Molecular Simulations: Applications in Biology Structure of Biomolecules: An Overview

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Understanding Molecular Simulations: Theory and Applications UMS2010 11th November 2010

Outline of this talk

What are Biomolecules?
 Significance of knowing the structure of a biomolecule?

Why simulate a biomolecule?

What is the current status?

Biomolecular simulation: An example



Building Blocks

Proteins Nucleic acids Carbohydrates Lipids

Amino acids Nucleotides Sugars Fatty acids

Proteins play crucial roles in all biological processes

Trypsin, Chmytrypsin – enzymes

Hemoglobin, Myoglobin – transports oxygen

Transferrin – transports iron

Ferritin – stores iron

Myosin, Actin – muscle contraction

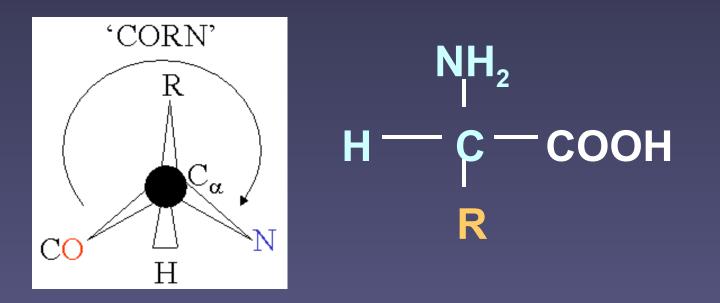
Collagen – strength of skin and bone

Rhodopsin – light-sensitive protein Acetylcholine receptor – responsible for transmitting nerve impluses

Antibodies – recognize foreign substances

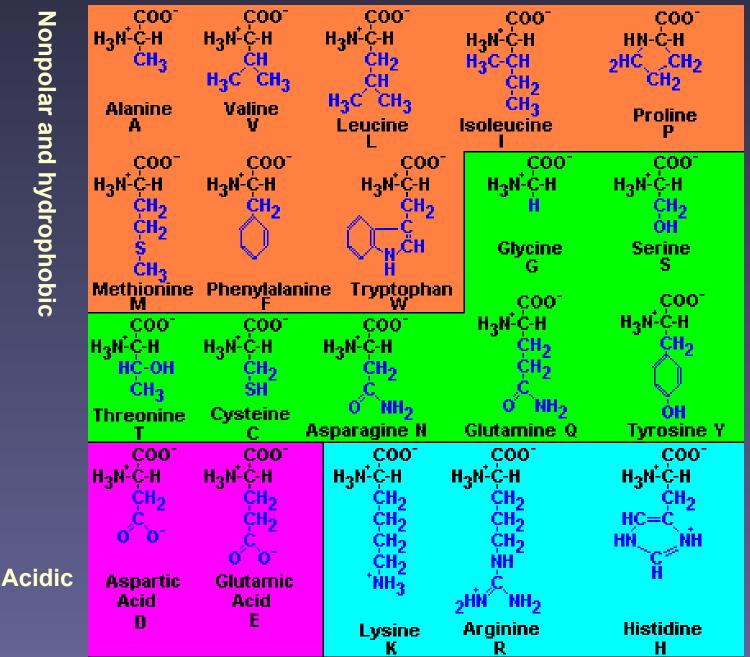
Repressor and growth factor proetins

Proteins are made up of 20 amino acids



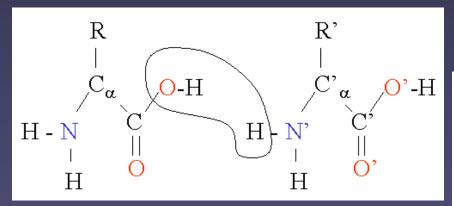
R varies in size, shape, charge, hydrogen-bonding capacity and chemical reactivity.

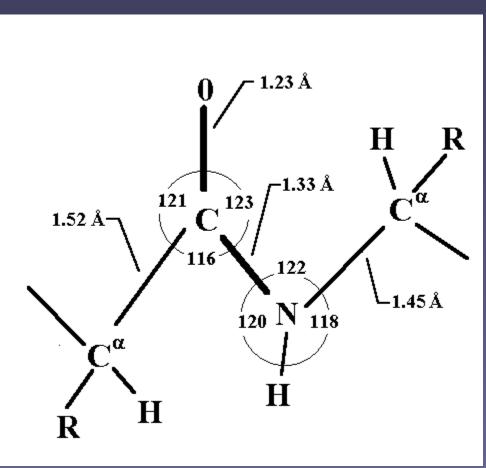
Only L-amino acids are constituents of proteins



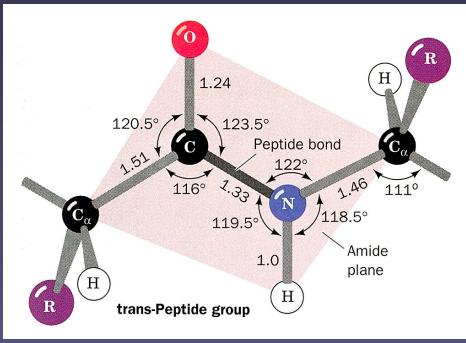
Basic

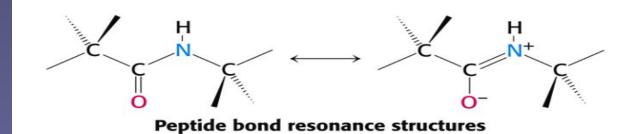
20 amino acids are linked into proteins by *peptide bond*



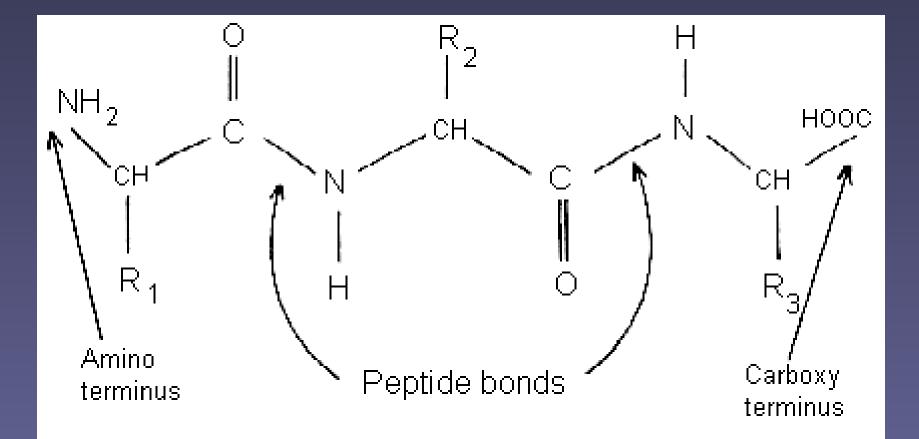


Peptide bond has partial double-bonded character and its rotation is restricted.





