

Outline.

- 1) Review of entropy / second law
in thermodynamics.
- 2) Fluctuation theorems
 - Transient (Jarzynski, Crooks, Seifert)
 - SSFT (Grakovitch - Cohen, Kurchan, Spohn - Lebowitz)
(Long time)
- 3) Derivation for systems with
Markovian dynamics

Lecture 1 (Transient)

①

FLUCTUATION THEOREMS IN NONEQUILIBRIUM PHYSICS.

Brief review of thermodynamics, notion of entropy, ~~and~~ equilibrium statistical physics, linear response theory.

First law of thermodynamics. — Conservation of energy.

$$dU = dQ + dW$$

↓ ↓
heat flow Work done
into system on system

Second law of thermodynamics:

Clausius: Impossible for a refrigerator, working in a cycle, to transfer heat from cold to hot reservoir with (no other changes).

- ① Leads to concept of absolute temperature.
- ② For a reversible process $\frac{dQ}{T} = dS$ is a perfect differential.

(2)

Hence entropy S is a function of the state variables.

Fundamental equation $S = S(U, V, N)$

Statistical mechanics.

$$S = K \ln \Omega(U, V, N).$$

Phase space Volume
in energy shell around
 U

No. of states

Relation between Thermodynamics and Mechanics.

$$\Omega = \frac{1}{h^{3N}} \int d^N q d^N p \ \delta(H(q, p) - U)$$

Note : S is a function of the thermodynamic state and not a function of phase space variables (q, p) .

In stat mech, it is a property of an ensemble or a probability distribution in phase space.

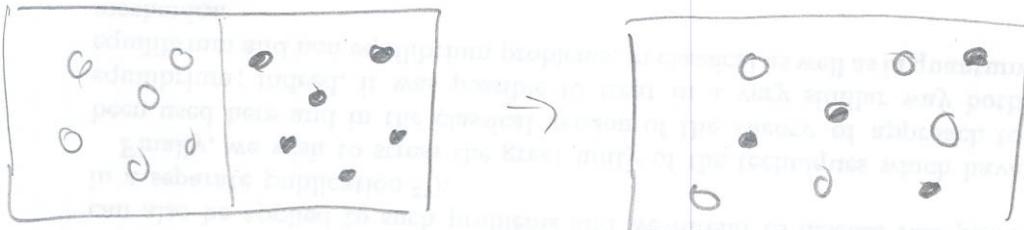
$$S = -k \sum_i P_i \ln P_i$$

Probability of being in.
 P_i = i th microscopic

(3)

The second law also tells us that for an irreversible process the total entropy of the universe increases.

e.g.



- * Entropy is defined only for equilibrium states. If a system goes from one equilibrium state A to another equilibrium state B then

$$\text{then } S_B - S_A \geq 0.$$

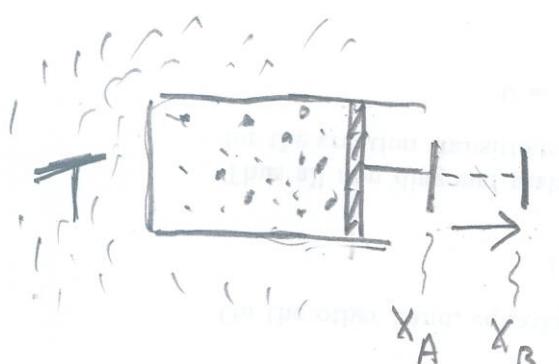
for an isolated system

Otherwise \leftarrow
total entropy change ≥ 0 .

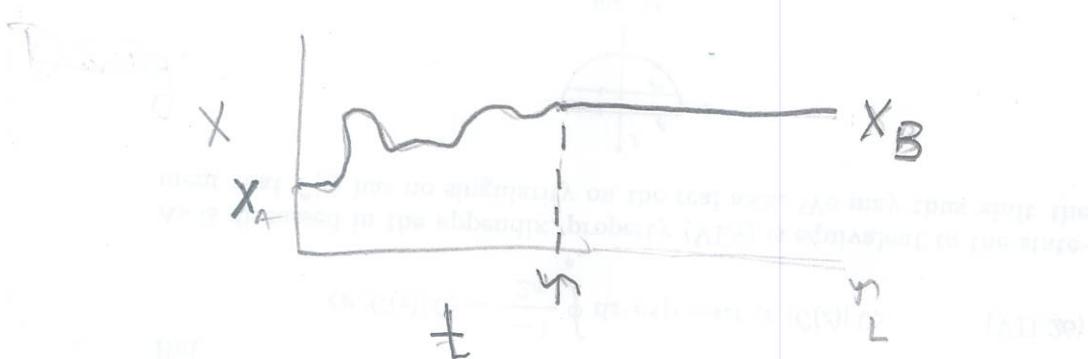
For systems that are not isolated but in contact with either heat baths or particle reservoirs it is useful to construct other "thermodynamic free energies".

