Unstable Chalcogenide Glasses & Application of Crystal Nucleation: Phase-change Non-volatile Memory

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School on Glass-formers & Glasses: Bangalore (4-20 January 2010)

Or:

A Replacement Memory Technology for Flash

Synopsis

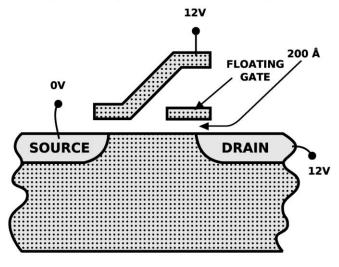
- Introduction Flash memory
 - crystal nucleation/growth
- Molecular-dynamics (MD) simulation method
- Homogeneous nucleation & crystallization (Ge-Sb-Te) = GST*
- Heterogeneous nucleation & crystallization (GS,Sb) – 'interface engineering'
- Structural analysis of amorphous GST-225, GS
- Summary

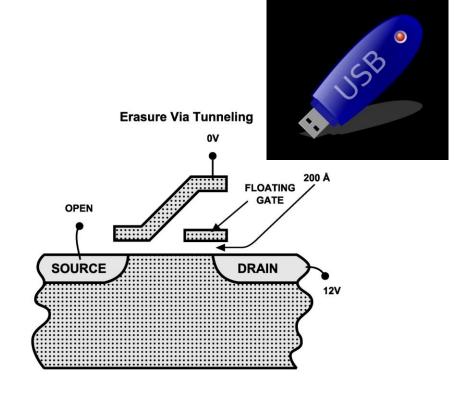
^{*} Hegedus & SRE, Nature Materials, 7, 399 (2008)

Flash Memory: What is it?

MOSFET with extra 'floating' gate (FG)

Programming Via Hot Electron Injection





- electrons are stored ('permanently') in electrically isolated FG (non-volatile memory)
- → change threshold voltage controlling whether or not an inversion layer forms, allowing a source-drain current to flow ('1') or not ('0')
- NOR flash: slower, less memory → mobile phones
- NAND flash: faster, more memory → MP3, hard-drive replacement
- Market = \$20 bn p.a. (2008)

Flash Memory: The problem

- Flash (MOSFET) memory has been obeying Moore's Law:
 - memory capacity doubles every 2 years as feature size reduces
- Samsung and Toshiba announced, in October 2007, a 64Gb NAND flash memory
 - with **30nm** feature size
- But there is an ultimate limitation:
 - as the insulator thickness decreases below **7nm**, the electrons trapped in the floating gate (FG) will leak out \rightarrow *volatility*
 - as the FG gets smaller → limit of 1 electron stored
- → FLASH technology will stop evolving in ~2012
- What next?

Phase-change Technology

Ge-Sb +Te alloys, eg Ge₂Sb₂Te₅ (GST-225),
 can be used for *non-volatile* memories

 based on ultrafast (~10ns) reversible, rewriteable (RW) phase change (PC) between amorphous & crystalline states

property contrast between a- & c-states
 → (0,1) bit

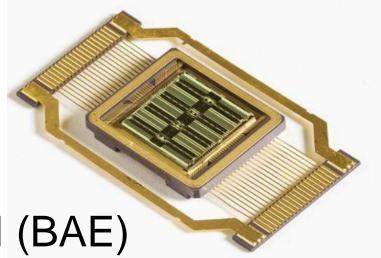
optical reflectivity contrast (DVD-RW, Blu-ray)



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electrical resistivity contrast (PC-RAM)

4 Mbit 'rad-hard' PC-RAM (BAE)

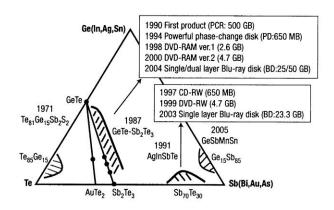


New PCRAM technology

- Scalable
- Extremely fast
 - can rewrite data without erasure
 - → 30x faster than Flash
- Fewer processing steps than NOR Flash
- Samsung mass-producing 512 MB PCRAM chips (see Nature online, Sept 25, 2009)
- → 1GB in late 2009?



Materials for PC (optical) data storage



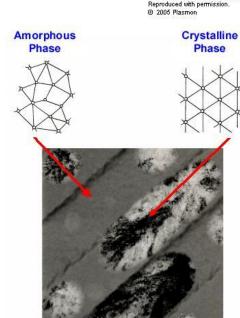
Main materials are:

- Ge-Sb-Te (GST), e.g. Ge₂Sb₂Te₅ ("225")
- Ge-Sb (GS), AgInSbTe (AIST)

How is the PC change caused?

