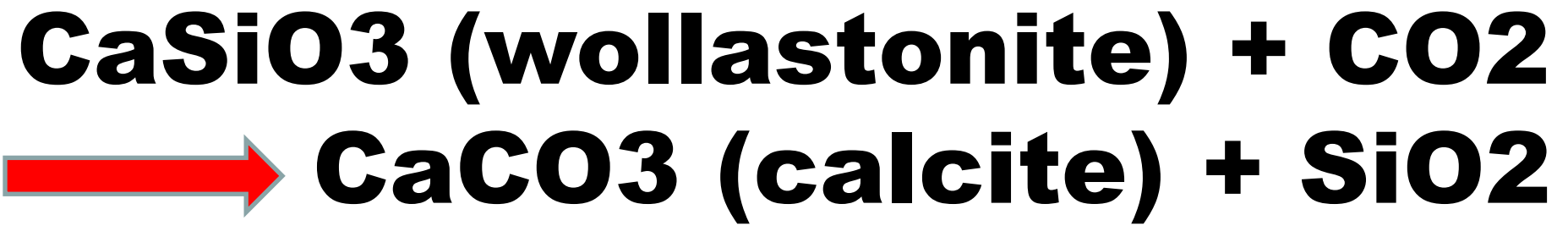
Composition of Venus atmosphere is controlled by chemical reaction on the planetary surface