Polypeptide backbone is a repetition of basic unit common to all amino acids

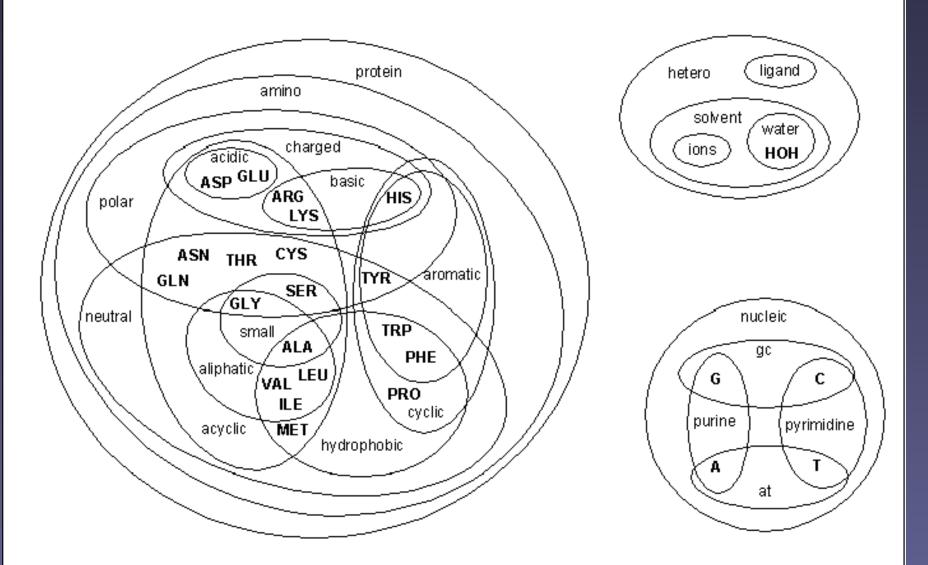


Frequently encountered terms in protein structure

Backbone

Side chain

•Residue



RasMol/Chime Venn diagram v1.2 Drawn by Kurt Giles (kurt@inn-prot.weizmann.ac.il) for the Israeli National Node of EMBnet

Α	Ala	alanine
С	Cys	cysteine
D	Asp	aspartic acid
Е	Glu	glutamic acid
F	Phe	phenylalanine
G	Gly	glycine
н	His	histidine
I.	lle	isoleucine
Κ	Lys	lysine
L	Leu	leucine
Μ	Met	methionine
Ν	Asn	asparagine
Ρ	Pro	proline
Q	Gln	glutamine
R	Arg	arginine
S	Ser	serine
Т	Thr	threonine
V	Val	valine
W	Trp	tryptophan
Υ	Tyr	tyrosine

One letter and three-letter codes for amino acids Proteins can exist in two types of environments

Globular proteins

Membrane proteins - Dr. Satyavani

Each protein has a characteristic three-dimensional structure which is important for its function

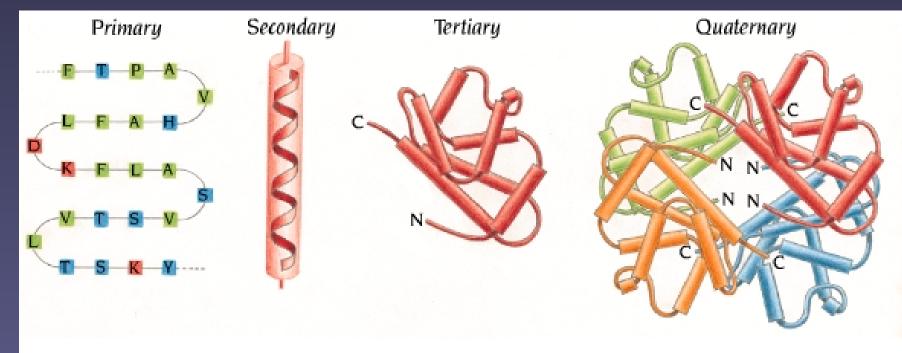
Protein Structure: Four Basic Levels

Primary Structure

Secondary Structure

Tertiary Structure

Quaternary Structure



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Protein - Primary Structure

Linear amino acid sequence
Determines all its chemical and biological properties
Specifies higher levels of protein structure (secondary, tertiary and quaternary)

Most proteins contain between ~200 to ~500 residues

Histone (human)

SETVPPAPAASAAPEKPLAGKKAKKPAKAAAASKKKPAGPSVSELIVQAASSSKER GGVSLAALKKALAAAGYDVEKNNSRIKLGIKSLVSKGTLVQTKGTGASGSFKLNK KASSVETKPGASKVATKTKATGASKKLKKATGASKKSVKTPKKAKKPAATRKSSK NPKKPKTVKPKKVAKSPAKAKAVKPKAAKARVTKPKTAKPKKAAPKKK

Rhodopsin (human)

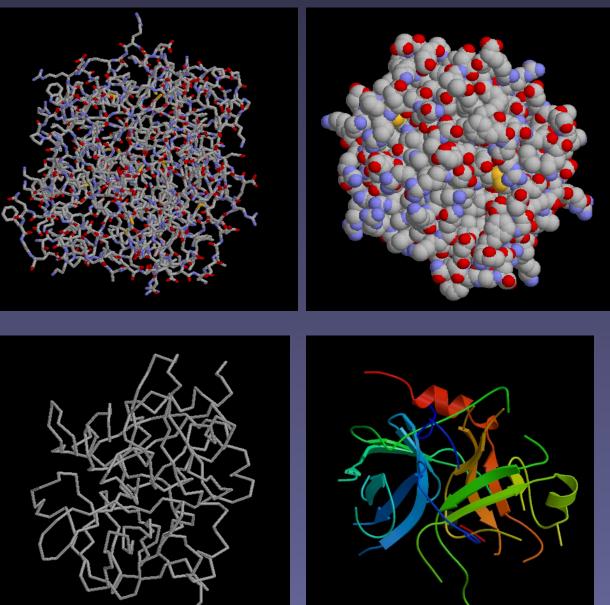
MNGTEGPNFYVPFSNATGVVRSPFEYPQYYLAEPWQFSMLAAYMFLLIVLGFPI NFLTLYVTVQHKKLRTPLNYILLNLAVADLFMVLGGFTSTLYTSLHGYFVFGPTGC NLEGFFATLGGEIALWSLVVLAIERYVVVCKPMSNFRFGENHAIMGVAFTWVM ALACAAPPLAGWSRYIPEGLQCSCGIDYYTLKPEVNNESFVIYMFVVHFTIPMIII FFCYGQLVFTVKEAAAQQQESATTQKAEKEVTRMIIMVIAFLICWVPYASVAF YIFTHQGSNFGPIFMTIPAFFAKSAAIYNPVIYIMMNKQFRNCMLTTICCGKNP LGDDEASATVSKTETSQVAPA