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Consider system in contact with heat bath at temperature T



Imagine moving a piston from position x_A to x_B in a finite time τ . At the end of τ we wait for a long time and allow the system to equilibrate to temperature T .



Let $Q =$ heat absorbed by system in time τ_L

$W =$ Work done on system

$$\Delta S = S(x_B) - S(x_A)$$

Then 2nd Law $-\frac{Q}{T} + \Delta S \geq 0$

1st Law $Q = \Delta U - W$

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$$\text{Hence } \frac{1}{T} [T\Delta S - \Delta U + w] \geq 0$$

For an isothermal process $F = U - TS$

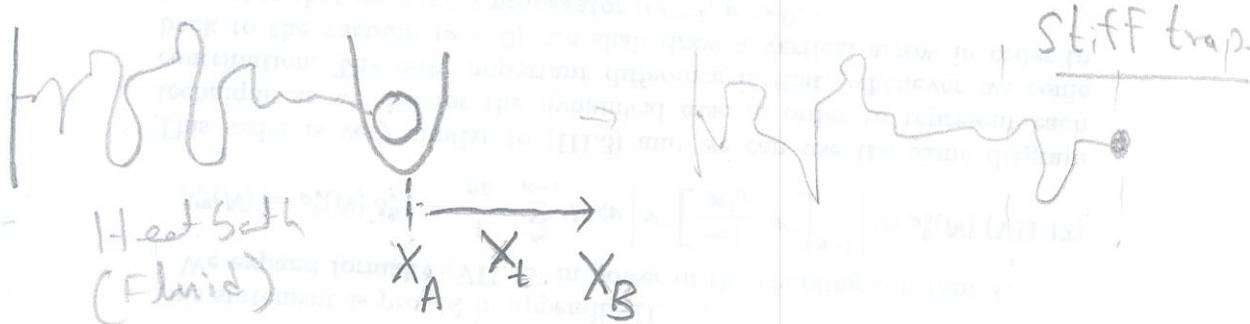
$$\Delta F = \Delta U - T\Delta S$$

\therefore We have Change in Entropy

$$\Delta S_{\text{tot}} = \frac{1}{T}(w - \Delta F) \geq 0$$

$$\frac{\Delta S_{\text{tot}}}{K_B} = \beta(w - \Delta F) \geq 0$$

Now imagine doing an experiment on a small system such as a RNA molecule.



Free energy is a function of T, X, N

$$F = F(T, N, X) \quad \text{We do an experiment}$$

where $X = X_t$ is the time dependent parameter

which changes from value X_A to $X_B \rightarrow$ Specifies the driving protocol.

(6)

- At time $t=0$ the polymer is in thermal equilibrium at temperature T and end-to-end distance fixed to x_A .
- We stretch the polymer from x_A to x_B , $x_B > x_A$ following the protocol x_t in time.
- The polymer has a Hamiltonian H which depends on the parameter x_t , $H = (q_1, q_2, \dots, q_n, p_1, p_2, \dots, p_n; x_t)$

The Work done $W_J = \int_0^T dt \frac{\partial H}{\partial x_t} = \int_0^T dt \frac{\partial H}{\partial x_t} \dot{x}_t$

Can be measured.

- * Unlike the thermodynamic work W , W in a small system, ΔW is now a fluctuating variable.
- If we repeat our experiment with the same protocol we will measure different values of W_J because:

W_J is a function of the path in phase space along which the system evolves. This fluctuates because

- (1) initial conditions fluctuate
- (2) path fluctuates because of interactions with heat bath.

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For a particular realization it is possible that $W_J - \Delta F < 0$.

"Transient Violation of the second law"
- discussed by Perrin.

Can show $\langle W_J \rangle - \Delta F > 0$.