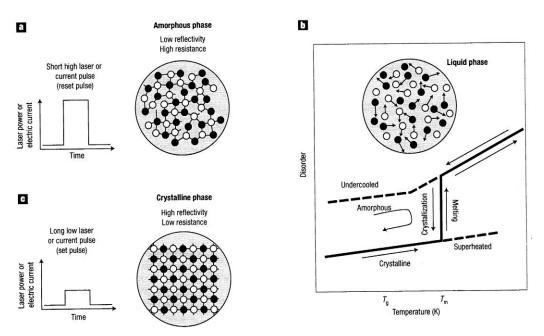
 In optical discs, brief laser pulses cause melting & quenching to amorphous state/crystallization of amorphous spot

 In PC-RAM devices, electricalcurrent Joule-heating pulses cause amorphization/crystallization



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Write/erase steps in PC data storage

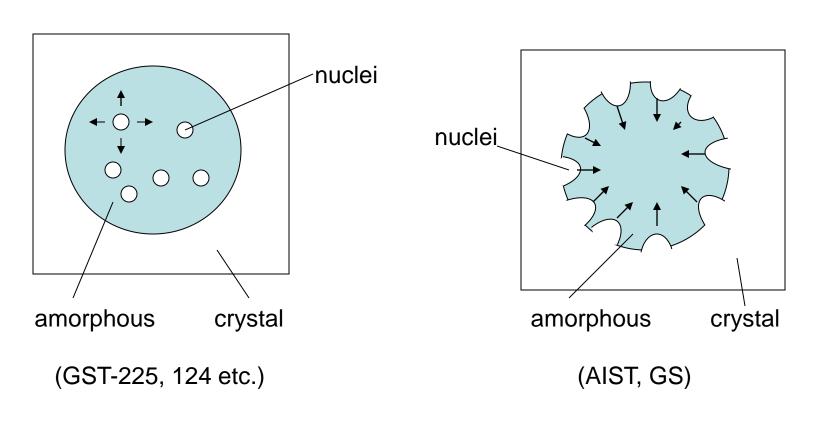


[Wuttig and Yamada Nat. Mat., **6**,824 (2007)]

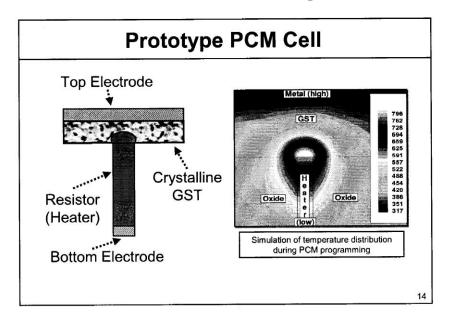
- Erased (optical) state is crystal
- written spot is amorphous
- Reset (electrical) state is amorphous
- set state is crystal

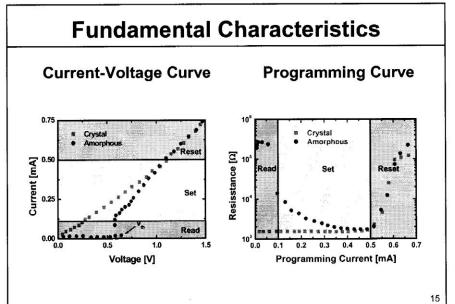
Erasure Mechanism: Crystal Nucleation vs. Growth

Homogeneous nucleation vs heterogeneous nucleation



PC-RAM Memory devices

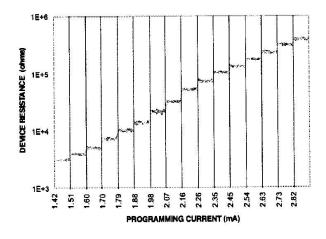




Attwood (Intel)

PC-RAM memory devices

- Advantages:
 - scalability (IBM has demonstrated PC behaviour down to 2.5nm)
 - multi-level storage capability (>1 bit/cell)

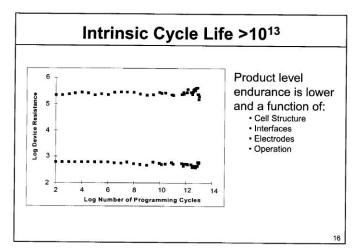


very fast "set" speed(crystallization occurs in ~1-10 ns)

- very long cycle life

high/low resistance \rightarrow 1 bit (0, 1)

 2^n intermediate resistance states $\rightarrow n$ bits



Attwood (Intel)

Features of phase changes in GST materials: I

Crystallization/amorphization is:

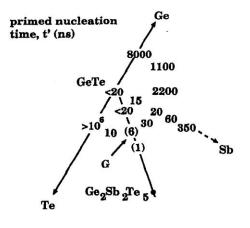
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-rapid (amorphous-crystal t > 10ns) (very 'bad' glass)
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-reversible

WHY??