- surface temperature
- chemical composition of planetary surface

albedo, greenhouse effect

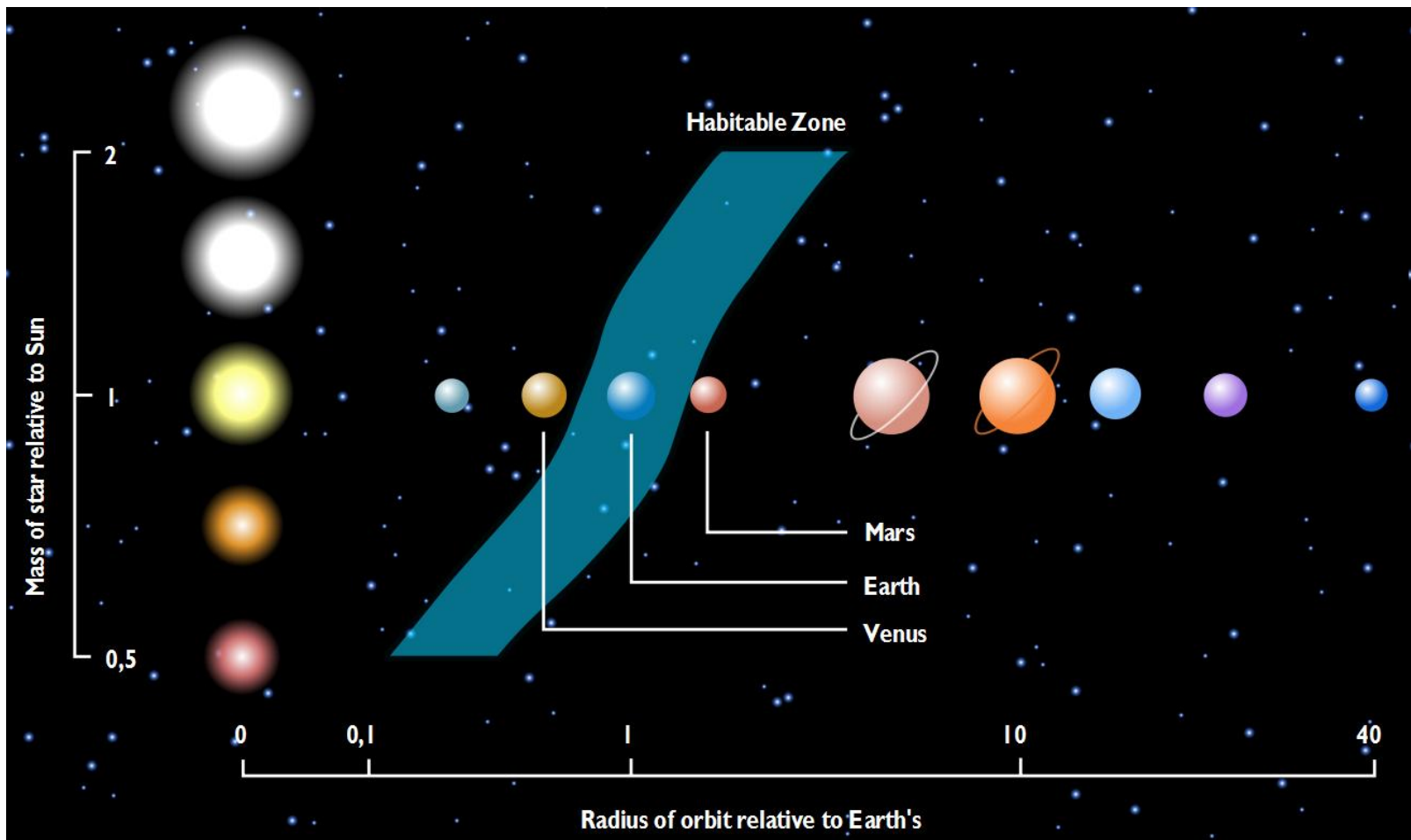