Define: $S = \beta(W_J - \Delta F)$, entropy generated on a particular realization of the stretching experiment. Fluctuating Variable with a distribution $P(S)$.

Fluctuation theorems: General results on these entropy production fluctuations.

Jarzynski's Equality

$$\langle e^{-S} \rangle = 1 \Rightarrow \langle e^{-\beta W_J} \rangle = e^{-\beta \Delta F} \quad [\text{EXACT EQUALITY.}]$$

Jensen's inequality $e^{-\langle x \rangle} \leq \langle e^{-x} \rangle$

$$\Rightarrow e^{-\beta \langle W_J \rangle} \leq e^{-\beta \Delta F}$$

$$\Rightarrow \langle W_J \rangle \geq \Delta F$$

$$\langle e^{-\beta W_J} \rangle = \int_{-\infty}^{\infty} e^{-\beta W_J} P(W_J) dW_J \quad \boxed{\text{Not the characteristic function.}}$$

(8)

Proof: If $H = H(\bar{q}, \bar{p}; \lambda_t)$ $\lambda_0 = \lambda_A$, $\lambda_T = \lambda_B$

Simplest Version:

- At $t=0$ system is in contact with a heat bath and so in thermal equilibrium at temperature T .

- The free energy of the system is

$$F(T, \lambda_A) = -k_B T \ln Z(\lambda_A)$$

$$Z(\lambda_A) = \int e^{-\beta H(\lambda_A)} d\bar{q} d\bar{p}$$

- The contact with heat bath is then removed and the work protocol λ_t is started.

Between times $0-T$ system evolves through Hamiltonian dynamics

$$\dot{q}_i = -\frac{\partial H(\lambda_t)}{\partial p_i} \quad \dot{p}_i = -\frac{\partial H(\lambda_t)}{\partial q_i}$$

- Measure work done in time T

$$W_J = \int_0^T dt \frac{\partial H}{\partial t}$$

Note

$$\frac{dH}{dt} = \underbrace{\frac{\partial H}{\partial q_i} \dot{q}_i + \frac{\partial H}{\partial p_i} \dot{p}_i}_{\text{}} + \frac{\partial H}{\partial t}$$

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$$\text{Hence } W_J = H(\bar{q}(r), \bar{p}(r), \lambda_r) - H(\bar{q}(0), \bar{p}(0), \lambda_0) \\ = H_B - H_A.$$

$\therefore \langle e^{-\beta W_J} \rangle \rightarrow \text{average over initial configurations}$

$$= \int d\bar{q}_0 d\bar{p}_0 e^{-\beta(H_B - H_A)} \frac{e^{-\beta H_A}}{Z_A}.$$

Prob. of initial state

$$= \int d\bar{q}_0 d\bar{p}_0 \frac{e^{-\beta H_B}}{Z_A}$$

Liouville's theorem $d\bar{q}_0 d\bar{p}_0 = d\bar{q}_f d\bar{p}_f$

Using $\int d\bar{q}_f d\bar{p}_f e^{-\beta H_B} = Z_B$, we get

$$\boxed{\langle e^{-\beta W_J} \rangle = \frac{Z_B}{Z_A} = e^{-\beta(F_B - F_A)}} \rightarrow \text{QED.}$$

(10)

Quantum systems (Kurchan) Tasaki)

Work has to be defined in terms of measurements.

- At $t=0$ system in equilibrium \rightarrow measure it's energy. System evolves unitarily over time $T \rightarrow$ measure energy again. $W_J = E_T - E_0$.
- At $t=0$ we find the system in an energy eigenstate $E_\alpha^{(0)}$ with probability $\frac{e^{-\beta E_\alpha^{(0)}}}{Z_A}$.
- The eigenstate $|\Psi_\alpha^{(0)}\rangle$ evolves to $U_T |\Psi_\alpha^{(0)}\rangle$ where $U_T = T \left[e^{-\frac{i}{\hbar} \int_0^T dt \hat{H}(\lambda_t)} \right]$
- At $t=T$ we will find the system in an energy eigenstate $E_\beta^{(T)}$ with probability $|\langle \Psi_\beta^{(T)} | U_T | \Psi_\alpha^{(0)} \rangle|^2$.
- Work done is $W_J = E_\beta^{(T)} - E_\alpha^{(0)}$

$E_\alpha(t)$ satisfies $H(\lambda_t) |\psi_\alpha(t)\rangle = E_\alpha(t) |\psi_\alpha(t)\rangle$

instantaneous eigenstates.