Thrombin

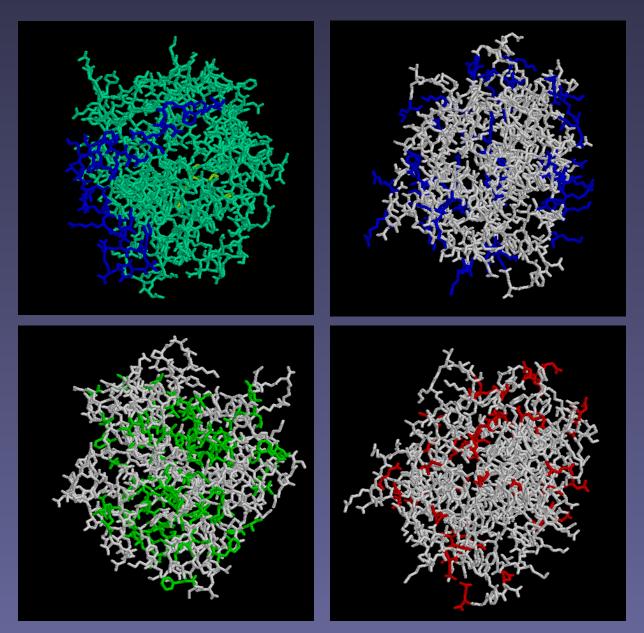
Heavy chain: IVEGSDAEIGMSPWQVMLFRKSPQELLCGASLISDRWVLTAAHCLLYPPW DKNFTENDLLVRIGKHSRTRYERNIEKISMLEKIYIHPRYNWRENLDRDIAL MKLKKPVAFSDYIHVCLPDRETAASLLQAGYKGRVTGWGNLKETWTANVG KGQPSVLQVVNLPIVERPVCKDSTRIRITDNMFCAGYKPDEGKRGDACEGDS GGPFVMKSPFNNRWYQMGIVSWGEGCDRDGKYGFY THVFRLKKWIQKVIDQFGE