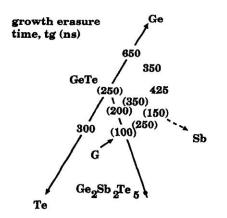
GST nucleation and growth times

Nucleation times, t_{nuc}



[Coombs *et al*, J. Appl. Phys. **78**, 4925 (1995)]

• Growth times, t_{ar}



- $t_{\text{nuc}} < t_{\text{gr}}$ for 225 GST (homogeneous nucleation dominant)
- $t_{\text{nuc}} > t_{\text{gr}}$ for GS (heterogeneous nucleation and growth dominant)

Features of phase changes in GST materials: II

 Crystal structure of GST involved in PC transformation is the metastable rocksalt (NaCl) structure + atomic vacancies:

NB: 4-rings, 90° bond angles

- e.g. for 225-GST, 1 vacancy/formula unit distributed among the Ge/Sb sub-lattice
- (2 Ge+2 Sb+1 vacancy = 5 Te)
- how are vacancies distributed?
- What is the structure of the amorphous phase?

Origin of c-a resistivity contrast

- Crystal octahedral coordination associated with resonant p-bonding
 - → metallic behaviour

- Amorphous structure destroys long-range resonant bonding
 - semiconducting behaviour

(Wuttig 2009)

What are the Problems?

- Why is crystallization so fast (~10ns)?
- How can it be made even faster?
- How to optimise property contrast between amorphous & crystalline phases of PC materials?
 - → ab initio Molecular Dynamics (AIMD)

- in silico rational materials design

Molecular-Dynamics (MD) Simulations

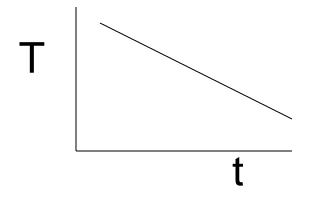
- Compute trajectories of an assembly of particles ('atoms'), interacting via a given potential, according to Newton's laws of motion
- Particles are in a simulation box with periodic boundary conditions (to eliminate surface effects)
- Interatomic potential can be computed quantummechanically (ab initio), albeit with approximations (DFT-LDA, GGA)

Ab initio MD of GS(T) phase changes

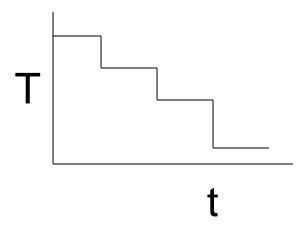
- VASP ab initio MD code
- 63-90 atom models for GST-225,124
- 150 atom models for GS, Sb
- Cubic cells + periodic-boundary conditions (+ crystal templates)
- Constant volume
- Quenching from melt + heating amorphous
- 6 month runs on 32 nodes of supercomputer

Simulated cooling protocols

Continuous ramp



 Discontinuous + isothermal

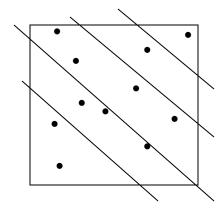


Dynamic structural characterization of quenched/annealed models

1. Long-range order

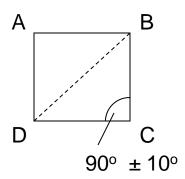
Maximal 3D Fourier intensity

 $I_{\text{max}} \rightarrow N$ if all atoms are on lattice sites/planes of a single crystal

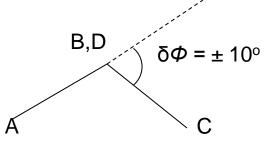


2. Medium-range order

4-rings (structural unit of rocksalt structure)



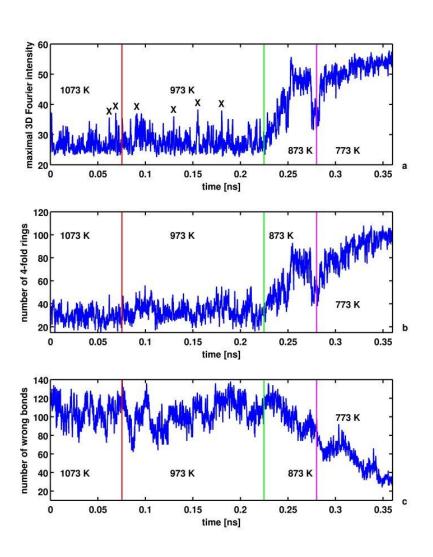
4-rings are included with some variations in bond/flex angles