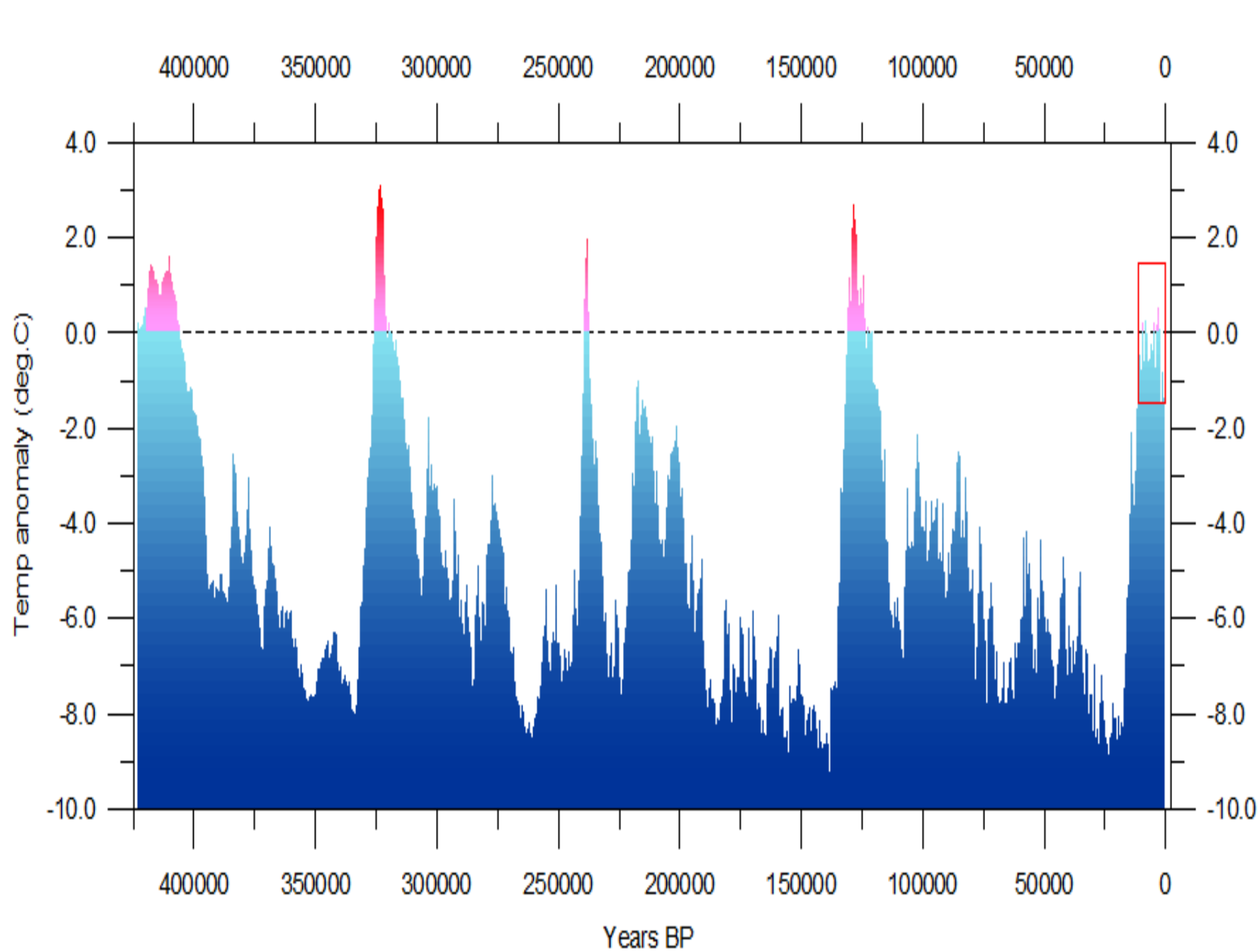
chemical reaction



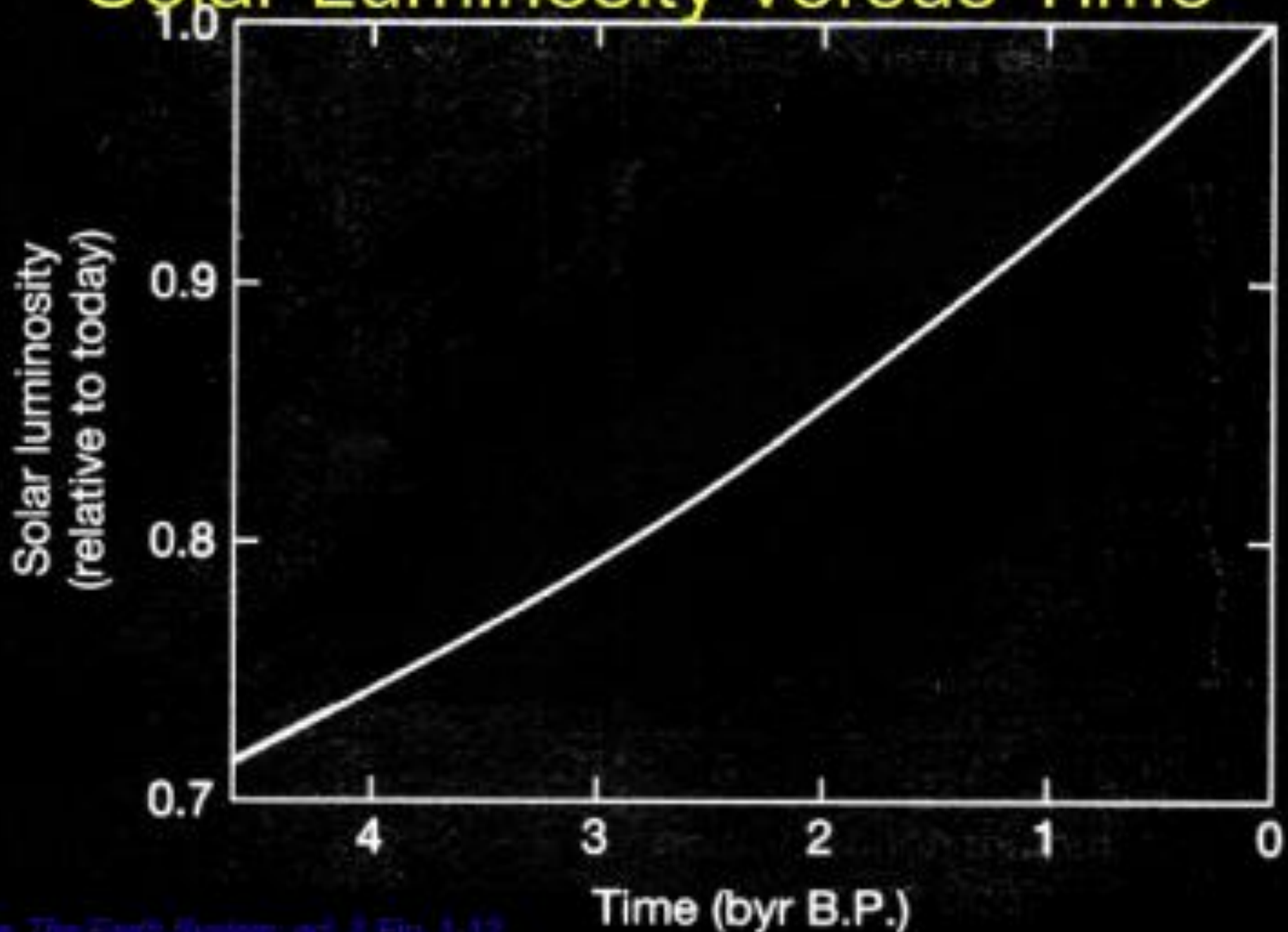
Carbonate-silicate reaction
shifted in the favour of
carbonate formation below
300 °C but shifts to silicate
formation above 300 °C