Light Chain: TFGSGEADCGLRPLFEKKSLEDKTERELLESYIDGR

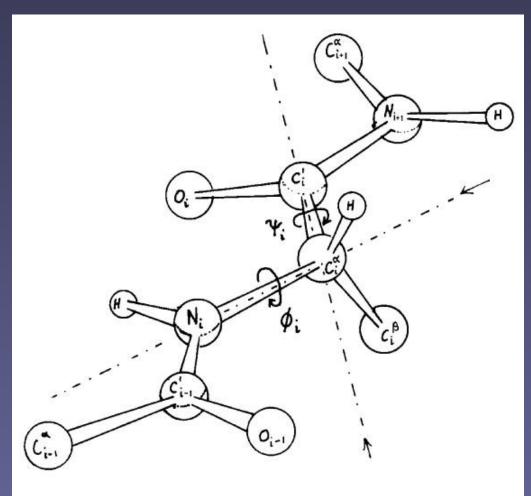
Thrombin Structure

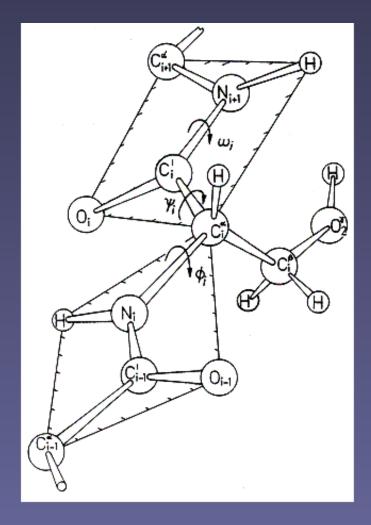


Thrombin Structure

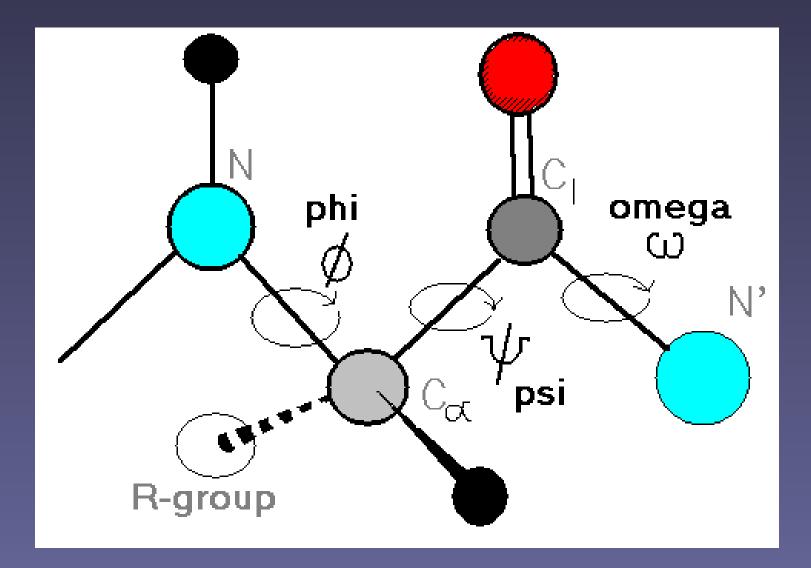


Primary to Secondary structure Importance of Dihedral Angle



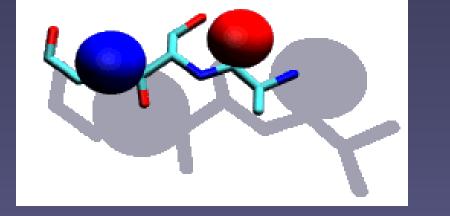


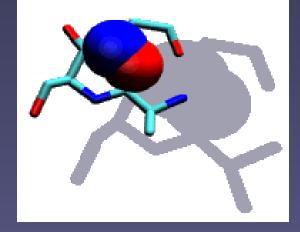
Dihedral angles ϕ , ψ and ω

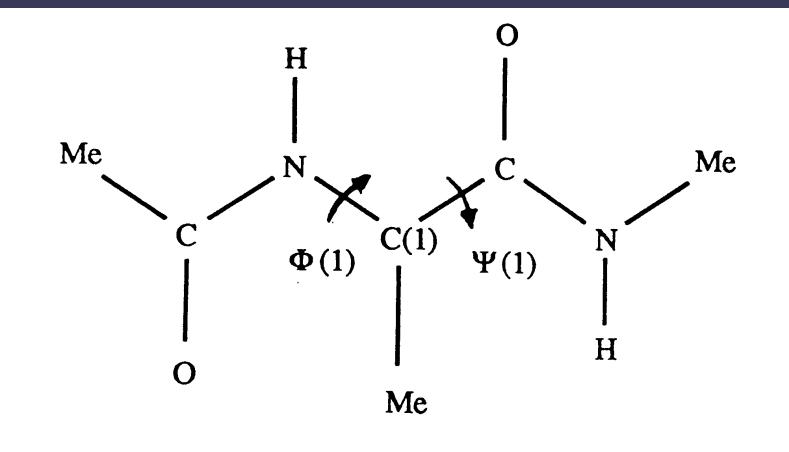


$\phi = 180^{\circ}; \psi = 180^{\circ}$

$\phi = 0^{\circ}; \psi = 0^{\circ}$



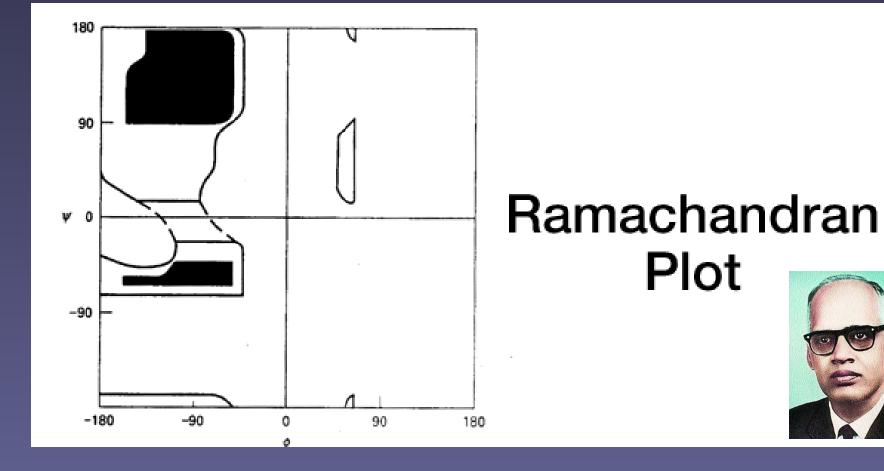




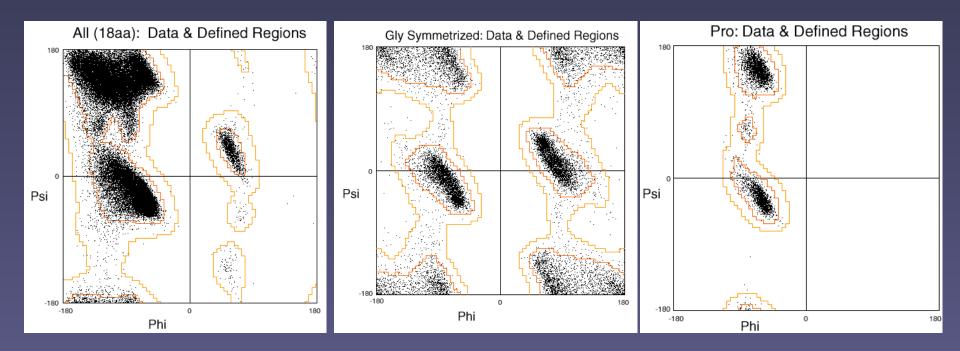
Limiting distances for various interatomic contacts

<u>Types of contact</u>	<u>Normal Limit</u>	<u>Extreme Limit</u>
НН	2.0	1.9
НО	2.4	2.2
HN	2.4	2.2
НС	2.4	2.2
OO	2.7	2.6
ON	2.7	2.6
OC	2.8	2.7
NN	2.7	2.6
N <i>C</i>	2.9	2.8
CC	3.0	2.9
СС(Н)	3.2	3.0
С(Н)С(Н)	3.2	3.0

Ramachandran & Sasisekharan (1968) Adv. Protein Chem.

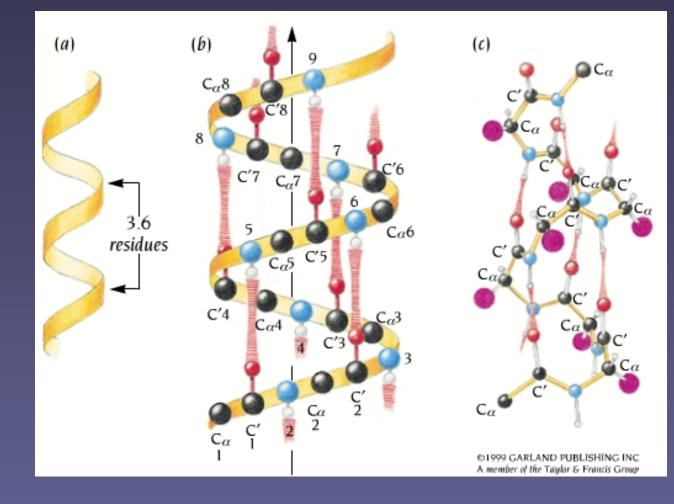


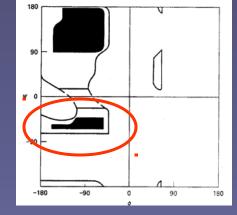
Ramachandran Plot Data from 500 high-resolution proteins



Secondary Structure

α -helix

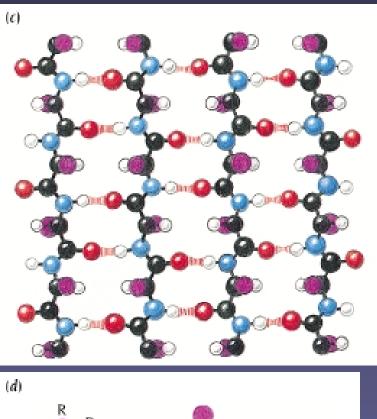


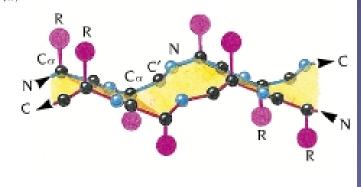


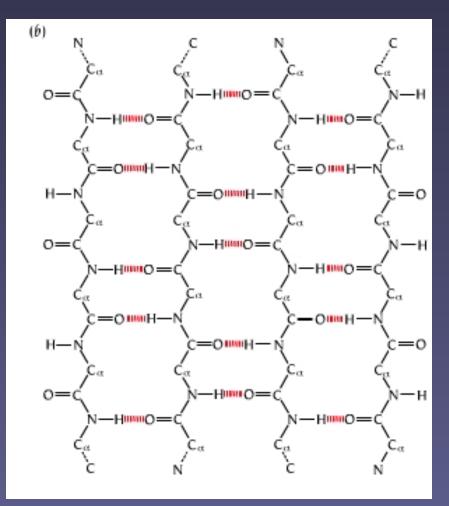
<u>α-helix</u>

3.6 residues per turn Translation per residue 1.5 Å Translation 5.4 Å per turn $C=O(i) \dots H-N(i+4)$ $\phi = -57^{\circ}; \psi = -47^{\circ}$ (classical value) $\phi = -62^{\circ}; \psi = -41^{\circ}$ (crystal structures) Preference of residues in helix Can proline occur in a helix? Average helix length ~ 10 residues