3. Short-range order

Wrong bonds (Ge-Ge, Sb-Sb, Te-Te, Ge-Sb in GST)

Liquid quench: GST 225



Simulated Crystallization

a) GST-225 (slow liquid-quenched) c) a-GST-225

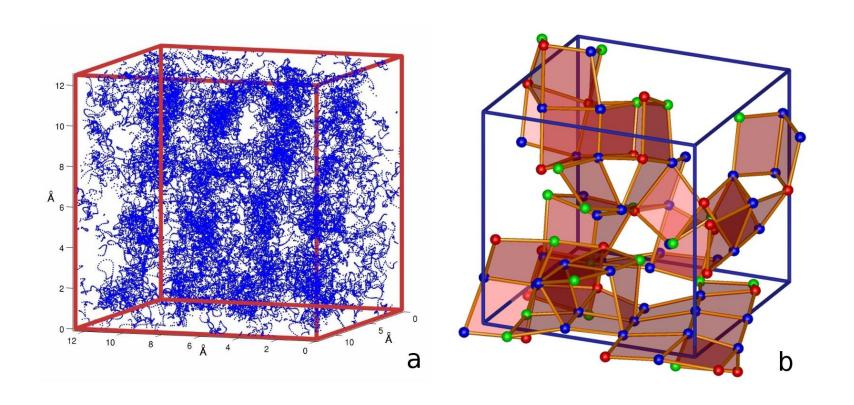
b) GST-124 (slow liquid-quenched)

c) a-GST-225 (fast liquid-quenched)

Ge= green; Sb= red; Te= blue

d) GST-225 (amorphous-annealed)