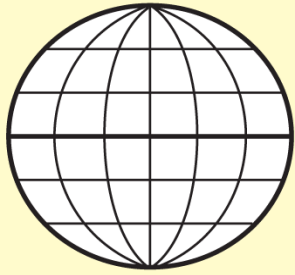
The (liquid water) habitable zone





Solar Luminosity versus Time





Solar flux (x present)

0.9 1.0 1.1 1.2 1.3

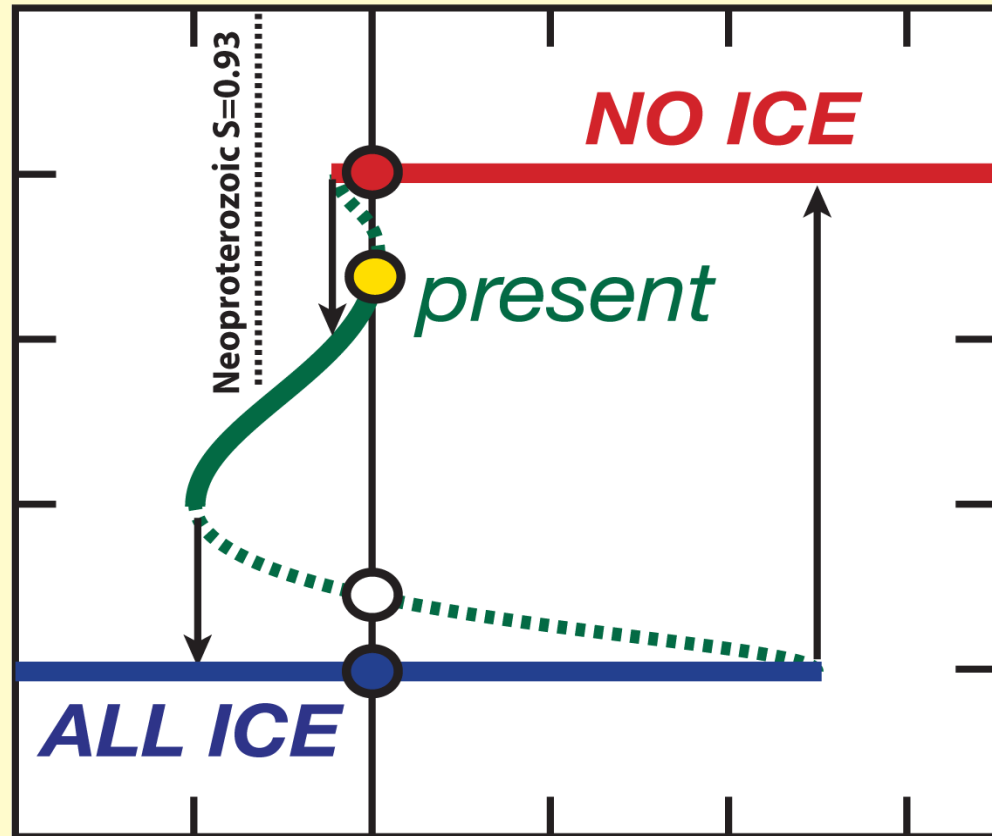
Ice line latitude

90°

60°

30°

Eq

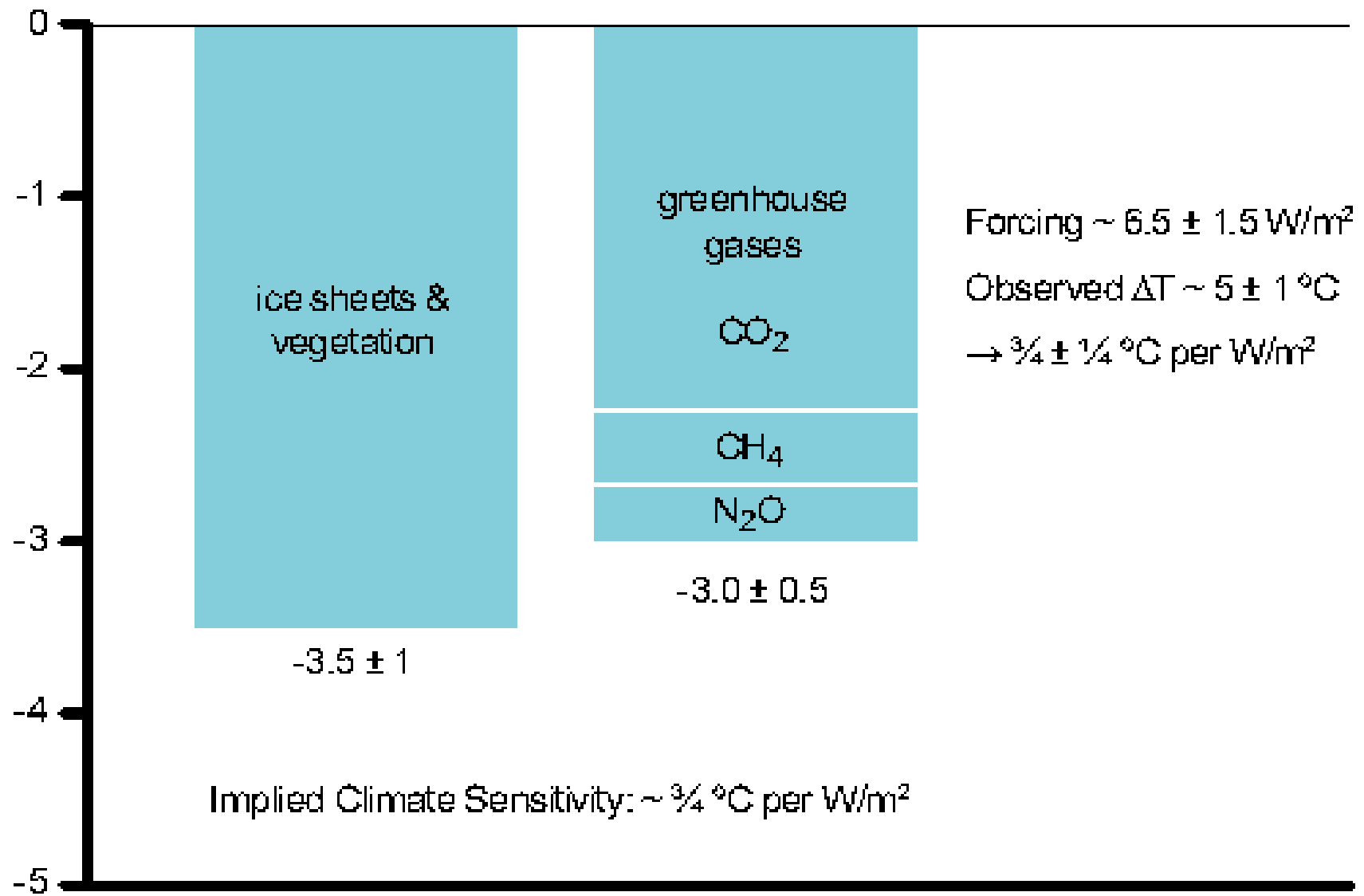


Budyko-Sellers type
energy-balance model
(1969)