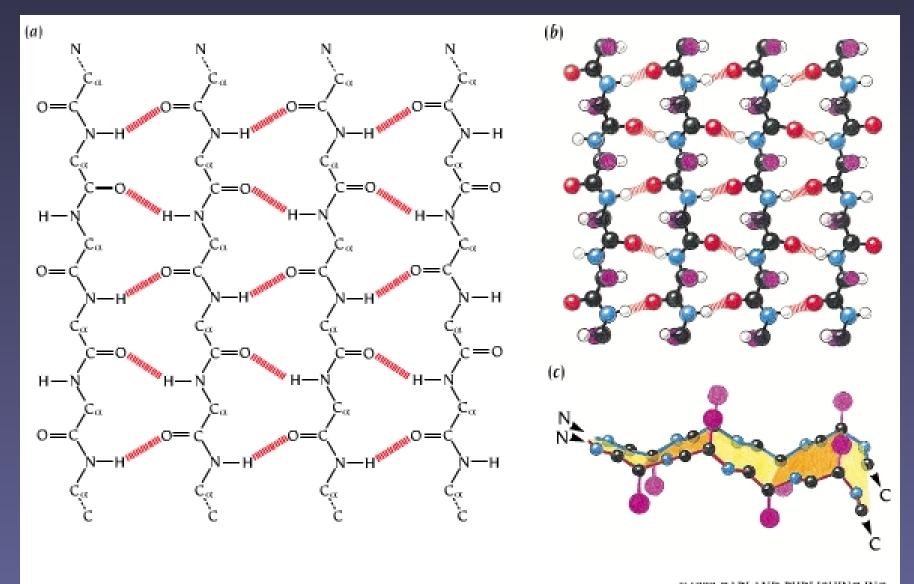
Antiparallel β -sheet





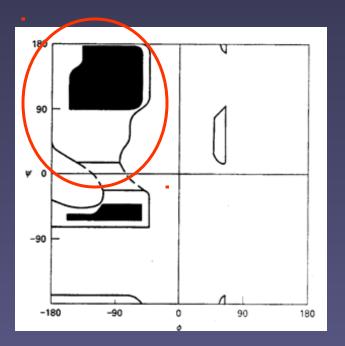


Parallel β -sheet



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<u>β-strand</u>

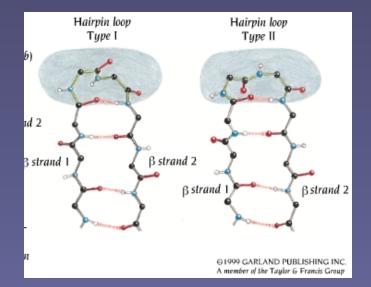


Polypeptide fully extended
2.0 residues per turn
Translation 3.4Å per residue
Stable when incorporated into a β-sheet
H-bonds between peptide groups of
adjacent strands
Adjacent strands can be parallel or
antiparallel

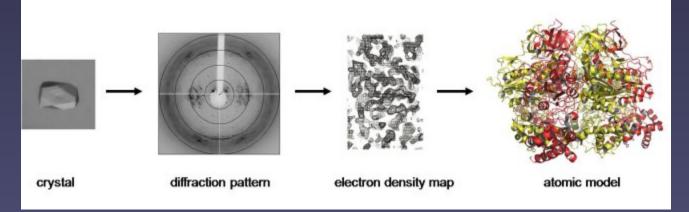
<u>Turns</u>

Secondary structures are connected by loop regions Lengths vary; shapes irregular Loop regions are at the surface of the molecule Rich in charged and polar hydrophilic residues Role: connecting units; binding sites; enzyme active sites Loops are often flexible; adopt different conformations

β-turns: Type I, Type II etc.
γ-turns; classical, inverse

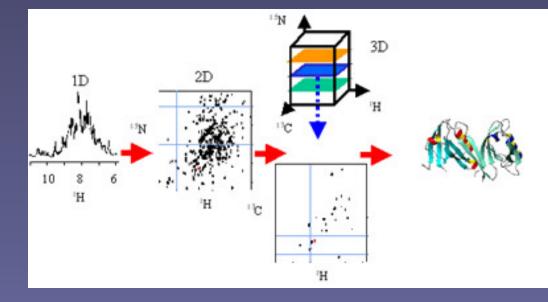


Structure Determination: Experimental Methods X-ray crystallography



http://www.uni-duesseldorf.de/home/Fakultaeten/math_nat/Graduiertenkollegs/biostruct/Research/BioStruct_Groups/AG_Groth/expertise.html

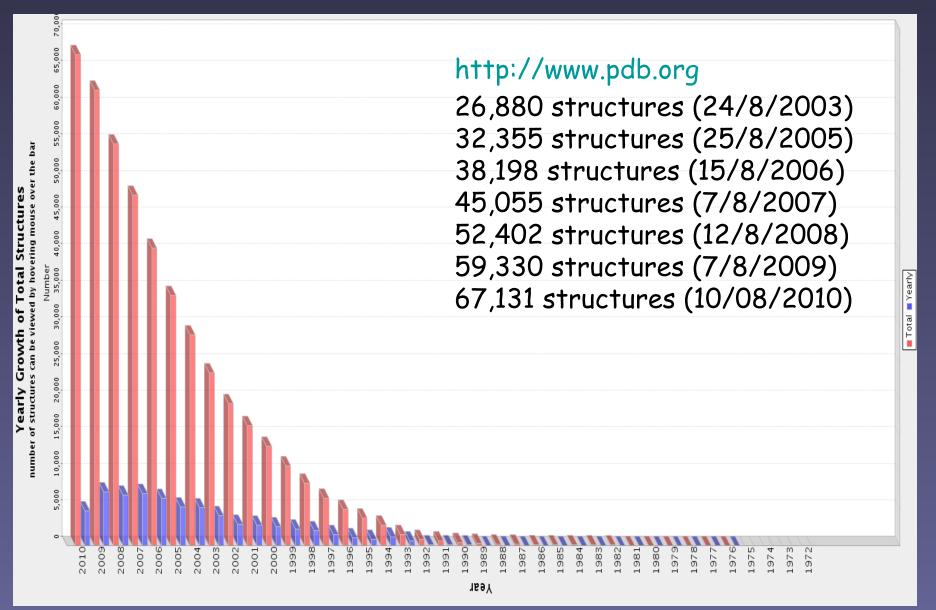
NMR



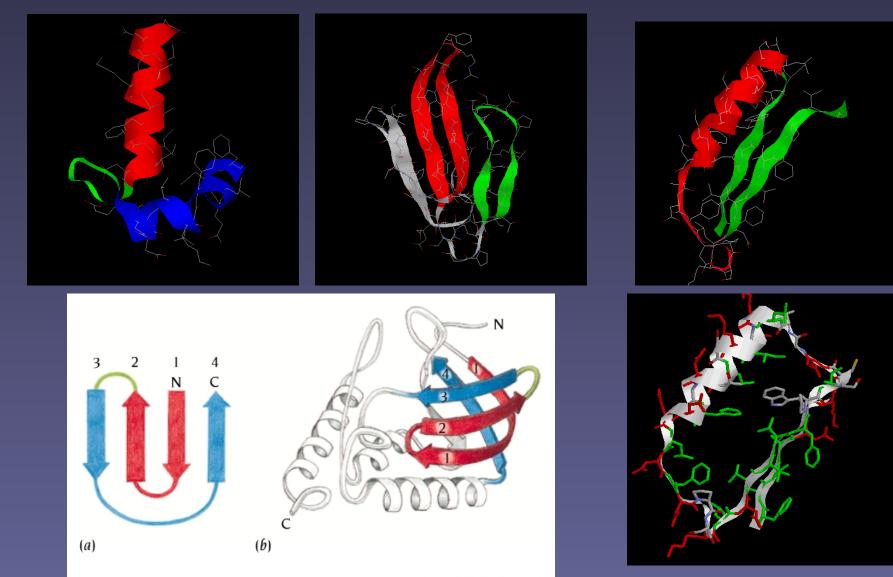
http://www.dbs.nus.edu.sg/staff/henry.htm



Growth of Protein Data Bank







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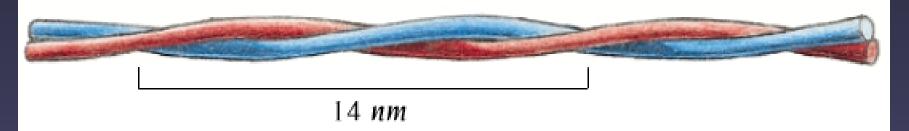
Main Classes of Protein Structures