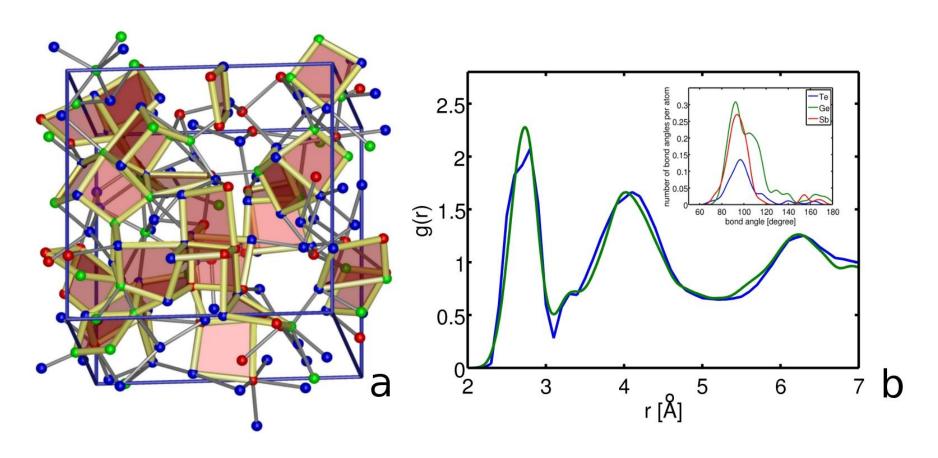
Structural order in *liquid* GST-225



a) Cumulative trajectories of all63 atoms for 20ps at 973K

b) Highly-ordered transient cluster of 4-rings at 1073K

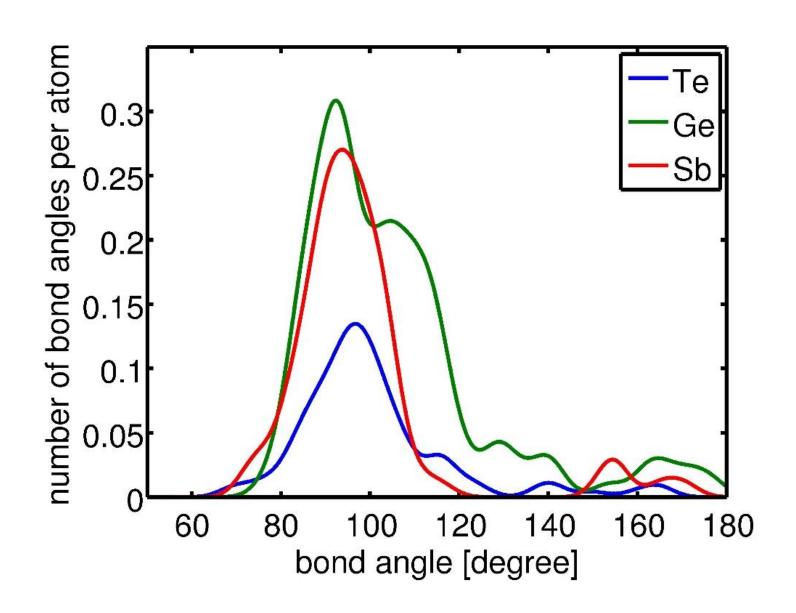
Structure of amorphous GST-225



Nb square 4-rings

Pair distribution function

Bond-angle distributions for amorphous GST-225



Structure of amorphous GST

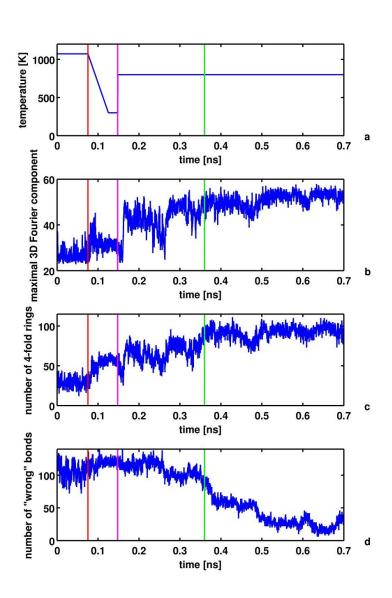
- Ge coordination is mainly 4 but mostly defective octahedral, not tetrahedral
- Ge bond-angle distribution (BAD) peaks mostly at ~90° (octahedral)
- Subsidiary Ge BAD peak at ~109° (tetrahedral)
- Sb coordination is mainly 3 but also 4 BAD peak at ~ 90°
- Te coordination is mainly 2 but also 3 BAD peak at ~ 90°

Structure of amorphous/glassy GST

- Structure of as-deposited (sputtered) a-GST is NOT the same as device-quenched glass
 - (EXAFS, XANES)
- More tetrahedral Ge in as-deposited material
- XANES spectra calculated from AIMD model are in good agreement with experimental data for device-quenched material

(Kolobov, SRE 2010)

GST-225: amorphization on rapid quenching, and crystallization on annealing



Why is crystallization so rapid?

 Crystal-nuclei seeds (fragments of octahedral rocksalt structure, e.g. 4-rings) already present in the liquid & amorphous states

 Therefore minimal atomic diffusion is required for crystal nucleation & growth

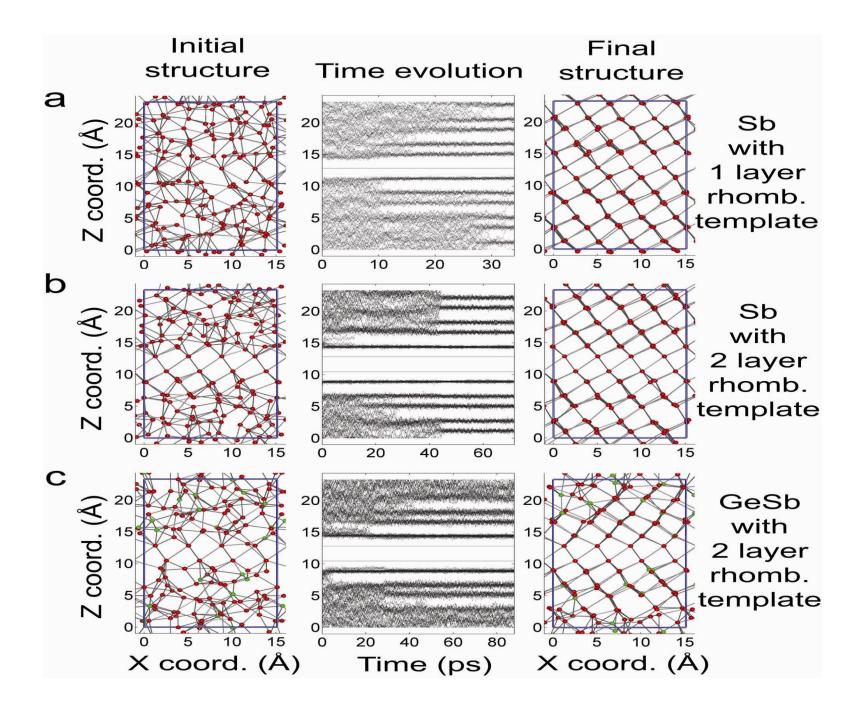
GS - why change from GST?