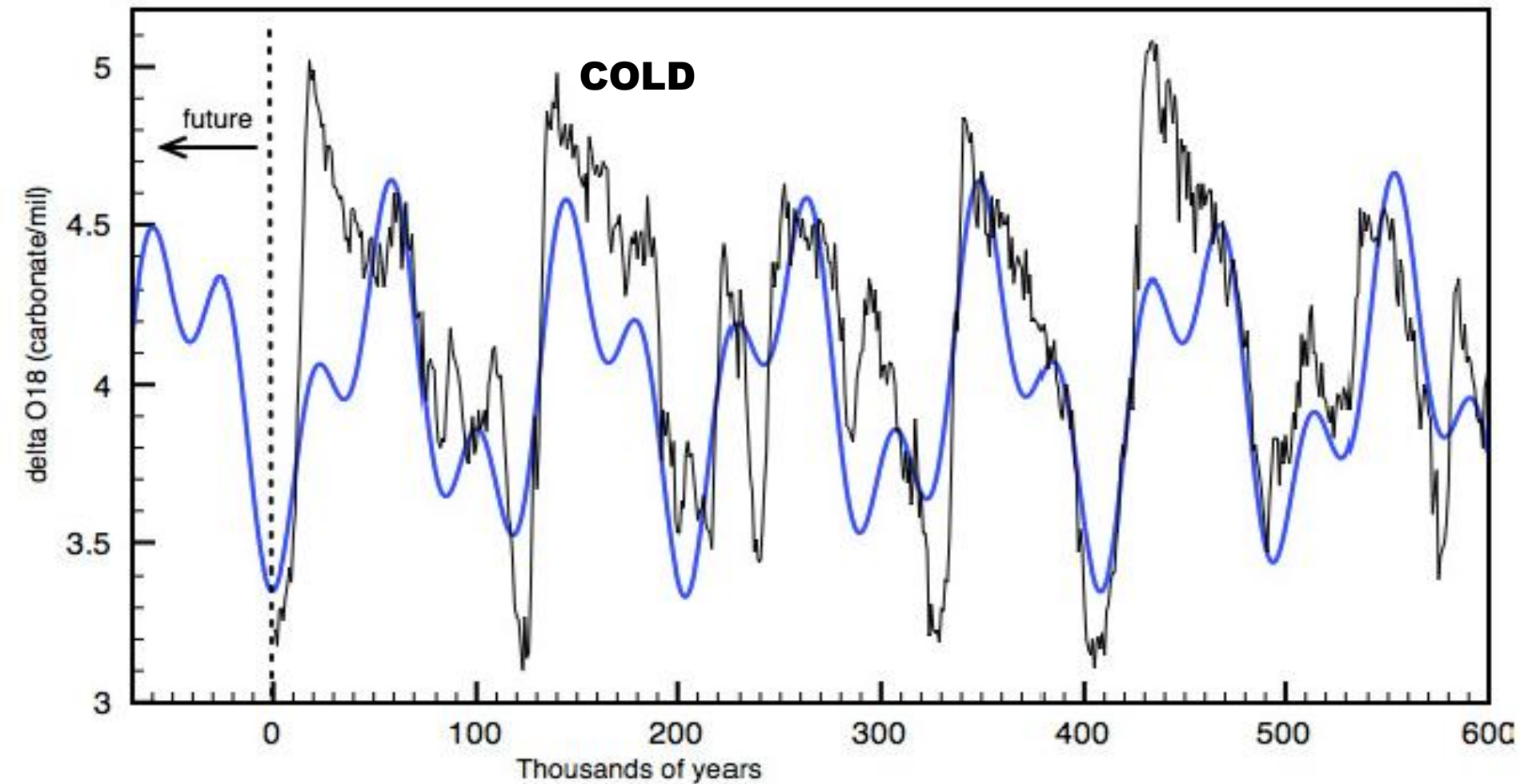
0.1 1 10 100 1000

log $p\text{CO}_2$ (x present)

Ice Age Climate Forcings (W/m^2)



Climate forcings during ice age 20 ky BP, relative to the present (pre-industrial) interglacial period.
(Source: Hansen et al. 2008)



δO_{18} record from microfossils in the ocean. Blue line solar radiation at 65° N