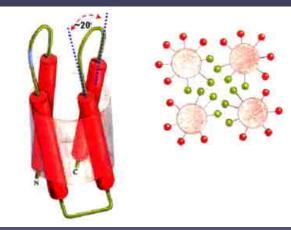
- α domains α -helices
- $\begin{array}{l} \beta \text{ domains} & \text{Antiparallel }\beta\text{-sheets} \\ \\ \hline \beta \text{ domains} & \text{Combinations of }\beta\text{-}\alpha\text{-}\beta \text{ motifs} \\ \\ \hline \alpha/\beta \text{ domains} & \text{Discrete }\alpha \text{ and }\beta \text{ motifs} \end{array}$

 α + β domains

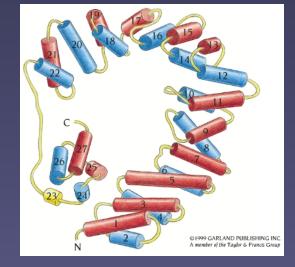
Disulfide bonds/metal atoms

Coiled-coil



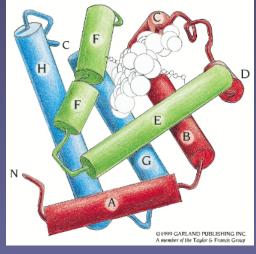


Alpha-domain



Large alpha-helical domain

Four-helix bundle

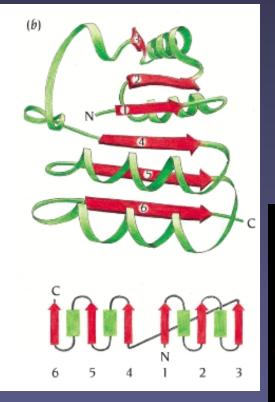


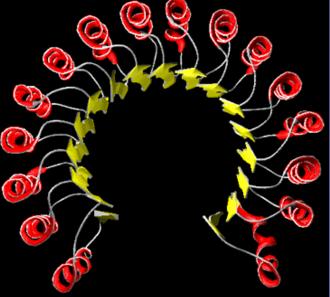
Globin fold



TIM-barrel

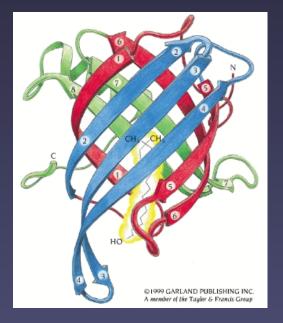
Rossman fold

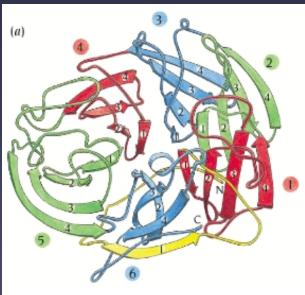




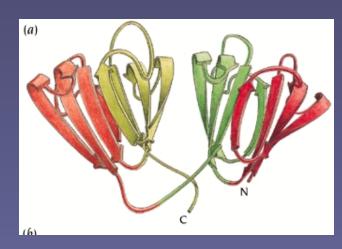
a/B structures

Horseshoe fold

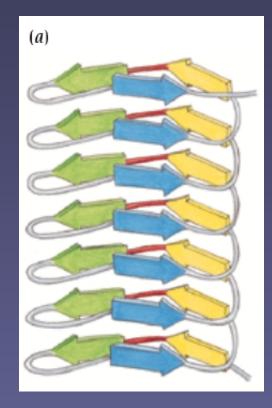




Up-and-down beta-barrel



B-domain



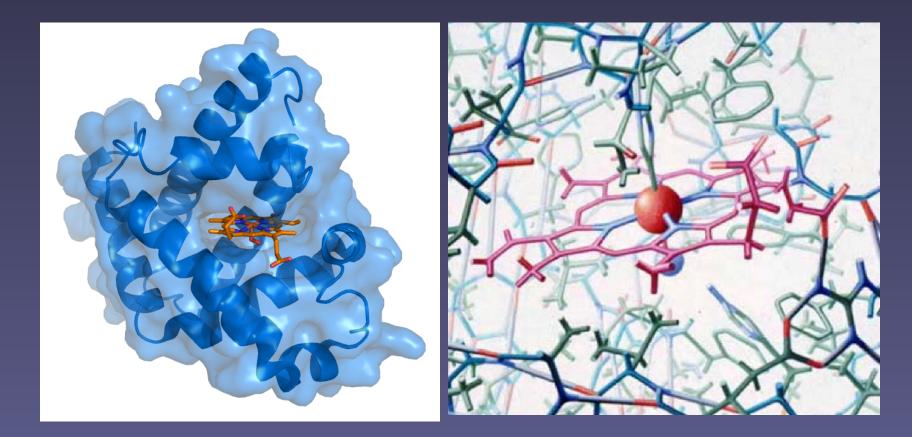
Beta-helix

Greek-key

Is knowledge of 3-D structure enough to understand the function?

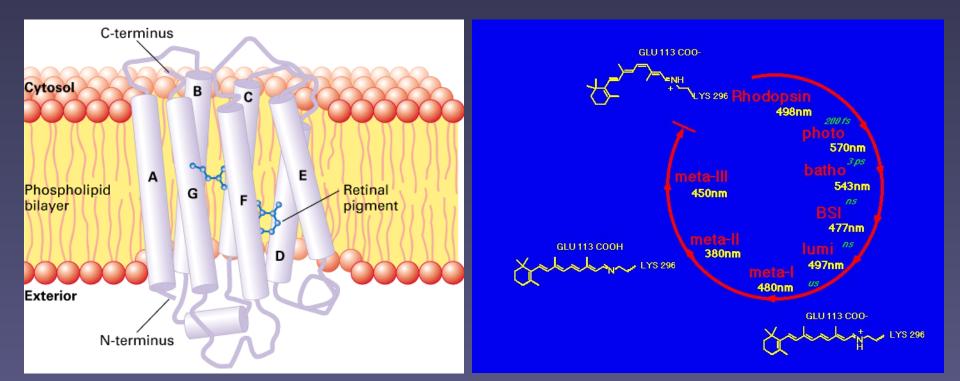
What we don't know?

Example 1: Myoglobin



Breathing motions in myoglobin opens up pathways for oxygen atoms to enter its binding site or diffuse out

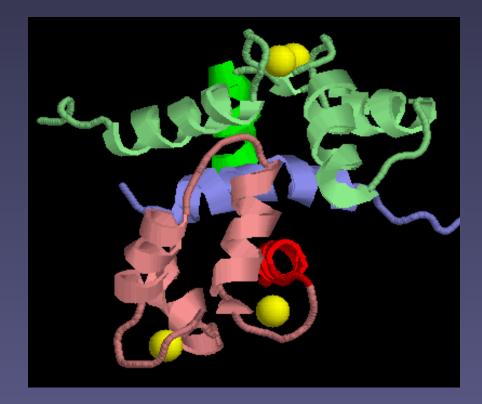
Example 2: Rhodopsin



GPCRs like rhodopsin undergo conformational changes during signal transduction

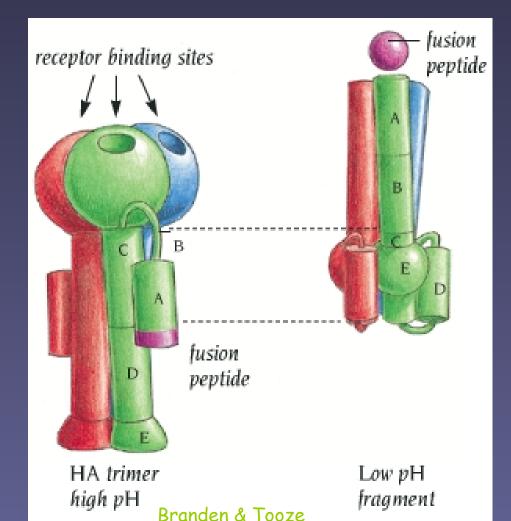
Example 3: Calmodulin





Largest ligand-induced interdomain motion known in proteins

Example 4: Hemagglutinin



Hemagglutinin from influenza virus undergoes large conformational changes

At low PH, the N-terminal helix moves 100 Å to bring the fusion peptide closer to the host cell membrane

Why Molecular Dynamics?