- No Te in CMOS fab. line (IBM)
- GS crystallizes faster
- But crystallization is growth-limited
- And crystal phase is rhombohedral (A7) Sb
- But material phase-separates on cycling (Sb, Ge)

Simulation of heterogeneous nucleation: I

- Simulate growth-limited crystallization in GS,Sb
- Isothermal MD runs at 600K
- Introduce crystal template into simulation box
 - -rhombohedral (A7) template of Sb

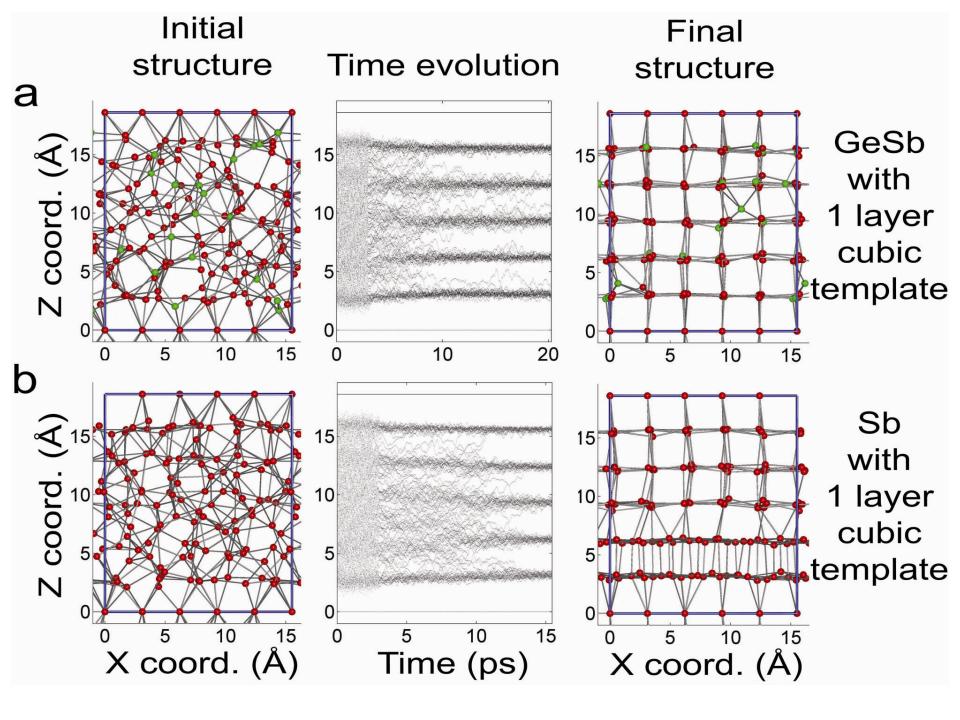
Nb **NO** untemplated crystallization in >100ps



Simulation of heterogeneous nucleation: II

Design new materials by choice of appropriate templates

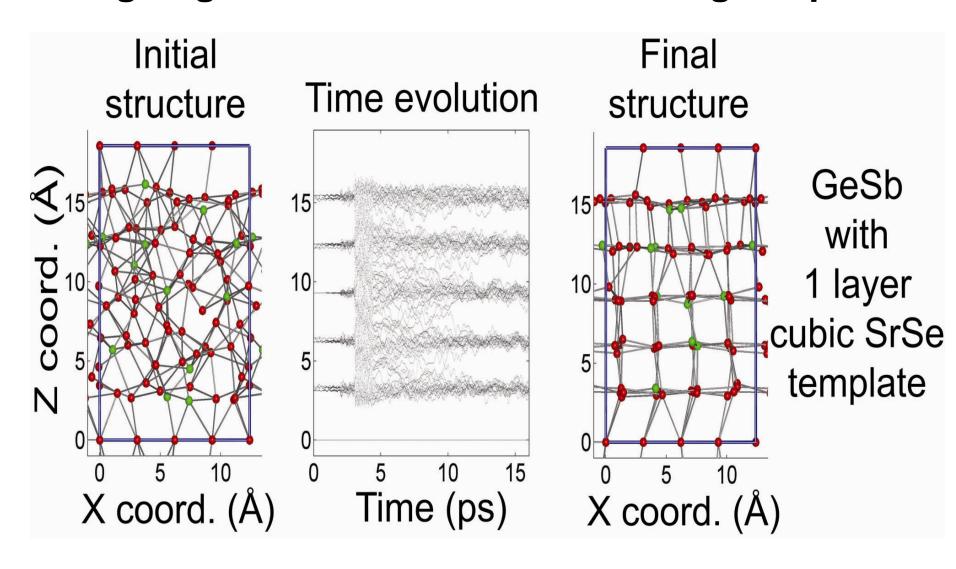
 Choose cubic template to try to force GS, Sb materials to crystallize into cubic crystal structure (like GST)