(11)

$$\therefore \langle e^{-\beta W_0} \rangle$$

$$= \sum_{\alpha, \beta} e^{-\beta(E_\beta(r) - E_\alpha(0))} \frac{e^{-\beta E_\alpha(0)}}{Z_A} \langle \psi_\beta(r) | U_r | \psi_\alpha(0) \rangle \langle \psi_\alpha(0) | U_r^+ | \psi_\beta(r) \rangle$$

$$= \frac{\sum_{\beta} e^{-\beta E_\beta(r)}}{Z_A} \frac{Z_B}{Z_A} = e^{-\beta \Delta F}$$

[OED]

J.E is an example of an "integrated F.T."

We now give an example of a "defined F.T." — Crooks' identity.

(12)

Again a classical system isolated from heat bath at $t=0$.

$$P(W_j) = \int d\bar{q}_0 d\bar{p}_0 \delta(W_j - (H_B - H_A)) \frac{e^{-\beta H_A}}{Z_A}$$

$$\downarrow$$

$$H_A = H_B - W_j$$

$$= \int d\bar{q}_r d\bar{p}_r \delta(-W_j - (H_A - H_B)) \frac{e^{-\beta H_B}}{Z_B} \left(e^{\beta W_j} \frac{Z_B}{Z_A} \right)$$

Now if we consider the transformation (time reversal)

$$q'_i = q_i, p'_i = -p_i, t' = T-t \quad \text{we see that}$$

q'_i, p'_i satisfy the equations of motion with the time-reversed protocol $\dot{x}_t = \lambda_{T-t}$.

Hence under this protocol the system will evolve from $(\bar{q}_r, -\bar{p}_r)$ at time $t=0$ to $(\bar{q}_0, -\bar{p}_0)$ at time $t=T$.

$$\text{Work done is } H_A - H_B = -(H_B - H_A)$$

Work done in forward protocol.

(13)

$$\therefore \int d\bar{q}_r d\bar{p}_r \delta(-w_j - (H_A - H_B)) \frac{e^{-\beta H_B}}{Z_B}$$

$$= P_R(-w_j)$$

$P_R(w_j) \equiv$ Distribution of work under the protocol $\bar{\chi}_t$

Hence $P(w_j) = P_R(-w_j) e^{\beta(w_j - \Delta F)}$

OR for $s = \beta(w_j - \Delta F)$

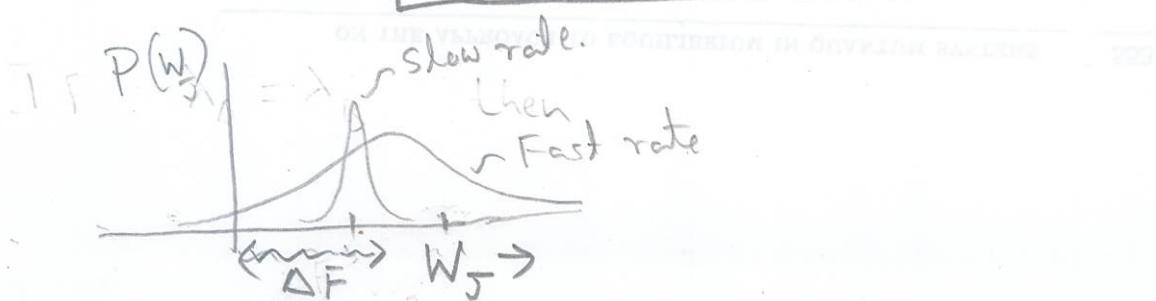
$$\frac{P(s)}{P_R(-s)} = e^s$$

Crook's Identity

J.E Follows from C.E.

$$\left(\int_{-\infty}^{\infty} w_j P(w_j) e^{-\beta(w_j - \Delta F)} \right) = \int_{-\infty}^{\infty} P_R(-w_j) dw_j = 1$$

$$\Rightarrow \langle e^{-\beta w_j} \rangle = e^{-\beta \Delta F}$$



Transient Fluctuation Theorems are ⑭
Interesting because:

- General results valid far from equilibrium.
Tells us about how improbable rare fluctuations are.
- Possible method for estimating equilibrium free energy profile from non-equilibrium measurements. (experiments and simulations)
- Verified in experiments on RNA stretching and trapped colloidal particles.