Experimentally determined structures are static

They represent the average structure of an ensemble of structures

They do not provide the dynamic picture of a biomolecule

Molecular dynamics is one way to understand the conformational flexibility of a biomolecule and its functional relevance

Biological molecules exhibit a wide range of time scales over which specific processes

•Local Motions (0.01 to 5 Å, 10⁻¹⁵ to 10⁻¹ s)

- Atomic fluctuations
- Sidechain Motions
- Loop Motions

Rigid Body Motions (1 to 10Å, 10⁻⁹ to 1s)
Helix Motions
Domain Motions (hinge bending)
Subunit motions

Large-Scale Motions (> 5Å, 10⁻⁷ to 10⁴ s)
Helix coil transitions
Dissociation/Association
Folding and Unfolding

http://cmm.info.nih.gov/modeling/guide_documents/molecular_dynamics_document.htm

Potential Energy Function (Equations)

• Potential Energy is given by the sum of these contributions:

$$V_{bonded}(R) = \sum_{bonds} k_l (l-l_0)^2 + \sum_{angles} k_{\theta} (\theta - \theta_0)^2$$
$$+ \sum_{impropers} k_{\omega} (\omega - \omega_0)^2 + \sum_{torsions} A_n [1 + \cos(n\phi - \phi_0)]$$
$$V_{nonbonded}(R) = \sum_{i < j} (\varepsilon_{ij} [(\frac{r_{ij}^{\min}}{r_{ij}})^{12} - 2(\frac{r_{ij}^{\min}}{r_{ij}})^6] + \frac{q_i q_j}{4\pi \varepsilon_r \varepsilon_0 r_{ij}}$$

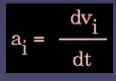
Molecular Dynamics

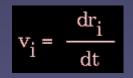
- Calculate Energy 'E' using the potential Energy function
- Calculate Force by differentiating the potential Energy
- Calculate Acceleration 'a' using Newton's second Law
- Calculate Velocity at a later time 't+dt'
- Calculate Position at a later time 't+dt'
- Calculate Energy at new position.
- Create a Trajectory by repeating the above steps 'n' number of times.

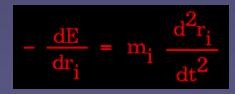
force = mass x acceleration (
$$F_i = m_i a_i$$
)

 dr_i

 $dE = F_i$









m.info.nih.gov/modeling/guide_documents/molecular_dynamics_document.htm

Some Popular Simulation Force Fields

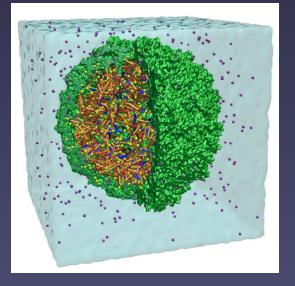
- AMBER (Assisted Model Building with Energy Refinement)
- CHARMm (Chemistry at HARvard Macromolecular Mechanics)
- CVFF (Consistent-Valence Force Field)
- GROMOS (GROningen MOlecular Simulation package)
- OPLS (Optimized Potentials for Liquid Simulations)

First Biomolecular simulation was performed in 1977

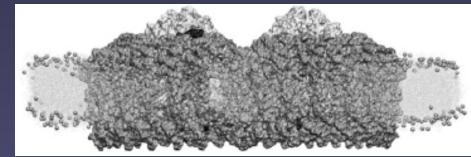
🗿 Dynamics of folded proteins - Microsoft Internet Explorer		
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Journal Home Current Issue AOP Archive	article Nature 267, 585 - 590 (16 June 1977); doi: 10.1038/267585a0	38
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The dynamics of a folded globular protein (bovine pancreatic trypsin inhibitor) have been studied by solving the equations of motion for the atoms with an empirical potential energy function. The results provide the magnitude, correlations and decay of fluctuations about the average structure. These suggest that the protein interior is fluid-like in that the local atom motions have a diffusional character.

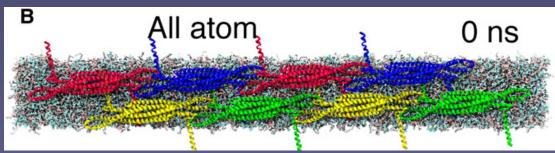
Simulations reaching the million-atom mark



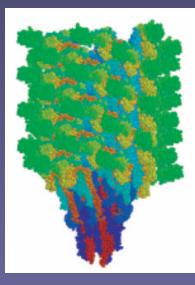
Complete virus: 1 million atoms (Freddolino et al., 2006)



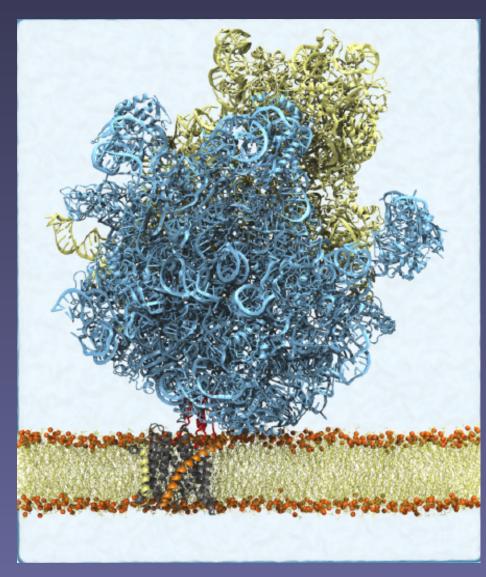
Arrays of light-harvesting proteins – 1 million atoms (Chandler et al., 2008)



BAR domain proteins - 2.3 million atoms (Yin et al., 2009) The flagellum - 2.4 million atoms (Kitao et al., 2006)



MD of protein-conducting channel bound to ribosome



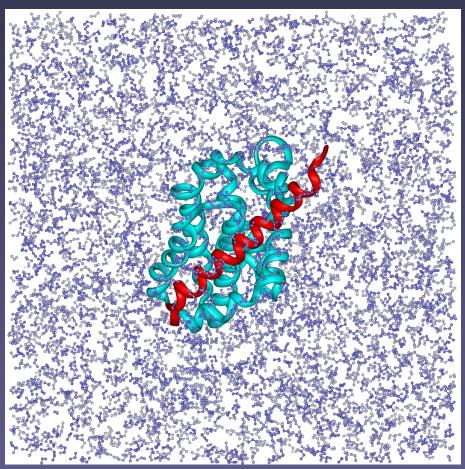
Bacterial ribosomes are important targets for antibiotics