Crystal templating

- Unfeasible to use cubic Sb/GS template in real PCRAM cells because rhombohedral phase is stable (and would melt during reset operation)
- Use different, refractory cubic material for template layer with matched lattice constant:
 - -eg rocksalt SrS (a=6.024A; MP=2000°C), or SrSe (a=6.236A; MP=1600°C)
 - -cf cubic Sb (a=3.1A); MP(eutectic GS)=600°C

Designing new PCRAM materials using templates

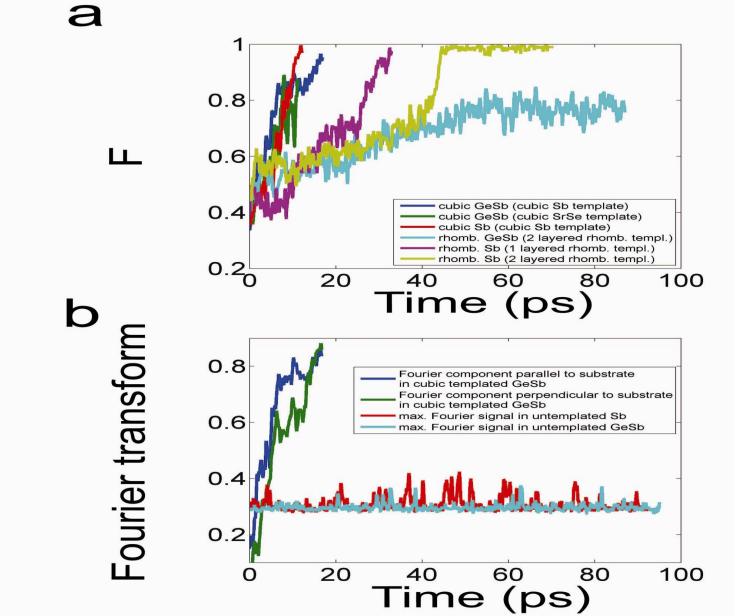


Time dependence of crystallization

 a) Calculate time-dependent fraction, f(t), of atoms lying closer to crystal plane than a chosen value

b) Calculate time-dependent 3D Fourier transform

Time Dependence of Crystallization



Templated crystallization speeds

Cubic templated GS in simulation box of width ~20A

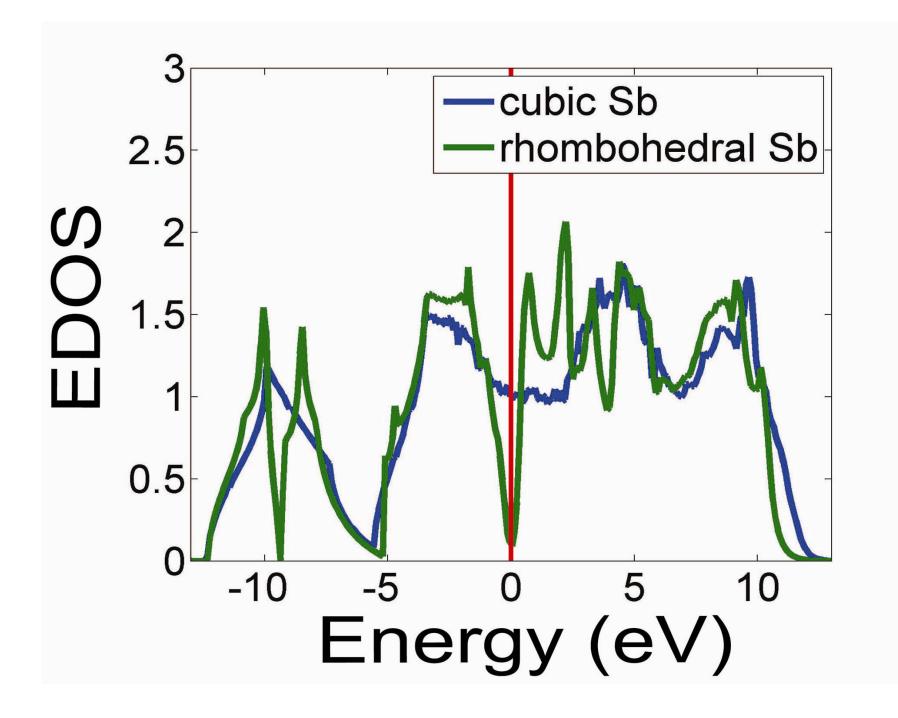
-crystallizes to cubic phase in ~ 20ps

- Extrapolated to 25-nm wide PCRAM cell, cubic crystallization is predicted to occur in ~ 0.25ns
- PCRAM using cubic-templated Sb/GS may be suitable for non-volatile DRAM replacement (t < 1.5ns) ?

Electronic Properties of cubic Sb

Electronic density of states (EDOS)
 (how many electron states at a given energy)

cubic vs rhombohedral



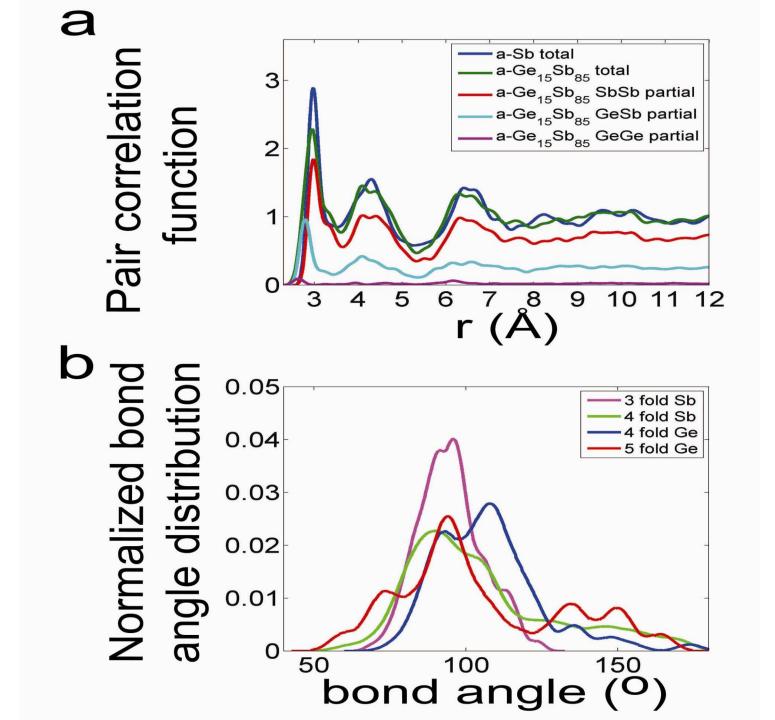
Electronic properties of cubic GS/Sb

- DOS at Fermi energy (E_F) is much greater for *cubic* (cb) than for rhombohedral (rh) phase
- Electrical resistivity contrast is predicted to be greater for {cb/am} than for {rh/am}
- Thus, cb-Sb/GS is predicted to be more suitable for multilevel memory operation

Structure of amorphous Sb, GS

Atom-Atom Pair Correlation Function

Bond-Angle Distribution



Structural features of a-Sb, GS

- Average bond angle in a-Sb is ~90°
- <3>,<4> Sb and <4>,<5> Ge in a-GS have 'octahedral' configurations
- <4> Sb and <4> Ge in a-GS also have tetrahedral configurations (average bond angle ~109°)

'octahedral' Ge → reduced phase separation in cb-GS?

Summary: AIMD

 First(?) example of AIMD simulations matching experiment in time and size

Summary: homogeneous nucleation

- Ab initio MD simulations of Ge-Sb-Te PC materials show crystallization, on slow cooling (~ 1ns) from the melt, of 225, 124, 101, 023, but NOT 160 (Ge-Sb, GS)
- Results can be interpreted in terms of homogeneous crystal nucleation
- Crystal seeds in liquid state are (connected) 4-rings, characteristic of octahedral rocksalt structure
 - 4-rings → fast crystallization on annealing

Summary: amorphous state

- Amorphous GST-225, obtained by rapid quenching, contains:
- 'octahedral' + tetrahedral <4> Ge configurations
 - <3> coordinated Sb, <2> coordinated Te
- <3> Sb in a-Sb, and <3> Sb and <5> Ge in a-GS have 'octahedral' configurations
- <4> Sb and <4> Ge in a-GS also have tetrahedral configurations (average bond angle ~109°)

Summary: heterogeneous nucleation

- Eutectic GS and Sb do not nucleate homogeneously
- Heterogeneous nucleation and crystal growth can be simulated by templating in simulation box
- Crystallization to the experimental rhombohedral phase can be simulated using rh-Sb template
- Crystallization to metastable cubic phase occurs with cubic templates – eg Sb, or SrS, SrSe
- Cubic crystallization is much faster than rhombohedral
 - 0.25ns for a 25nm PCRAM cell
- Cubic phase has much higher EDOS at E_F than for rhomb.
- Cubic-templated GS etc might be suitable for DRAM?
 - → in silico rational design of new PCRAM materials

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