2.7 million atoms50 ns simulation

Largest system simulated to date

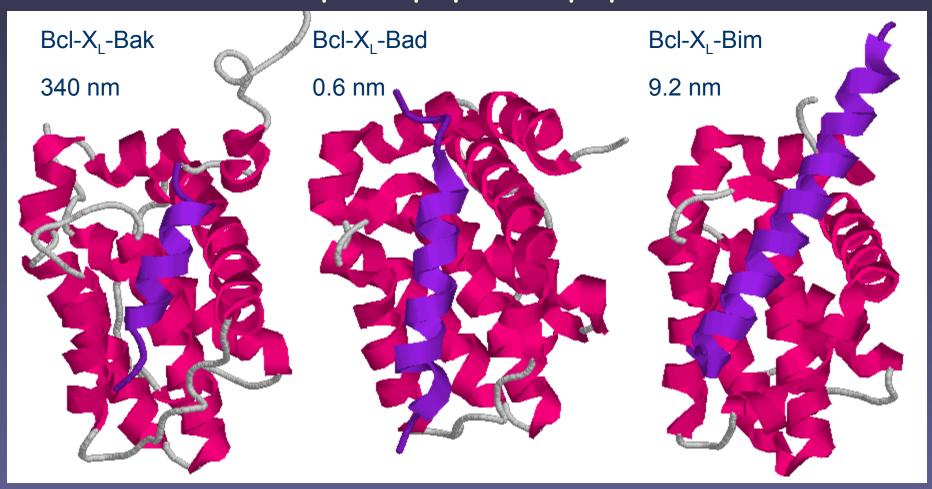
Gumbart et al. (2009)

Biomolecular structures should be simulated under native environment



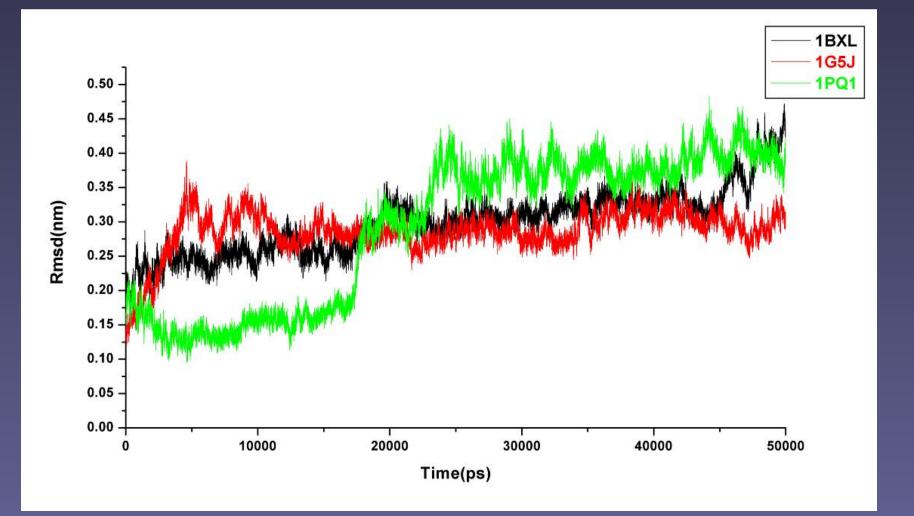
Simulation conditions should be similar to that observed under physiological conditions

Bcl-X_L protein has different affinities for different BH3 pro-apoptotic peptides



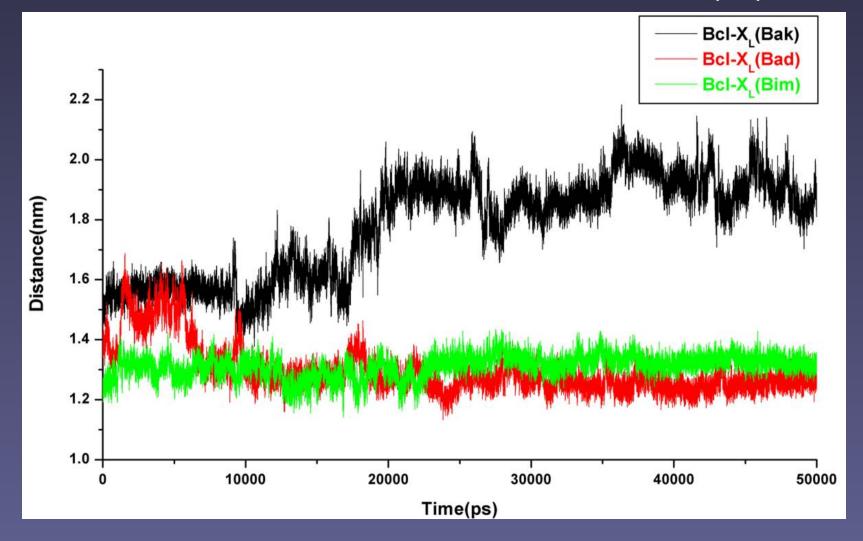
What are the factors that contribute to the different affinities of BcI-X_L?

RMSD Analysis



Lama and Sankararamakrishnan, Proteins (2008)

Distance between helix H3 and the BH3 peptide



Bak peptide moves away from helix H3

Lama and Sankararamakrishnan, Proteins (2008)

Protein-peptide interactions

Bad residue ^{b, c}
F162 - 3.64 (85)
F162 - 3.47 (97)
F162 - 3.63 (88)
F158 – 3.53 (91)
M154 - 3.73 (85); S155 - 4.11 (71)
L151 - 3.78 (78); M154 - 3.71 (86)
Y147 - 3.21 (98); E150 - 3.89 (76); L151 - 3.92 (71)
L151 - 3.79 (82)
Y147 - 3.22 (95)
N140 - 3.83 (64); A143 - 3.55 (90); Y147 - 3.11 (92), W142 - 3.41 (94)
A144 - 4.03 (62)
N140 - 4.20 (56)
N140 - 3.53 (78); A144 - 3.52 (96)
L141 - 3.58 (87); A144 - 3.63 (89); Q145 - 3.19 (99)
A144 - 3.37 (98); G148 - 3.68 (87); L151 - 4.13 (50)
Q145 - 3.63 (81); G148 - 3.84 (72); R149 - 3.37 (93); R152 - 3.07 (99)
G148 – 3.75 (75); L151 – 3.70 (90); R152 – 3.40 (94)
D160 - 3.25 (95)
V159 - 3.36 (97)
F158 - 3.62 (79); V159 - 3.78 (80); D160 - 3.72 (71)
R152 - 3.62 (80)
L151 - 3.71 (86)
K164 - 3.64 (80)
K163 - 3.27 (99); K164 - 3.31 (95)
D160 - 3.01 (93)

Lama and Sankararamakrishnan, Proteins (2008)

Acknowledgements

Anjali Bansal Dilraj Lama Alok Jain Tuhin Kumar Pal Priyanka Srivastava Vivek Modi Ravi Kumar Verma Krishna Deepak Phani Deep



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